

## Data mining new energy materials from structure databases

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### ABSTRACT

New energy materials that act as clean power sources and data science are developing rapidly in the past decades and the advancement of the two research areas have significantly benefited the development of each other. At the meantime, structural information of materials have been obtained and stored in various structure databases, such as the Cambridge Structure Database (CSD) and the Inorganic Crystal Structure Database (ICSD). Researchers have developed various structure-property relationships of the energy materials, which could be applied to screen the potential suitable materials from structure databases; this has become an efficient route to explore and design new energy materials. In this article, we review recent progresses on the data mining study of new energy materials based on structure databases such as CSD and ICSD, in the context of dye-sensitized solar cells and perovskite solar cells, and also include other energy systems such as water splitting systems, lithium batteries, thermoelectric devices and gas adsorbent materials. The structure descriptors that are more fundamental in the data mining procedure employing the structure-properties relationships are focused; the structural descriptors are complementary to the quantum descriptors and are efficient in the materials design process. We believe that with the successful formulation of more advanced and case-by-case structure-property relationships of energy materials, many new energy materials could be efficiently identified with much lower cost and shorter design period via the data mining process.

### 1. Introduction

There has been global energy crisis and environmental issues in the past decades due to the overuse of fossil fuels. New energy materials should be identified and developed to provide clean energy [1–5]. For example, efficient solar cell materials should be explored to harness the clean energy from the non-exhaustible solar system to provide the electricity; materials incorporated in water-splitting systems should be screened to provide solar fuels; lithium-based materials could be exploited to be tailored to undergo charge/discharge process reversibly and store electrical energies for portable devices and vehicles; metal-organic frameworks could be discovered to capture the carbon and gas molecules effectively; thermoelectric materials and piezoelectric materials should be identified to convert the thermal and mechanical energy into the electric current. At the current stage, the discovery of these energy materials relies predominantly on the experimental serendipity and the try-and-error experimental process that are inefficient and time-consuming. Nevertheless, many evidences have shown that the discovery process of new energy materials could be greatly accelerated by the data mining process, which have already shown their excellence

in pharmaceutical drug discovery, finance, medicine, and marketing [6,7].

A large amount of structure information of new materials have been determined and stored in structure databases, which are mainly prepared by the crystallization processes and solved by the X-ray diffraction techniques and other crystallographic techniques [8]. Most of these materials have their structures accurately determined within the resolution of 0.01 Å and serve as outstanding platforms for the structural analysis and more advanced structure-property analysis. Two outstanding databases of materials structures have emerged: the Cambridge Structure Database (CSD) and the Inorganic Crystal Structure Database (ICSD), which focus on the accurate crystal structures of organic and inorganic materials, respectively [9,10]. These databases formed a good foundation for the data mining process of new energy materials. These databases store structures of materials that have already been synthesized in laboratory and exhibit significant advantages to expedite the process towards achieving promising candidates [11,12].

To help the fundamental understanding of the energy materials and facilitate the following-up materials engineering methods, a series of

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design principles based on materials structures ‘structure-property relationships’ have been developed. Many researches have been devoted to the formulation of the structure-property relationships of new energy materials to better understand the mechanisms involved. A bonus of the structure-property relationship study is that new energy materials with targeting properties could be efficiently identified from the structure database by proper algorithms that consider the structure-property relationship [13,14]. By simply inputting the algorithms that consider the structure-property relationships using specific codes, potential materials available for experimental validation can be retrieved from structure databases [15–21].

In this manuscript, we focus on recent progresses on data mining new energy materials exploiting the structure–property relationships from structure databases, especially CSD and ICSD, in the context of solar cells, water splitting systems, lithium batteries, thermoelectric devices and gas adsorbents. The structural descriptors are mainly focused, with the quantum descriptors also discussed to reveal the data mining process of the new energy materials. While we emphasize on the fundamental crystal structure databases such as CSD and ICSD, it should be noted that a number of derived databases exist such as those storing the calculated materials properties. The readers could be referred to the existing literatures reviewing the high-throughput calculations and derived databases [14,22–28].

## 2. Structural databases

### 2.1. Crystal structure databases

The CSD database presented by the Cambridge Crystallographic Data Centre (CCDC) starts from 1965 to represent the world’s largest repository storing organic crystal structures; it provides a foundation for the data mining of organic materials-based energy devices [14,29]. The CSD includes both small organic molecules and metal organic frameworks (MOFs). The structure information is stored in a standard crystallographic information file (CIF) format which includes the crystal structure, the atomic positions, the bond length and the bond angles [30]. The packing mode including the space group and the symmetry elements are also available [31]. CSD now contains over 900,000 entries of accurate 3D structures from X-ray and neutron diffraction analyses [30,32–34]. User-friendly interface software tools are accessible for the detailed structural analysis [35]. The materials stored in CSD have been already synthesized and structurally characterized, but most of them have never been investigated for energy applications [28,36]. Therefore, new materials could be mined directly from the structure database either by a user-ready search interfaces such as *Conquest*, or writing a code with algorithms describing the searching criteria. For example, searches based on CSD have been conducted using *ConQuest*, performed with short contact analyses and packing diagrams [37].

In contrast with the organic material-based CSD database, the ICSD database contains the crystal structure information of inorganic compounds, with more than 199,000 crystal structure entries in June 2018 [38–40], and ca. 7000 new entries are added per year [16,41]. The ICSD-based data mining process could be coupled with the CSD database, where both organic compounds and inorganic compounds are desired [42].

Other crystal structure database other than CSD and ICSD include the powder diffraction database [43], the Protein Data Bank (PDB) [44–48], and PDB-derived databases such as PDBSite [49,50] and PhosphoSite [51], to name a few. These provide additional structure information on the new materials could be employed to design suitable materials for the targeting applications. In addition, it will not be surprising that with the development of the new devices that requires integration of the energy materials and the biological systems such as skin, the protein-based structure database might be very useful for the development of new energy materials in the future.

In addition to the experimental crystallographic techniques that are employed to obtain the CSD and ICSD databases, the *in silico* methods have also been used to predict the crystal structures of new materials, since in many cases the crystal structures are impossible to solve experimentally [52–55]. The crystal structure prediction (CSP) method has been widely performed to obtain the hypothetical structures [56–58]. The CSP method relies on the force-field method or the quantum mechanics to identify the proper positions of the neighboring molecules that interact stably with the central molecule to build up the crystal structure. In many cases, a proper space group should be set in advance. The CSP could be accelerated by evolutionary algorithms [53,59–61], R-group enumeration [62], and more advanced analysis [61]. In addition, the existing academic publications storing various structures or property data could form a natural basis for the data mining process [63]. Various techniques have been employed to data mine the existing publications, such as text mining, link mining, citation network analysis, adaptive neuro-fuzzy inference systems, neural networks and multilayer perceptrons [64,65].

### 2.2. Property databases

The property databases store property information that are closely related to the materials requirements, but are usually computation-intensive to prepare. The Materials Project is a well-known property database and stores the calculated materials properties based on the structures stored in ICSD [66]. The Materials Project database is prepared to identify the “materials genome” that resembles the human genome that could be conceptualized to represent the gene that dictates their properties and applications. The Materials Project contains structural, thermodynamic, electronic, optical and mechanical properties that are calculated by various materials simulation techniques [25,41,45,67–70]. For clarification purpose, Materials Projects and the related calculated databases based on CSD or ICSD are not focused in this study, but are discussed especially in the areas where the structure-property relationships of energy materials are not well-developed at the moment. Similar to the Materials project, other computed properties databases based on CSD or ICSD have been constructed such as Open Quantum Materials Database (OQMD) [38,71], *ab-initio* electronic transport database [72], MOF databases [11,73,74], JARVIS-DFT database [75], and the Harvard Clean Energy Project [76,77].

## 3. Data mining techniques

### 3.1. Structure descriptors and structure-property relationships

The structure-property relationships of energy materials could be coded via proper selections of structure descriptors for the data mining procedure in structure databases. The atomic structures that could be solved by the well-developed crystallographic techniques include the information of the structural parameters specified in terms of bond length, bond angle, conjugation,  $\pi\cdots\pi$  stacking, intramolecular hydrogen bond (HB), intermolecular HB, molecular weight, packing mode and symmetry that are closely related to the optoelectronic properties of materials [36,62,78–83]. New materials could be mined directly from the structure databases either by a user-ready search interfaces such as *Conquest*, or a code incorporating algorithms describing these searching criteria related to the structure-property relationships.

The bond length is defined as the separation between the two covalently bonded atoms and the number of bonded electrons. The single/double/triple bonds defined by the bond length reveal the strength of interactions between the atoms and the charge transfer characters [84]. Based on the bond length values, the bond length alternation (BLA) values, the harmonic oscillator stabilization energy (HOSE) values, and the quinoidal structures could be determined, while BLA, HOSE and the quinoidal structures are quantitative descriptors to reveal the charge transfer characters [85,86]. The bond angles formed

between multiple atoms across at least two bonds include dihedral/torsion angles where four atoms are involved. The planarity is another structure parameter: and the geometry is planar if all the atoms are embedded on a particular plane [87,88]. Factors such as valency and electronegativity are easily obtained from the structures and have been applied to filter chemically implausible compositions [89].

The hydrogen bond is a moderate bond resulting from an electrostatic attraction between a proton in one molecule and an electronegative atom in the other. The hydrogen bond could be either intramolecular or intermolecular. Similar to the double-helix DNA chain, the hydrogen bond present in materials could greatly influence the structure geometry and stability as well as various properties [90–92]. The  $\pi\cdots\pi$  stacking is another important structural parameter that dictates the crystal formation of many materials. The  $\pi$  bonds in the aromatic rings leads to intermolecular attractive interactions and result in the arrangements in a sandwich, T-shaped or parallel-manner. Unfortunately, there is a lack of unified theory that accurately quantifies the  $\pi\cdots\pi$  interactions. Nevertheless, the  $\pi\cdots\pi$  stacking shows evidence to strongly influence the intermolecular charge transfer and optical properties. The intermolecular interactions including the hydrogen bonds, the  $\pi\cdots\pi$  stacking, halogen bond, van der Waals forces, etc. from various chemical motifs lead to a number of packing modes of crystals that dictates the related electronic/optical properties such as the emission properties and UV–vis absorption properties. The packing efficiency can be determined by the ratios of the total volume and the volume of the unit cell [93–96].

Certain functional groups are especially important for particular applications (vide infra); therefore a more efficient data mining procedure could include special chemical motifs. For example, the search and detailed analysis of the chemical substructure and ligand conformations have been performed to explore the chemical motif candidates [97]. The donor- $\pi$ -acceptor motifs are especially important for the design of solar cells and organic light emitting diode (OLED), and these functional groups could be mined to design new energy materials [62].

Symmetry is considered to be critical for the crystal structure formation since the repeating units arrange and stack upon one another in particular ways such as the rotation or translation according to certain symmetry operations. The symmetry is employed to determine the crystal structures during the structural refinement [98–103]. Also, the symmetry of a crystal has been employed to calculate the thermal conductivity and optical properties. The CSD and ICSD databases contain many other structures information that are less studied such as cocrystals and disorder; these structure parameters could be explored in the future [104].

### 3.2. Quantum descriptors and high-throughput computation

Various quantum descriptors have been proposed to represent physical properties of materials calculated via the Schrödinger equations and related theories such as the band structure and the density of states. The descriptors calculated from the wave function theory include atomic charges, molecular orbital energies including highest occupied molecular orbital (HOMO) and lowest unoccupied molecular orbital (LUMO), energy levels, frontier orbital densities, superdelocalizabilities, dipole moment, polarity indices, polarizabilities and stability [41,69,105–108]. The quantum descriptors are often considered to be more versatile since they better represent the properties, but they are most often more difficult to obtain and time-consuming compared with the structural descriptors. High throughput calculations, usually based on the first principles calculations, have been employed to calculate the above quantities of the materials stored in database, once their crystal structures are known. The calculations are usually followed by procedures to exclude inappropriate structures in CSD and ICSD such as to eliminate duplicates, disorders and inaccurate structures.

### 3.3. Other techniques

The computational cost is high for many materials discovery processes and various methods have been employed to speed up the calculation. Genetic algorithms and machine learning approaches have been employed to accelerate the process [62]. The machine learning technique that uses artificial intelligence automatically extracts predictive models from existing materials data. It has been used to extract meaningful chemical trends from training data [109], for example the prediction of solid-state properties with the local spin-density approximation results as a training set [110–112]. Genetics-based machine learning methods have shown scalability capacity which leads to their capability in the large-scale data mining jobs [113,114]. The machine learning models have been used to guide the materials researches such as solvent choice when crystallising a compound [115], soft materials engineering [116], crystal engineering [115], and classifying microstructures [56,117]. High-throughput calculations could be combined with the machine learning procedures for better materials prediction and validation [118]. Artificial neural networks (ANN) have been used for the prediction of the materials properties, on the basis of their dielectric and ionic properties [119,120].

The fast and efficient analysis of the chemical data is highly desirable [16,62,121,122]. Association analysis is used to reveal the patterns of the data [16]. The cluster analysis is used to discover the correlations of closely related groups [78]. Predictive modelling helps build models for targeted objectives as a function of input. Anomaly detection involves the opposite way by identifying data that differ from the normal observation [16]. The classical force field-based methods could be employed for larger systems. Monte Carlo simulations have been used to calculate the adsorption of methane in 650,000 structures [25,123]. Aggregation of experimental data from the published literature have been employed to allow the creation of interactive databases and additional metadata to be visualized involving energy materials [63,124]. Text mining and visualization tools have been applied to search publication databases [125] including papers, books, patents, and literature compilations [126,127]. Many efforts have been made to prepare databases that extract useful information from the pre-existing results in publications [14].

## 4. Examples of data mining energy materials from structure databases

### 4.1. Data mining dye-sensitized solar cell (DSSC) materials

The Dye sensitized solar cell is a new-generation solar cell that mimic the solar synthesis in the plant and utilize molecular chromophores that reside stably on semiconductor substrates to capture the solar energy and convert into the electrical power [128–142]. The DSSCs require specific dye structures such as the anchoring group and the D- $\pi$ -A structure, i.e., the dyes should have an electron donor, a conjugate  $\pi$  bridge and an electron acceptor. The anchoring moiety is usually a cyanoacrylic acid group. The structural aspects including the chemical substitutions, the conjugation, planarity and bond length alternation (BLA) correspond to the charge transfer characters of the DSSC-active dyes [143].

A series of dye structures that are incorporated in DSSC photoanodes have been solved by the X-ray crystallography [144–147]. Ultra-strong X-rays that are produced by the synchrotron have been used to determine the crystal structures of dyes that present weak diffraction signal and previously structurally unattainable [148–152]. The crystal structures of a series of polyoxotitanate clusters have been solved crystallographically to understand the binding modes of dyes on the semiconductor substrates, with their electronic and optical properties determined upon the modification with the dopants and the ligand structures. Apart from the single-crystal X-ray diffraction techniques, the neutron diffraction, the terahertz spectroscopy, the powder

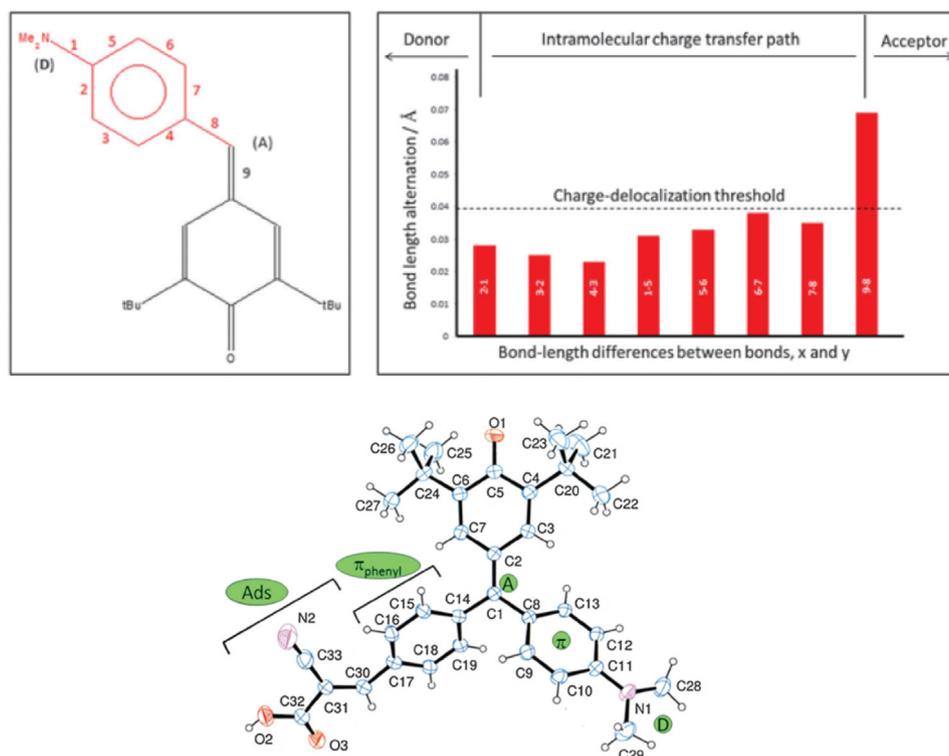


Fig. 1. The bond length analysis involving the bond length alternation (BLA) to filter DSSC-active dye molecules from the CSD database (top) and the new dye identified using the data mining procedure [157]. Reproduced with permission from Ref. [157]. Copyright RSC (2014).

diffraction, NMR, FTIR, as well as other crystallographic and spectroscopic techniques have been involved to understand the structural origin of DSSCs either directly or indirectly [153–155].

Apart from the structure-relationships study on the experimental-proven dyes, the data mining approach has been used to reveal entirely new classes of DSSC dyes, based on the structural analysis of DSSC-active molecular dyes that exhibit certain congruent bond length pattern. Cole et al. employ graph theoretical algorithms and classification tests to discover DSSC-active dyes, using the recursive depth-first, backtracking and graph traversal algorithms (Fig. 1) [156]. The CSD database is crucial since this database contains potential DSSC-active dyes that could be utilized to predict new dyes [157]. In particular, chemical groups and molecular architectures with the D- $\pi$ -A-phenyl-A backbone, that commonly features in a well-performing DSSC dye have been spotted and translated into the generic design principles for the data mining process in CSD. The crystallographic R1 smaller than 0.07 threshold is used for data mining to ensure the accuracy, with all disordered materials, polymers and ionic salts excluded [157]. The screening process involves the BLA analysis, the anchoring group position, the dipole moments calculation and the Hammett constants calculation [158]. A simulated ‘molecular mutation’ with the addition of a cyanoacrylate group is performed on the ‘top 10’ molecular candidate to ensure good electron withdrawing capabilities. Therefore, the screening process relies heavily on the structural descriptors of the DSSC-active dyes.

Apart from the predication of new functional dye molecules, the data mining techniques have also been realized on the photocathode material in a p-type DSSC by Moot et al. [159]. After screening inorganic compounds in ICSD, they identified new photocathode materials such as the lead titanate (PbTiO<sub>3</sub>), a perovskite material, as the promising photocathode material. The search procedure includes the chemical similarity parameters used in cheminformatics, Tanimoto similarity coefficient, and starts from the known p-type photocathodes including NiO, Co<sub>3</sub>O<sub>4</sub>, Cu<sub>2</sub>O, CuI, CuAlO<sub>2</sub>, CuGaO<sub>2</sub>, NiCo<sub>2</sub>O<sub>4</sub>, and ZnCo<sub>2</sub>O<sub>4</sub> as reference query materials [159].

Since the bond lengths stored in the structure databases are closely related to the intramolecular charge transfer properties in dye molecules, analysis has been carried out on retrieving the bond lengths of different organic dyes to understand the intramolecular charge transfer characteristics [160,161]. For example, the bond length values of the azo group stored in CSD are compared to reveal the efficient charge transfer characters in azo dyes toward the DSSC application [162]. Identifying suitable chemical fragments is important for the DSSC materials design. Donor groups have been screened [163], and DSSCs with novel structures have been obtained [164,165]. Phenothiazine-based dyes have been assembled from fragments in a synthetically tractable manner, with predicted PCEs over 9% [166]. In order to help users analyse the DSSC structure data more efficiently, a DSSC database ‘DSSCDB’ has been generated to afford researchers with information from literatures on the detailed dye structures including triphenylamines, carbazoles, coumarins, phenothiazines, ruthenium and porphyrins [167].

#### 4.2. Data mining perovskite solar cell materials

Perovskite solar cells have their device structures originating from dye-sensitized solar cells, and the perovskite solar cell research could be dated back to 2009 when Miyasaka [168] et al. substitute dye molecules in DSSCs with halide perovskites, and found a power conversion efficiency above 3%. By changing the liquid electrolyte into the solid state, as well as additional optimization of the perovskite crystal formation process, the power conversion efficiencies of perovskite solar cells rocket to 22% [169–200]. Despite of the astonishing development on power conversion efficiency, the origin of the superiority of the halide perovskite materials as well as the detailed mechanisms happening inside the crystal environment or at the interfaces still remains elusive. For example, the halide perovskite structures for solar cell applications have been accurately solved except the disordered cation molecules, and the exact roles of the methylammonium cation are debatable [201,202].

There have been researches carried out to data mine the new halide perovskite materials and similar materials that can replace the traditional lead halide perovskites in solar cells, which suffer from the lead contamination and instability issues [69,159,203–205]. The new materials that are mined from structure database should possess the small effective masses of electrons and holes, the long charge diffusion length, outstanding optical properties and small charge recombination possibilities, all of which could be calculated bases on the ICSD databases. Perovskites could have multiple organic and inorganic constituents ( $\text{CH}_3\text{NH}_3$ ,  $\text{HC}(\text{NH}_2)_2$ , Cs, Rb, Pb, Sn, I, Br), and various new structures with layered, double, and metal-deficient perovskites, which naturally give flexibility for rational strategies to replace the current experimental trial-and-error-based approaches [203].

The ICSD database has been successfully exploited to identify potential lead-free hybrid perovskites, with the stoichiometry as the starting point to identify possible lead free perovskites. The halide moiety is critical in the functioning of the halide perovskite materials. Using the CSD database, the strength and geometric preferences of non-covalent interactions in the halide polymers have been investigated [206–214]. The structures of the halide polymer with different dimensions could be used for data mining new halide perovskites for solar cells. The 3D halide perovskites are not unique structures for the perovskite solar cells. Replacement of lead with di-, tri- and tetra-valent cations has been performed, and iodosalts, dimers, layered structures, and mixed and multi-dimensional perovskites have been employed (Fig. 2);  $12\text{A}^{1+}$  (e.g.,  $\text{MA}^+/\text{FA}^+/\text{Cs}^+$ ),  $27\text{M}^{2+}$ ,  $35\text{M}^{3+}$ , and  $25\text{M}^{4+}$  site cations could be chosen; three halogens ( $\text{I}^-$ ,  $\text{Br}^-$ ,  $\text{Cl}^-$ ) and a mixture of these three elements could be chosen. For example, the divalent element could be Ge, Sn, Bi, Ag and In. The tetravalent metallic cations such as Sn in  $\text{Cs}_2\text{SnI}_6$  has been tried, while the double perovskites ( $\text{A}_2^+\text{M}^+\text{M}^{3+}\text{X}_6$ ) uncover the opportunity of including  $3^+$  cations into the perovskite structures [203]. Taking into consideration of the cationic valence states and volume ratios, possible perovskites combining entries from ICSD databases and chemical intuition are  $> 24,138$  for  $\text{AMX}_3$ ,  $> 31,290$  for  $\text{A}_3\text{M}_2\text{X}_9$ ,  $> 22,350$  for  $\text{A}_2\text{MX}_6$ , and  $> 9 \times 10^6$  for  $\text{A}_2\text{MM}'\text{X}_6$  [203].

Efforts have been made to find that ionic radii, tolerance factor, and octahedral factor are the important structural descriptors for the perovskite halides using a support vector machine classification model. With the training model, several novel perovskite structures have been predicted [203,215,216]. Mechanically stable and electronically suitable electrons and holes extracting contacts materials for  $\text{CH}_3\text{NH}_3\text{PbI}_3$  have been selected according to the electron affinity, ionisation potential, lattice parameters and crystal structure criteria. Potentially effective new hole and electron transporting layers according to these descriptors include  $\text{Cu}_2\text{O}$ ,  $\text{FeO}$ ,  $\text{SiC}$ ,  $\text{GaN}$ , and  $\text{ZnTe}$  [205].

#### 4.3. Data mining other solar cell materials

Similar to those in DSSCs and perovskite solar cells, proper descriptors are needed to identify candidate solar cell materials. Starting from the ICSD database, alternative polar chalcogenides have been predicted to substitute the widely studied CIGS material [41]. In the study, the stoichiometry with the  $\text{AB}_2\text{CX}_4$  chemical formula is targeted, where  $\text{A} = \text{Li}$ ,  $\text{Cu}$ ,  $\text{Ag}$ ;  $\text{B} = \text{Zn}$ ,  $\text{Cd}$ ;  $\text{C} = \text{Ga}$ ,  $\text{In}$  and  $\text{X} = \text{O}$ ,  $\text{S}$ ,  $\text{Se}$ ,  $\text{Te}$ , leading to a family of 48 members [41]. The criteria could be strong optical absorption coefficients, a proper bandgap value such as  $\sim 1.3\text{eV}$  satisfying the Shockley–Queisser limit, low cost and compatibility with existing technologies. A set of high SLME materials are identified based on ICSD, including the best already known thin-film solar absorbers, such as  $\text{CuInSe}_2$ ,  $\text{CuGaSe}_2$  and  $\text{CuInS}_2$ , while materials different with the 1:1:2 stoichiometry (for example,  $\text{Cu}_7\text{TeS}_4$ ,  $\text{Cu}_3\text{TeS}_2$  and  $\text{Cu}_3\text{TeSe}_2$ ) are also identified [217]. Other approaches involving ordinary least squares, sparse partial least squares, elastic net/least absolute shrinkage, selection operator regression methods coupled to rough set and principal component analysis methods have been developed for the

band gap prediction of new chalcopyrite compounds [218]. A data mining workflow have been developed and applied to the analysis of solar cell libraries based on titanium and copper oxides [26]. Constituent elements have been focused to find new thin film solar cell materials. Thermodynamic stability, electrical transport, electronic structure, optical and defect properties have been evaluated to predict the  $\text{Cu}_2\text{SnS}_3$  as the promising photovoltaic materials [219].

High-throughput calculations are most often used to identify new solar cell materials [28,220]. The Materials Project database that provides relaxed structures and related optoelectronic properties are naturally selected as the starting point to identify solar cell materials. For example in the area of the transparent p-type oxide, 3600 quaternary oxides in the Materials Project database are searched and the materials with a small effective mass are obtained according to their band structures [221]. Phosphides and zinc blende boron phosphide BP are found to be promising p-type transparent conducting materials (requiring large band gap for transparency and low hole effective mass for high mobility) by exploiting the weak absorption for indirect optical transitions [222]. The CSD database stores a large amount of organic structures that could be utilized to design organic electronics. Apart from the low carrier mobilities in organic semiconductors, the electronic couplings and intramolecular reorganization energies are identified to be the two main descriptors for charge mobility, and materials with long-range charge percolation pathways are identified [28]. In the ICSD, for every nitride (3008 total entries) there are more than ten oxides (41,529 total entries) [223]. As a consequence, new nitride materials with structural motifs and crystal structures that have never been reported previously are discovered with good regularity [223]. Similar electronic structure databases and initiatives include the JARVIS-DFT database [75] and the Harvard Clean Energy Project (CEP) toward the new organic solar cell materials [76].

#### 4.4. Data mining water splitting materials

Water splitting materials employing the photocatalyst to harvest the solar energy and produce hydrogen fuel from the water provides an effective alternatives to fossil fuels [176,224–228]. The photocatalysts that split water have been searched from ICSD [229,230]. For example, by analysing the bulk metals and the most stable single- and bi-metal oxides in ICSD and the Materials Project databases, new water splitting materials are proposed [169], based on the descriptors of appropriate band gap, band edges relative to the water redox levels, high mobilities, and chemical stability under light irradiation [169]. 10 oxides and 5 oxynitrides including  $\text{AgNbO}_3$ ,  $\text{BaSnO}_3$ ,  $\text{BaTaO}_2\text{N}$ ,  $\text{CaTaO}_2\text{N}$ ,  $\text{SrTaO}_2\text{N}$ , and  $\text{LaTiO}_2\text{N}$  are identified as the candidates for light harvesting materials, which agree with the literature [169]. Oxynitrides and compounds containing d10 cations ( $\text{Ga}^{3+}$ ,  $\text{In}^{3+}$ ,  $\text{Ge}^{4+}$ ,  $\text{Sn}^{4+}$ ,  $\text{Sb}^{5+}$ , and  $\text{Bi}^{5+}$ ) or d0 cations ( $\text{Ti}^{4+}$ ,  $\text{Zr}^{4+}$ ,  $\text{Hf}^{4+}$ ,  $\text{V}^{5+}$ ,  $\text{Nb}^{5+}$ ,  $\text{Ta}^{5+}$ ,  $\text{Cr}^{6+}$ ,  $\text{Mo}^{6+}$ ,  $\text{W}^{6+}$ ,  $\text{Sc}^{3+}$ , and  $\text{Y}^{3+}$ ) are targeted in ICSD. To solve the problems of insufficient oxynitrides materials available in the ICSD, some new candidates are proposed to be proper photocatalysts, including three binary nitrides, two ternary oxynitrides and eleven quaternary oxynitrides [229]. The detailed procedure starting from the ICSD structure search and the follow-up electronic property calculations, including the energy level position match and the band gap screen, are summarized in Fig. 3.

In order to identify potential water splitting materials, calculations have been performed using the Open Quantum Materials Database (OQMD) [204]. 139 materials are identified as potential new candidates for thermochemical water splitting (TWS) application, including  $\text{CeCoO}_3$  and  $\text{BiVO}_3$  which are not observed previously, using the large data set of compounds containing stabilities, oxidation states, and ionic sizes [204]. High-throughput screening of materials based on the electronic band gap calculations have been performed, which leads to the discovery of five candidates as the promising new water splitting material [231]. The oxygen evolving reaction (OER) activities of

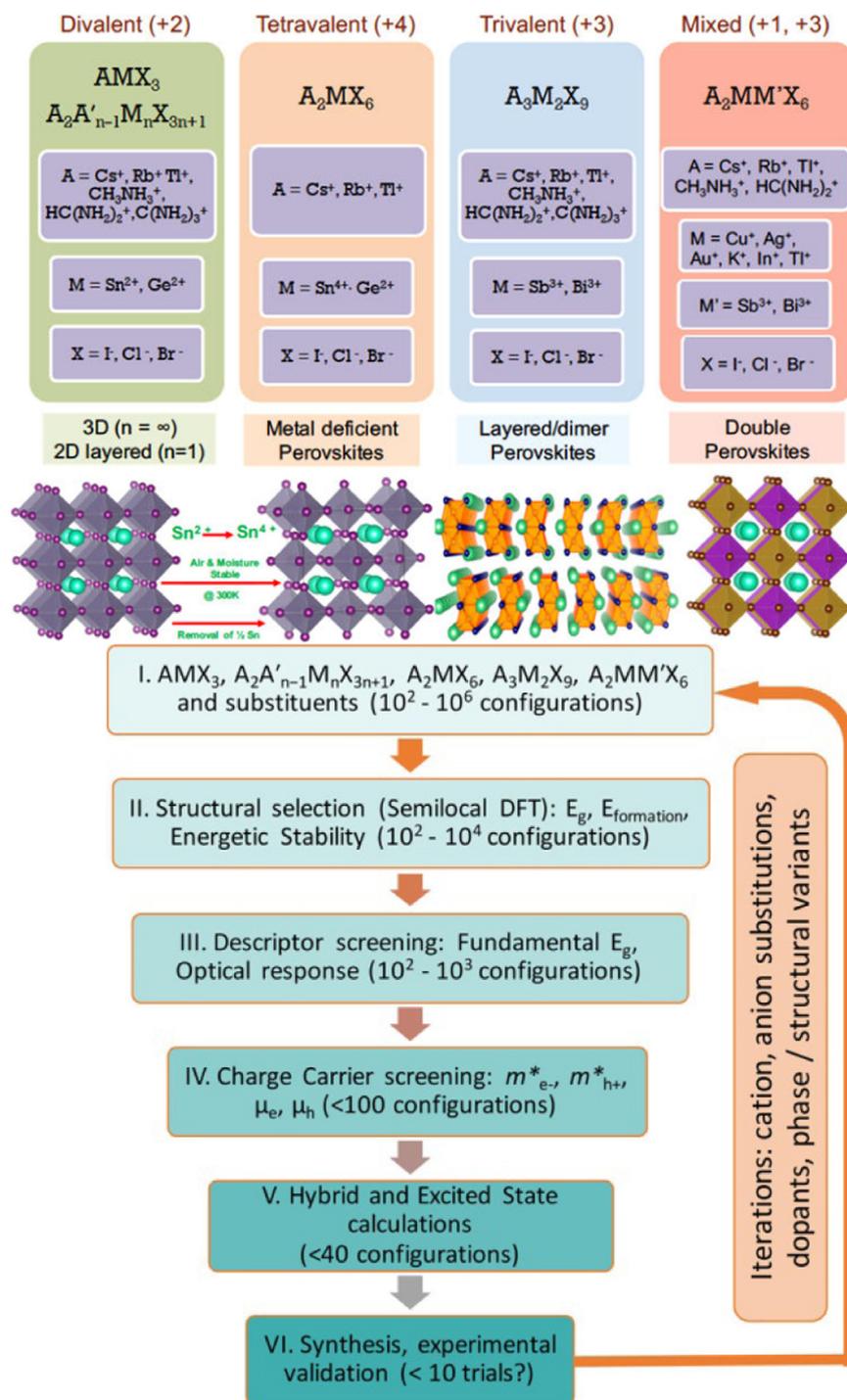


Fig. 2. Perovskite structure with different valence and elements (top) and rational design flowchart starting with the ICSD database and subsequent computational screening and experimental validation (bottom) [203]. Reproduced with permission from Ref. [203]. Copyright ACS (2017).

perovskites are explored according to the descriptors of the metal–oxygen bond strength using factor analysis and linear regression models. Electron occupancy and metal–oxygen covalency are identified as the dominant influences on the OER activity, and multiple descriptors required for better prediction [232]. The associations between composition and catalytic activity are examined for OER for 5429 catalyst compositions in a (Ni–Fe–Co–Ce)<sub>x</sub> library, based on informatics-based clustering of composition property functional relationships [233].

#### 4.5. Data mining organic light emitting diode materials

The data mining process of the OLED molecules based on the structural descriptors are not well-developed compared with the quantum descriptors. Nevertheless, considering the structural similarity in terms of the D-π-A backbone between the OLED materials based on molecules and the DSSC dyes, a similar data mining process borrowing from the DSSC concept could be employed in the future to identify new OLED functional molecules. Compared with the DSSC materials, the OLED materials have larger molecular weight and do not require anchoring groups. The search of OLED molecules could also be based on

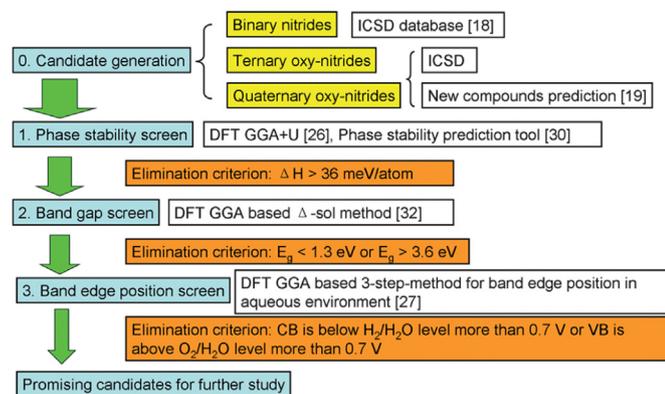


Fig. 3. The searching procedure of water-splitting photocatalysts starting from ICSD [229]. Reproduced with permission from Ref. [229]. Copyright RSC (2013).

completely new fragments. According to the HOMO and LUMO positions and the optical properties, a new set of fragments have been screened [62].

#### 4.6. Data mining thermoelectric materials

The thermoelectric materials generate electricity from thermal energies. Various descriptors have been tried to search the potential thermoelectric materials. Thermoelectric power factors of  $> 3000$  inputs in ICSD are compared [69,234], while a metric to provide a consistent evaluation of the inherent physical properties that dictate the thermoelectric figure of merit  $zT$  is proposed [235,236]. After searching for new thermoelectric materials in the ICSD based on the 570 Sb-containing compounds, Zintl compound  $\text{LiZnSb}$  are suggested as potentially interesting n-type thermoelectric materials [237]. Other descriptors include Lorenz number that lead to different  $zT$  enhancement, and the computational search for materials with exceptionally low Lorenz number and high thermoelectric quality factor has been carried out [24]. Based on over 100 publications, databases storing thermoelectric materials have been created [238], which includes 18,000 data points with associated properties measured at several temperatures. The interplay between data mining, informatics, and machine learning approaches allow for a new paradigm in thermoelectric materials development [63,239,240]. A data-driven approach is accomplished by understanding discrete scalar descriptors regarding the crystal and electronic structure and the Curie temperature, and quantitatively exploring the different materials descriptors.

#### 4.7. Data mining lithium battery materials

The Materials Genome Project coupled to ICSD has been data mined to design lithium battery materials where the electrodes consist of inorganic compounds [15,238]. The data-driven approach has proved to be successful to the rational design of Li-ion battery materials [241], since the energy density, power density, discharge capacity, lithiation potential, capacity retention upon cycling and many others could be accurately calculated for the compounds in ICSD [220]. The organic part in the lithium battery such as the electrolyte solvent has also been screened; the preferred electrochemical stability, melting and boiling temperatures, viscosity, dielectric constant, Li-ion conductivity,  $\text{Li}^+$  diffusion, electronic conductivity ranges and activation barriers have been calculated in the presence of the organic molecules [62,242,243].

#### 4.8. Data mining gas capture and storage materials

Data mining techniques have been employed to search the gas capture and storage materials, which are predominantly based on

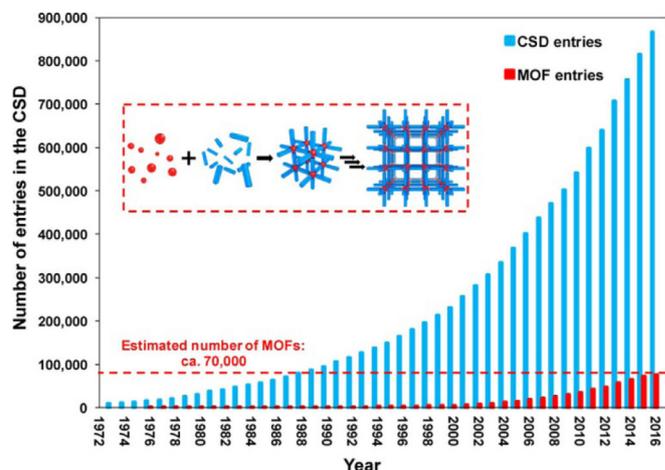


Fig. 4. The increase of MOF entries since 1972 in comparison with the CSD entry. The inset shows the structures of the MOF that is self-assembled by the following moieties: metals (red) and organics (blue) [250]. Reproduced with permission from Ref. [250]. Copyright ACS (2017). (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article).

metal-organic frameworks (MOF) owing to their structural integrity and porosity for molecular adsorption [97,244–248]. These MOFs are mainly for carbon capture and storage, gas separation, as well as hydrogen fuel storage [12,73,74]. The MOFs are essential hybrid systems that are built from the blocks and frameworks consisting of either organic or inorganic moieties. Since CSD stores various synthesis-ready MOF crystal structures and the synthesis of MOF is not straightforward, screening MOFs from the synthesis-ready CSD databases has been found efficient in the design process [11,12,249]. In CSD, the MOF entries have risen up to 6000 in CSD in 2011, and risen to 70,000 in CSD in 2016 (Fig. 4), which means the CSD database is a natural database selection for the carbon capture and storage as well as the hydrogen storage [250].

In order to identify available MOF structures in CSD, researchers have searched for structures with bonds between metallic elements and organic elements, while additional structural screening procedure include the solvent molecules, varying degrees of disorder, missing H atoms and overlapping atoms [22,73,74]. Databases with MOFs free from solvents/disorders and with a pore limiting diameter (PLD) larger than  $2.4 \text{ \AA}$  structures have been provided; the optimal pore size that maximizes net capacity such as  $6\text{--}10 \text{ \AA}$  and void fractions of  $0.1\text{--}0.5$ , have been set to the optimal criterion for the MOF search [12].

Potential substitutes with possible ligands has been selected according to the chemical substructure searching and ligand conformational analysis [97]. Pores larger than  $3.25 \text{ \AA}$  for  $\text{CO}_2$  adsorption are selected based on the possible zeolite-like  $\text{SiO}_2$  structures starting from 230 space groups, various unit cell dimensions and densities [251]. Regarding the oxygen uptake, the relationships between structural properties and oxygen adsorption performance at dissimilar pressures have been studied in five dimensions, and UCM-152 that delivers 22.5% more oxygen has been obtained [249]. NU-12528 is identified from 10,000 hypothetical MOFs and confirmed by experimental synthesis and adsorption measurements [252]. An efficient approach automated assembly of secondary building units (AASBU) has been developed [253], with the building blocks randomly distributed at points, and provides insights for the topological preferences [22]. The methane uptake in over 650,000 materials has been investigated starting from both existing and predicted nanoporous materials [123]. The propylene/propane adsorptive separation have been investigated regarding their selectivity and working capacity, via structural descriptors such as the  $\text{N}_2$  surface area, accessible surface area of propane, and PLD. Porous materials that includes elements such as In, Te, Al, and

I, along with the low LCD stipulation has been proposed owing to high crossover distance [254]. In addition, neural networks and grand-canonical Monte Carlo simulations, coupled with Lorentz-Berthelot mixing rules and Lennard-Jones parameters have been employed for the H<sub>2</sub> storage and methane adsorption [12,123]. Apart from CSD, a number of databases for MOFs are available such as Hypothetical MOF Database, MOF-5 Analogues, Reticular Chemistry Structure Resource and porous materials databases including Zeo++ , Poreblazer, MOFomics, TOPOS, CoRE MOF and MOFIA [11,22,251].

#### 4.9. Data mining other materials

The data mining technique has generated new insights in various new materials [78,255]. Piezoelectric materials convert mechanical energy into electricity, and the ICSD database has been examined to find previously unrecognized piezoelectric systems from half-Heusler semiconductor compounds, with 987 candidates scanned in terms of the structural, dielectric, insulating and piezoelectric properties [256]. Previously unknown two-dimensional materials have been identified starting from CSD and the first principles calculations [69,257]. A novel data mining algorithm involving the dimensionality of weakly bonded subcomponents has been employed in the Materials Project database, and 325 materials are predicted to exhibit piezoelectric properties [78]. Topological insulators have been data mined by examining the changes in band inversion in the presence or absence of spin-orbit coupling [258]. Stable Dirac-point node are predicted via the combined group theory and data mining approach using the Organic Materials Database, and particular space groups such as P212121 is established to be positive to the Dirac nodes construction [259]. Data mining using density of states similarity search in OMDB is performed to identify novel organic High-Tc Superconductors [260]. The relative stability and the relative solubility of cocrystals has been investigated by a data mining force field in CSD [104]. Soft-matter materials is also assisted by the data-driven approaches [261,262].

### 5. Suggestions and outlook

Although the structure descriptors are sometimes considered to be lower-level quantities than the quantum descriptors, they offer a more direct and faster data mining process than the quantum descriptors to identify new energy materials, while the quantum descriptors often requires more expensive computational resources. The data mining procedures rely heavily on the structural details of energy materials.

Finding a proper algorithm and structural descriptor is critical for the efficient screening. Data mining new energy materials from the CSD and ICSD databases can be accelerated using proper structural descriptors, which describe the crystal structures that could be solved by the well-developed crystallographic methods. The structural parameters such as bond length, bond angle, conjugation,  $\pi \dots \pi$  stacking [36,78–82], intramolecular/intermolecular hydrogen bond, molecular weight [62], packing mode and symmetry [83] are closely related to the targeting applications. There can be hundreds of structure descriptors representing certain properties of the materials, and the individual structure descriptors could be artificially combined to form new structure descriptors, owing to the newly developed machine learning techniques. Many experimental researches have given insights on the chemical motifs that could be particularly popular in certain type of devices. For example, the chemical substructure searching, the donor- $\pi$ -acceptor motif probing, and the ligand conformational analysis are effective to select potential materials. Particular chemical moieties such as amine, triphenyl amine and cyanoacrylic acid group are well-correlated with certain applications. The bond length and angles, the intermolecular interactions, stoichiometry, hydrogen bonds (HBs), intermolecular distance, BLA pattern, quinoidal structures, packing modes, halogen bond and space group should be explored further in relation to particular applications. Many crystal structures of proteins

are stored in databases such as the Protein Data Bank (PDB) [44–48] and PDB-derived databases, which might provide additional structure information on the new biological materials, such as the DNA- and protein-based energy devices. It will not be surprising that with the development of the new devices that requires integration of the energy materials into the biological systems such as the skin, the protein-based structure database might be very useful for the development of new energy materials in the future.

Some of the design procedures of energy materials follow an interesting “Lego” type style where suitable chemical subgroups with desired quantities such as the Hammett values are combined in a particular way [263,264]. However, the Hammett values alone only vaguely estimate the electron donating/withdrawing abilities of the materials and many Hammett values of new functional groups are unavailable [86,265]; therefore improvements in the Hammett value determination and more advanced descriptions other than the Hammett values are called for. It would be desirable to develop parameters to better quantify the electron/withdrawing abilities for particular applications.

While most of the structural details have been revealed from the structure-property relationships investigation in the literature, some structural impacts still remain elusive at the moment; for example, the disorder (happening in halide perovskite solar cells) and cocrystals that are different from the crystals of pure components that are considered to be useless previously might be re-investigated [104].

The ever-increasing data mining studies from the structure databases require more crystallographic contributions that are dedicated to obtain the accurate structure information. X-ray crystallography continues to be the undisputed primary technique [266] for picturing molecules and assemblies with the utmost accuracy compared with the in-silico crystal structure prediction, because the structure prediction of even the simplest crystalline solids only starting from the composition is still fundamentally questionable. For example, in many case a possible space group should be set in advance for the structure prediction, which might not be correct and brings out inaccuracies and inconsistencies [69,267,268]. Databases storing materials properties from the Schrödinger equations still depends on the accurately crystal structures inputs in databases, and the crystal structure determination and the structure-property relationships formulation are considered be of paramount importance to realize better design of energy materials in the future.

Since the excited states and the time-resolved dynamics are particularly important for many energy conversion applications, it could be very useful to obtain the time-resolved structure of energy materials with both atomic and electronic structures determined in the time-domain, such that new mechanisms could be revealed and new materials could be mined from the more accurate structure-properties relationships based on the time-resolved structures. To this end, there are evidences showing that the time-resolved structures of energy materials are achievable via the time-resolved crystallography techniques, which provides the ultra-strong X-rays produced by the synchrotron to determine the crystal structures of new materials with weak signal that are previously unattainable [148–152,269–282]. For example, the time-resolved structures (in nanosecond or picosecond) of several silver- and copper-based organometallic compounds have been reported are reported to be closely related to the optoelectronic properties of the materials especially the UV–vis absorption and the light emission [149], which are critical factors for the solar cells and OLEDs. The extent of the displacement in angstrom of the atoms and the electrons upon the excitation are suggested to be directly associated with the electronic and optical properties, with the larger displacement leading to stronger emission [150,151,270,277,283]. Nevertheless, how the time-resolved structures are related to the energy-related device performance quantitatively remain elusive at the moment, considering the fact that many processes related to the electron transfer and light absorption happen in different time-scales [102,284–287]. In

addition, the energy gap, defects, and the energy level alignment significantly affect the time-domain activities of the energy materials. A formulation of the time resolved structure-property relationship would be thus highly desirable (for instance the differences of the structural changes in femto-, pico-, nano-, micro and milli-second on the optoelectronic properties) to guide the materials design, which emphasize the importance of the accurate determination of materials structures in various timescales using the ultra-strong X-rays or neutron pulses.

## 6. Conclusions

Data mining energy materials from the structure databases such as CSD and ICSD have been facilitated by the formulation of proper structure-property relationships, and successful algorithms coded with the structural descriptors that consider the structure-property relationship have been rapidly developed to facilitate the data mining process. The structural descriptors could be the chemical moiety, bond length and angles, intermolecular interactions, stoichiometry, valency, elements, ionic substitutions, surface area, pore diameter, HBs, halogen bonds intermolecular distance, BLA pattern, quinoidal structures, stoichiometry, packing modes and space groups, etc. There can be hundreds of structure descriptors representing certain properties of the materials, and the new structure descriptors could be designed via the machine learning techniques to obtain more proper structure descriptors. The data mining study of new energy materials from the crystal structure databases such as CSD and ICSD has been found to be applicable in solar cells, water splitting systems, lithium batteries, thermoelectric devices, piezoelectric materials and gas adsorbent materials. Continuous efforts should be spent on the basic understanding of structure-property relationships of new energy materials, since a bonus of the structure-property relationship study is that new energy materials with targeting properties could be identified quickly and accurately from the structure database by algorithms that include the structure-property relationships. It is believed that with more crystal structures obtained from the X-ray and neutron crystallographic techniques, and a more unified theory of structure-property relationship of energy materials toward particular applications, a Google-like approach based on the data mining process could be realized in the future to design energy materials.

## CRediT authorship contribution statement

**Lei Zhang:** Conceptualization, Data curation, Formal analysis, Funding acquisition, Investigation, Methodology, Project administration, Resources, Software, Supervision, Validation, Visualization, Writing - original draft, Writing - review & editing. **Zhiqiao Chen:** Data curation, Formal analysis. **Jing Su:** Funding acquisition, Resources. **Jingfa Li:** Funding acquisition, Resources.

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## Author contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

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