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Biological monitoring: lichens as bioindicators of air pollution assessment — a review

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“Capsule”: *Lichen species may be used as biological monitors for environmental prevention programs.*

Abstract

Often as part of environmental impact studies and, above all, to obtain authorisations in accordance with prescriptions from the Ministry for the Environment (Italy), surveys and controls that use biological indicators are required. This is because such indicators are valid instruments for evaluating the quality of the air ensuing from the subject (often an industrial plant) of the Environmental Impact Assessment (EIA). In this context, this paper aims to analyse some of the theoretical aspects of biological monitoring and to provide a progress report on the use of lichens as bioindicators of air quality, with a particular eye to the situation in Italy. The object of this paper is that of pointing out the most important lines in the current state of knowledge in this field, evaluating the methodological applications and their advantages/disadvantages with respect to traditional surveying methods. © 2001 Elsevier Science Ltd. All rights reserved.

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1. Biological monitoring

The use of cosmopolite organisms to assess pollution has developed notably during the last few decades. Such organisms assume environmental contaminants and may be used as indicators of the bioavailability of a given contaminant over time, allowing, in certain cases, comparison between contamination levels in geographically different areas.

It is in this context that OECD countries have taken many initiatives for examining potentially dangerous products by proposing general programmes for the monitoring and evaluation of environmental impact (Tessier et al., 1980; Connell, 1986; Herman, 1987; Krumgalz, 1989; Bero and Gibbs, 1990).

From an ecotoxicological perspective, we can consider as contaminants or producers of environmental stress, all chemical compounds that are fundamentally released into the environment as a result of human activities, and which cause damage to living organisms (Moriarty, 1999).

In general, bioindicators are organisms that can be used for the identification and qualitative determination of human-generated environmental factors (Tonneijk and Posthumus, 1987), while biomonitors are organisms mainly used for the quantitative determination of contaminants and can be classified as being sensitive or accumulative.

Sensitive biomonitors may be of the optical type and are used as integrators of the stress caused by contaminants, and as preventive alarm systems. They are based upon either optical effects as morphological changes in abundance behaviour related to the environment and/or upon chemical and physical aspects as alteration in the activity of different enzyme systems as well as in photosynthetic or respiratory activities.

Accumulative bioindicators have the ability to store contaminants in their tissues and are used for the integrated measurement of the concentration of such contaminants in the environment. Bioaccumulation is the result of the equilibrium process of biota compound intake/discharge from and into the surrounding environment.

The first studies of bioindicators date back to the 1960s. Beginning with the theoretical calculations of

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Stöcker (1980) and Phillips (1977, 1980), we can define the main characteristics of a bioaccumulator.

Bioaccumulators must:

1. accumulate the pollutant without, however, being killed by the levels with which it comes into contact;
2. have a wide geographical distribution;
3. be abundant, sedentary, or of scarce mobility, as well as being representative of the collection area;
4. be available all year round and allow for the collection of sufficient tissues for analysis;
5. be easy to collect and resistant to laboratory conditions, as well as being usable in laboratory studies of contaminant absorption, if necessary;
6. have a high concentration factor for the contaminant under study, and thus allow direct analysis with no prior increase in concentration;
7. have a simple correlation between the quantity of contaminant contained in the organism and the average contaminant concentration in the surrounding environment; and
8. have the same contaminant content level correlation with the surrounding environment in every site studied and under any condition. This must be true for all organisms examined.

2. The problems of biomonitoring

For a variety of reasons, it is of fundamental importance to define the reference levels for pollutants in an ecosystem when making biological monitoring studies to:

1. evaluate the state of conservation or degradation;
2. predict the incidence of possible future human activities in order to establish the necessary interventions; and
3. control evolution over time, using monitoring programmes, if necessary.

To correctly evaluate the degree of contamination in an ecosystem, or to carry out biomonitoring operations, it is necessary to first establish the background level of the contaminant, both in the environment (air, water, soil), and in the organisms. The background level may be interpreted in different ways: it may be understood as a pre-industrial level (prior to any human activity); as a natural level (the average conditions of an area or a region where there may be human activity, but which is in a good state of conservation); a standard level (based upon global geographical references); or even a zero level (the concentration of an element in the environment or in an organism prior to the development of a particular activity that is independent of the degree of conservation; Carballeira et al., 2000; Cecchetti and Conti, 2000).

Once the background level has been established according to Carballeira et al. (2000), the contamination factor may be used to evaluate the state of conservation of an ecosystem, or to monitor its state. This is the relationship between the level of a contaminant found in the biota or environment and a reference value that represents a determined stage (pre-industrial, natural, zero):

$$CF_b = C_b/BL_b \quad \text{or} \quad CF_a = C_a/BL_a$$

where CF = the contamination factor for the biota (_b) or the environment (air, water, soil) (_a); C = the concentration of contaminant in the biota (_b) or in the environment (_a), respectively; BL = the background level of the pollutant in the biota (_b) or in the environment (_a), respectively.

If the background level is a reference of the zero phase, it will allow us to observe the evolution of a pollutant (in terms of both space and time), during a contamination process. This concept may also be used to observe the decontamination rate in an ecosystem (positive impact).

The CF is said to have been corrected when:

$$CF_{\text{corrected}} = C_{b2}/C_{b1}$$

where C_{b1} = the concentration of pollutant present in the biota at the time or the point 1; C_{b2} = the concentration of pollutant present in the biota at the time or the point 2;

This signifies that the CF has been corrected when, during comparisons between different environmental situations, no data is available for the background levels (BL).

The system for environmental classification is realised by starting with the *Contamination Factors* obtained for each contaminant present in the environment or organisms. When evaluating the CFs obtained, it is also necessary to take into account the uncertainties that derive from the following: sampling; space and time variations for the samples; the age and condition indexes of the organisms, etc. In general, a CF that is above a given number (generally 1.5, 2, or 3 times the BL), is taken to be the minimum level under which it is no longer possible to refer to certain contamination. The qualification of a contamination situation may follow a linear scale, or, in high-level pollution conditions, a scale of the exponential type.

3. Lichens as bioindicators of air pollution

Lichens are considered the result of a symbiotic association of a fungus and an alga. More precisely the term “alga” indicates either a Cyanobacteriae or a

Chlorophyceae; the fungus is usually an Ascomycetes, although on rare occasions it may be either a Basidiomycetes or a Phycomycetes.

In this association, the alga is the part that is occupied with the formation of nutrients, since it contains chlorophyll, while the fungus supplies the alga with water and minerals. These organisms are perennial and maintain a uniform morphology over time. They grow slowly, have a large-scale dependence upon the environment for their nutrition, and, differently from vascular plants, they do not shed parts during growth. Furthermore, their lack of cuticle or stoma means that the different contaminants are absorbed over the entire surface of the organism (Hale, 1969, 1983).

As far back as 1866, a study was published on epiphytic lichens used as bioindicators (Nylander, 1866). Lichens are the most studied bioindicators of air quality (Ferry et al., 1973). They have been defined as “permanent control systems” for air pollution assessment (Nimis et al., 1989).

During the last 30 years, many studies have stressed the possibility of using lichens as biomonitors of air quality in view of their sensitivity to various environmental factors, which can provoke changes in some of their components and/or specific parameters (Brodo, 1961; Rao and LeBlanc, 1966; Schönbek, 1968; Hawksworth, 1971; Gilbert, 1973; Mendez and Fournier, 1980; Lerond, 1984; St Clair and Fields, 1986; St Clair et al., 1986; Galun and Ronen, 1988; Showman, 1988; Nimis, 1990; Oksanen et al., 1991; Loppi et al., 1992a; Seaward, 1992, 1996; Halonen et al., 1993; Gries, 1996; Loppi, 1996; Hamada and Miyawaki, 1998). For indeed, many physiological parameters are used to evaluate environmental damage to lichens, such as: photosynthesis (Ronen et al., 1984; Calatayud et al., 1999); chlorophyll content and degradation (Kardish et al., 1987; Garty et al., 1988; Balaguer and Manrique, 1991; Zaharopoulou et al., 1993); decrease of ATP; variations in respiration levels (Kardish et al., 1987); changes in the level of endogenous auxins; and ethylene production (Epstein et al., 1986; Garty et al., 1993).

Furthermore, laboratory exposure to SO₂ causes relevant membrane damage to lichen cells (Fields and St Clair, 1984). Many studies show a positive correlation between the sulphur content of lichens and SO₂ present in the atmosphere (Takala et al., 1985; Rope and Pearson, 1990; Silberstein et al., 1996).

Various authors report that the concentration of chlorophyll *a+b* is altered by vehicle traffic pollution (LeBlanc and Rao, 1975; Ronen and Galun, 1984; Carreras et al., 1998), and by urban emissions (Zambrano and Nash, 2000). In general, lichens that are transplanted into areas with intense vehicle traffic show an increase in chlorophyll *a+b* concentration that is proportional to increases in emissions. Such effects are generally caused by traffic emissions and in particular,

sulphur and nitrogen oxides. In areas with intense vehicle traffic and elevated levels of industrial pollution, high values are obtained for Chl *b*/Chl *a* ratios.

Air traffic, and in particular the effects of kerosene and benzene, seems to have a lesser effect on the lichen population than vehicle traffic. This has been demonstrated in a study of Hamburg airport (Rothe and Bigdon, 1994).

Lichens may be used as bioindicators and/or biomonitors in two different ways (Richardson, 1991; Seaward, 1993; Gries, 1996):

1. by mapping all species present in a specific area (method A); and
2. through the individual sampling of lichen species and measurement of the pollutants that accumulate in the thallus; or by transplanting lichens from an uncontaminated area to a contaminated one, then measuring the morphological changes in the lichen thallus and/or evaluating the physiological parameters and/or evaluating the bioaccumulation of the pollutants (method B).

4. Lichens in the control of environmental contamination

4.1. The index of atmospheric purity (IAP)-method (method A)

The compositional changes in lichen communities are correlated with changes in levels of atmospheric pollution. The application of method A allows the elaboration of an IAP. This method (LeBlanc and De Sloover, 1970) makes it possible to map out the quality of the air in a determined area. The IAP gives an evaluation of the level of atmospheric pollution, which is based on the number (*n*), frequency (*F*) and tolerance of the lichens present in the area under study. There are twenty different formulae for IAP calculation, and these are able to predict, to a good level of approximation, the degrees of eight atmospheric pollutants measured using automatic control stations (SO₂, NO_x, Cl, Cd, Pb, Zn, and dusts; Amman et al., 1987).

The formula with the highest correlation with pollution data is that which considers as a parameter only the frequency (*F*) of the lichen species present in a sampling network comprising 10 areas:

$$IAP = \sum F_i$$

F is the frequency (max. 10) of every *i*th species that is calculated as the number of rectangles in the grid (a rectangle of the dimensions 30×50 cm, subdivided into 10 areas measuring 15×10 cm each), in which a given species appears (Herzig and Urech, 1991). It has been

shown that the frequency method makes it possible to predict pollution levels with a certainty of over 97% (Lo Porto et al., 1992; Gottardini et al., 1999).

Method A foresees a choice of sampling stations on the basis of the presence of suitable trees on which it is possible to observe lichens. The difficulty of this method lies in finding the same tree species in the study sites so as to enable homogenous observations to be made. For example, in Italy trees of the *Tilia*, *Acer*, *Quercus* and other species are used. In the event that the species are not totally homogenous, observations can be made using other, different species of tree. When selecting suitable trees, it is necessary to take into account the state of damage to the bark as well as the inclination of the trunk (this must be < 10%) and the circumference (min. 70 cm).

Periodic notations are made of all lichen species present within the network (made on a weekly and monthly basis, etc.). A frequency value (F) is given for each species noted and this corresponds to the number of sub-units within the network in which it is present (minimum = 1, maximum = 10). The IAP is then computed for each tree and each station.

The values obtained may be plotted in order to create an air quality map. The IAP values are grouped into five quality levels which are given in Table 1 (Kommission Reinhaltung der Luft im VDI und DIN, 1995).

The main part of the studies that concern air quality in Italy deal fundamentally with atmospheric pollution in towns and cities or in larger geographical areas, where different sites with different impacts are compared. Of the numerous works, we quote as examples data collected in different Italian sites: the city of Isernia (Manuppella and Carlomagno, 1990); the province of Potenza (Lo Porto et al., 1992); the cities of Trieste (Nimis, 1985), Udine (Nimis, 1986), Pistoia (Loppi et al., 1992b), Siena (Monaci et al., 1997), Montecatini Terme (Loppi et al., 1997a), Trento (Gottardini et al., 1999), and La Spezia (Nimis et al., 1990; Palmieri et al., 1997); the Veneto region (Nimis et al., 1991); the Valle del Susa in Piedmont region (Piervittori, 1998); the city of Teramo (Loppi et al., 1998b), the volcanic areas of Italy (Grasso et al., 1999); the city of Pavia (Brusoni et al., 1997), and the province of Viterbo (Bartoli et al., 1997).

Cislaghi and Nimis (1997) report a high degree of correlation between lung cancer and the biodiversity

of lichens as a result of atmospheric pollution. These conclusions are based upon thousands of observations made in 662 sites in the Veneto region (northern Italy). These high correlation levels have been found for the more common atmospheric pollutants, such as SO_2 , NO_3 , dusts and SO_4^{2-} , which is, respectively: $r^2 = 0.93$, 0.87, 0.86 and 0.85: $P < 0.01$ in all cases. Of the many lichen species present (Nimis and Tretiach, 1995), *Physcia tenella* is among the most common in Italy, above all below the mountain areas. It is a species that is considered to be toxin-tolerant, even if experts are in disagreement as to its sensitivity to sulphur dioxide. In urban environments or cultivated areas, it is possible to find toxin-tolerant species that belong — from a phytosociological viewpoint — to the lichen association category of *Xanthorion parietinae* (Nimis, 1987). The more toxin-tolerant species include *Pheophyscia orbicularis* and *Candelaria concolor*; this latter being present at the limit of the “lichen desert”.

Different lichen species react to different pollutants in different ways, and various authors give lists that classify them according to sensitivity (Nimis and Tretiach, 1995). The classification of lichen species is one of the most discussed points in literature. In particular, sensitivity to SO_2 is the base factor for most classifications. Several authors, however, suggest classification on the basis of a scale of semiquantitative characteristics (Wirth, 1991); while others classify lichen species according to SO_2 sensitivity on a scale that distinguishes between “acid” and “eutrophic” bark (Hawksworth and Rose, 1970).

The latter method, basically qualitative, considers the degree of atmospheric pollution varying from 10 (zero pollution pure air) to 0 (strongest pollution) as a function of SO_2 levels. Each level is defined by various epiphytic lichens of broad ecological amplitude grouped in different communities according to the acidic or alkaline character of the bark. This method, due to its rapidity and sensitivity, can be applied to draw cartographic representations of pollution indexes on ample geographic areas, also based on absolute values, provided the lichen flora is comparable to that used for the original reference study in United Kingdom. A partial drawback of this method is that the knowledge of 80 lichens species is required (Deruelle, 1978).

Van Haluwyn and Lerond (1986) proposed a qualitative method based on lichenosociology. The authors suggest the use of a 7-level scale (indicated by letters A to G) defined on the basis of easily recognizable species. According to this method, letters A to E refer to strongly polluted, and letters F and G to less polluted, areas. Studies performed on Northern France showed that the two group of zones relates to SO_2 levels higher and lower than $30 \mu\text{g}/\text{m}^3$, respectively. According to this method, even the presence of one species only can be

Table 1
Quality levels of index of atmospheric purity (IAP)

| | | |
|---------|-------------------------------|------------------------------|
| Level A | $0 \leq \text{IAP} \leq 12.5$ | Very high level of pollution |
| Level B | $12.5 < \text{IAP} \leq 25$ | High level of pollution |
| Level C | $25 < \text{IAP} \leq 37.5$ | Moderate level of pollution |
| Level D | $37.5 < \text{IAP} \leq 50$ | Low level of pollution |
| Level E | $\text{IAP} > 50$ | Very low level of pollution |

sufficient to characterize one zone. One of the major advantages of this method is that it is not directly correlated to the levels of SO₂ only, but it is based on the overall response capacity of epiphytic communities as a whole (Lerond et al., 1996).

The IAP method is the most commonly adopted in Italy. Despite the quantitative information that it can supply, this method also presents some disadvantages, primarily among them the fact that a deep knowledge of lichen flora is required, and that it refers to a specific group of environmental pollutants (Amman et al., 1987).

Another quantitative method is the index of poleotolerance (IP) proposed by Trass (1973) and subsequently reviewed by Deruelle (1978). This method, developed and applied in Estonia, allows to draw a map of pollution on the basis of a mathematical index that, in turns, is obtained following observation carried out in predefined conditions. This method considers trees of different age and species. Every observation is performed taking into account the area of the bark covered by lichens, which is related to a graded reference scale varying from 1 to 10 according to the percent of covered surface. Indeed, every species is classified according to the IP, calculated as:

$$IP = \sum_{i=1}^n a_i \times c_i / C_i$$

where n represents the number of considered species, a_i the degree of tolerance of each species, c_i the corresponding level of covering and C_i the overall degree of covering of all species as a whole.

According to this method, an IP value of 10 refers to a zone of lichen desert, while normal condition correspond to level 1–4. IP can also be correlated with SO₂ levels, where an IP of 1–2 corresponds to zero SO₂ and an IP of 10 to SO₂ concentration higher than 300 µg/m³.

Deruelle (1978), Van Haluwyn and Lerond (1988), Lerond et al. (1996) critically reviewed the main advantages and disadvantages of the above-mentioned qualitative and quantitative methods. The two qualitative methods are also compared by Khalil and Asta (1998), in a French study considering the recolonization by pollution sensitive lichens of the Lyon area.

Some authors (Insarov et al., 1999) propose a methodology for biomonitoring climatic changes by measuring the lichen communities of calcareous rocks and for determining the Trend Detection Index with which to verify the sum of lichen species, allowing variability coefficients to be applied to lichen communities that are sensitive to average annual temperature changes of up to 0.8°C. This application is of great interest, above all taking into account that realistic predictions for planetary global warming should be in the range of 2.5°C for the end of the 22nd century.

4.2. Use of native lichens and the transplant method (method B)

In areas where lichens are not killed by contaminants, it is possible to make biomonitoring studies through the direct analysis of contaminants in the thallus. Method B, which consists of transplanting lichen thalli, has the great advantage of being applicable even in “lichen desert” areas (in areas that are unsuited to lichen survival due to high pollution levels), or it can be used in areas where there are no suitable substrata.

The lichen thalli used are taken from tree bark in areas of low pollution and then fixed to suitable surfaces (e.g. cork) and exposed in monitoring areas where samples are taken periodically in order to evaluate the health of the thalli and their degree of damage. Lichen damage is expressed as a percentage of necrotised lichen surface.

The main problem with this method is that found in the difficulty in providing a valid interpretation of transplanted thalli damage percentages. There are also methods that allow identification of necrotic areas, defining them on photographs of lichen thalli. A certain error margin, due to the subjective interpretation of the images, has also been found for this procedure. Possible tendencies to over- or underestimate may be corrected through use of statistics tests (χ^2 , t student).

The transplant method is also used in classical bioaccumulation studies that analyse contaminants in tissue. Numerous works regarding this method are concerned with trace elements and in particular, bioaccumulation, absorption, retention, localisation and release, tolerance and toxicity (James, 1973; Nieboer et al., 1978; Burton et al., 1981; Brown and Beckett, 1984; Burton, 1986; Nash and Wirth, 1988; Puckett, 1988; Richardson, 1988, 1992, 1995; Nash, 1989; Brown, 1991; Deruelle, 1992; Garty, 1992, 1993; Sloof, 1995; Bargagli et al., 1997; Garty et al., 1997, 1998a; Benett and Wetmore, 1999; Freitas et al., 1999).

4.3. Heavy metals

The accumulation of metals in plants depends upon many factors, such as the availability of elements; the characteristics of the plants, such as species, age, state of health, type of reproduction, etc.; and other such parameters as temperature, available moisture, substratum characteristics, etc. (Baker, 1983). Contaminants deposit on lichens through normal and indirect (occult) precipitation. This latter includes mist, dew, dry sedimentation and gaseous absorption (Knops et al., 1991). Indirect precipitation occurs in highly stable atmospheric conditions and contains higher nutriment and contaminant concentrations of different orders of size when compared to normal precipitation (Nash and Gries, 1995).

In general, three mechanisms have been put forward with regard to the absorption of metals in lichens (Richardson, 1995):

1. intracellular absorption through an exchange process;
2. intracellular accumulation; and
3. entrapment of particles that contain metals.

Many experts have attempted to increase knowledge of these bonding processes — that is, the interaction between lichen and metal — using various analytical techniques, such as nuclear magnetic resonance, electron paramagnetic resonance, and luminescence. It should, however, be noted that knowledge regarding the understanding of the entire process that is responsible for metal absorption and accumulation in lichens is still scarce. A new approach has recently been attempted (Antonelli et al., 1998), where metal-lichen interaction is studied by applying microcalorimetric techniques with the aim of obtaining enthalpic measurement data. To carry out these tests and to process the microcalorimetric data, the metal-lichen complex is considered as an overall co-ordinating agent, given that it is not possible at this time to know which particular molecule is responsible for co-ordination with the metal. Considering the constant towards equilibrium and the enthalpy trend for *Evernia prunastri*, the following trend has been found: $Pb \gg Zn > Cd \approx Cu \approx Cr$; which indicates a good correlation between the metal bond and the enthalpy values in the absorption process (metal uptake).

Lichens are also excellent bioaccumulators of trace elements, since the concentrations found in their thalli can be directly correlated with those in the environment (Andersen et al., 1978; Herzig et al., 1989; Sloof and Wolterbeek, 1991; Herzig, 1993; Bari et al., 1998).

Studies made of transplanted *Evernia prunastri* highlight the fact that the capacity for Pb accumulation expressed as the relationship between the concentration in the latest sample and the initial concentration value, is 10.2 in the Fontainebleau site (France), 3.7 for the Würzburg site (Germany; Deruelle, 1992), and 4.4 for the city of Rome (Italy; Bartoli et al., 1994).

In Italy, different biomonitoring studies carried out using lichens have shown that Pb is still very widespread in spite of the introduction of lead-free petrol. This indicates that high levels of this metal are still released (and/or re-suspended) by vehicle traffic (Cardarelli et al., 1993; Deruelle, 1996; Monaci et al., 1997). Vehicle traffic seems to be the main source of atmospheric Cr, Cu and Pb in the central Italian sites (Loppi et al., 1998b).

Climatic factors most probably play an important role in the bioaccumulation of heavy metals, even if this role is as yet unclear. The direction in which pollutants are transported by the wind is most surely fundamental in determining their main fallout points. Nimis et al. (1989) correlates pollution from an industrial pole

(northern Italy) with that at a distant agricultural centre, situated in the predominant wind direction.

It is well known that heavy metal content in lichen thalli tends to alternate over time in phases of accumulation and subsequent release. The causes of these differences may lie in the incidence on this phenomenon of acid rain. Deruelle (1992) indicates that the periodic releases of Pb that occur in lichens may depend upon lixiviation induced by acid precipitation. Indeed, laboratory experiments show that lixiviation does not occur at pH 7 (Nimis et al., 1989). Heavy metals do, in any case, influence water loss in lichen thalli, and the accumulative effect of Pb, Cu and Zn on water loss, after absorption of a mix of metals in solution, has been observed in the laboratory (Chettri and Sawidis, 1997).

Altitude seems to play an important role in Pb and Cd concentrations, as studied on *Hypogymnia physodes* (Kral et al., 1989). In particular, Pb concentration increases in a linear fashion as altitude increases, while Cd increases in the same way up to altitudes of 900–1100 m. For higher altitudes, Cd concentrations follow a decreasing trend. What is more, *Hypogymnia physodes* is one of the most suitable bioindicators in the study of the bioaccumulation of trace elements (Jeran et al., 1996) in view of its high-tolerance capacities.

In general, the higher accumulation of heavy metals in the thallus found after the summer period, may be due to the increased hydration that results from autumn rainfall (Nieboer et al., 1978). In Mediterranean climates, the trace element content in lichens as they are (unwashed), is strongly influenced by soil dust contamination (Loppi et al., 1997a). Loppi et al. (1999) in spite of high correlation levels of Al, Fe and Ti in *Parmelia sulcata* does not find any linear correlation for these elements with their concentration levels in the soil. This would lead to the supposition that contamination through dust is highly variable and probably depends upon the local characteristics of the sites under study.

Cd is considered to be particularly toxic for various lichen species (Nieboer et al., 1979; Beckett and Brown, 1984). Concentration intervals of 1.26–5.05 and 1.56–6.40 $\mu\text{g g}^{-1}$ have been found for *A. ciliaris* and *L. pulmonaria* respectively. These values (considering average values) are considered to be close to the appearance of toxicity symptoms. Furthermore, Cd has a high negative correlation with protein and reducing sugar content (Riga-Karandinos and Karandinos, 1998).

Lichens from the *Usnea* species have been used to evaluate heavy metal deposition patterns in the Antarctic (Poblet et al., 1997). The activities carried out in the different scientific stations could be potential sources of pollution and contribute to the circulation of trace metals in this site.

The relationship between cationic concentrations in lichens, as shown for *Cladonia portentosa*, can be used as an index of acid precipitation. In particular, the

K^+/Mg^{++} ratio and the (extracellular) Mg^{++} /(intracellular) Mg^{++} in lichen apexes is strongly correlated to H^+ concentrations in precipitation. High concentrations of H^+ that are found in acid rain cause increases in extracellular Mg^{++} . In general, the variation in Mg^{++} concentration in lichens can be considered to be a good indicator of acid rainfall (Hyvarinen and Crittenden, 1996).

Acid-moisture depositions containing heavy metals can significantly reduce lichen survival in affected geographical areas. In lichens (*Bryoria fuscescens*) exposed to simulated acid rainfall containing two levels of Cu^{++} and Ni^{++} only or combined with acid rain (H_2SO_4) at pH 3 for 2 months in addition to environmental rainfall, it was observed that the alga and fungus components respond in different ways to pH levels and that they have a specific interaction that is correlated to the toxicity of the metals. In particular, the alga component is the more sensitive to acid rain and to the mix of heavy metals and, as a result, it has a higher quantity of degenerated cells, which causes significant changes in membrane permeability. Critical concentrations of heavy metals in alga thalli were $> 50 \mu g g^{-1}$ for Cu and $> 7 \mu g g^{-1}$ for Ni in the presence of acidity and $> 20 \mu g g^{-1}$ for Ni in absence of acidity (Tarhanen, 1998; Tarhanen et al., 1999).

Another recently developed field of application for biomonitoring with lichens is that of indoor pollution and in particular, the analysis of air particulates. Rossbach et al. (1999) found a high ratio between the concentrations of Cr, Zn, and Fe in air particulate samples taken from the filters of air conditioning systems in different hotels in different cities and in *Usnea* spp. samples found in the conditioned environments.

Table 2 reports some bibliographical data on heavy metal bioaccumulation on lichens. For over 20 years, lichens have been used as bioindicators and/or biomonitors in environment quality evaluations for such industrial realities as iron foundries and fertiliser manufacturing plants (Kauppi, 1976; Laaksovirta and Olkkonen, 1977; Palomaki et al., 1992); steel works and iron foundries (Pilegaard, 1978, 1979; Pilegaard et al., 1979); oil extraction plants (Addison and Puckett, 1980); sites contaminated by the petrochemical industry (Pakarinen et al., 1983); areas surrounding zinc foundries (de Bruin and Hackenitz, 1986); areas surrounding nickel foundries (Nieboer et al., 1972); coal-fired power stations (Olmez et al., 1985; Garty, 1987; Freitas, 1994); power stations in high-density industrial areas (Gonzalez and Pignata, 1997). Garty et al. (1997), in a study of a heavy oil combustion plant in the south-west of Israel (Ashdod area), carried out using transplanted of fruticose epiphytic lichens (*Ramalina duriaei*) found high concentrations of S, V and Ni in thalli that were transplanted to the industrial area. These values can be correlated with environmental measurements of SO_2 and V

made in the same region. Furthermore, the high V/Ni ratio in the lichens may be an indicator (tracer) of pollution in the area caused by heavy oil combustion plants. The same authors (Garty et al., 1998a) found a high Pb, V, Ni, Zn and Cu bioaccumulation potential for *Ramalina lacera* in the same site.

Lichen are also used to study copper dust emissions from mines. *Ramalina fastigiata* has been used as a bioindicator of the impact of a coal mine in Portugal. The threshold concentration of intracellular Cu above which total inhibition of photochemical apparatus occurs, is approximately $2.0 \mu mol g^{-1}$ (Branquinho et al., 1999). *Neophuscelia pulla* and *Xanthoparmelia tartarctica* were used to study the bioaccumulation of heavy metals in abandoned copper mines in Greece, where a significant correlation ($P < 0.05$) was found between the copper content in the soil and that of the lichen thalli (Chettri et al., 1997). *Hypogymnia physodes* has been used as a bioindicator of the presence of mercury and methylmercury in metal extraction areas in a site in Slovenia, where its excellent bioaccumulation capacities were confirmed (Lupsina et al., 1992). Riga-Karandinos and Karandinos (1998) took three native lichen species (*Anaptychia ciliaris*, *Lobaria pulmonaria*, and *Ramalina farinacea*) directly from 22 sites distributed over an area of 250 km² in Peloponnesus, Greece, where there is a coal-fired power station. There they found levels of Cd that were close to levels of toxicity for lichens (respectively, 3.09, 3.42, and 3.80 $\mu g g^{-1}$) and below-toxic levels of Pb (respectively, 8.60, 9.76, and 11.18 $\mu g g^{-1}$).

Lichens are excellent bioindicators of atmospheric pollution from geothermal power stations and in particular, of the pollutants that are correlated with this phenomenon, such as mercury (Bargagli and Barghigiani, 1991), boron (Koranda, 1980), radon (Matthews, 1981) and other metals (Connor, 1979). Several authors (Loppi, 1996; Loppi et al., 1998a) give data on atmospheric pollution from geothermal power stations in central Italy using the method of mapping lichen communities, where they found minimum IAP values within 500 m of the power station and progressive increases in frequency in line with increases in distance from the power stations themselves. Pollutants that are typically associated with geothermal activity are As, B, Hg, and H_2S (Loppi, 1996; Loppi and Bargagli, 1996). Nonetheless, it is not clear if the drop in the richness of the species in those areas close to geothermal power stations is due to the action of a single contaminant or to the combined actions of all contaminants. In any case, the author considers that the worst damage to lichen thalli is caused by H_2S , which is a highly toxic gas (Beauchamp et al., 1984) and which is continuously present in high levels as a local contaminant in areas surrounding geothermal power stations.

Different studies have established correlation factors between chlorophyll damage and the concentrations of

Table 2

Selected references of heavy metals ($\mu\text{g/g}$ dry weight) studied on lichens species from different geographical areas (mean values and ranges of concentrations^a)

| Species | Site | Note | Cd | Cr | Cu | Fe | Mn | Ni | Pb | V | Zn | References |
|--------------------------------|--|---|---------------------|-----------------------------------|-----------------------------------|---------------------|-----------------------|-----------------------------------|------------------------------------|--------------------|-----------------------------------|--------------------------------------|
| <i>Anaptychia ciliaris</i> (N) | Southern Greece | Area of 250 km ² of Peloponnesus where a lignite-burning power plant is located (RM) | 3.09 (1.26–5.05) | | 4.06 (1.10–5.60) | 2153 (1359–3092) | 43.87 (15.6–92.1) | | 8.60 (3.57–12.6) | | 31.22 (23.3–41.9) | Riga-Karandinos and Karandinos, 1998 |
| <i>Lobaria pulmonaria</i> (N) | Southern Greece | idem c.s. | 3.42 (1.56–6.4) | | 6.85 (4.6–12.3) | 1103 (339–2180) | 65.32 (17.7–137.1) | | 9.76 (3.87–21.1) | | 28.16 (16.9–59.4) | Riga-Karandinos and Karandinos, 1998 |
| <i>Ramalina farinacea</i> (N) | Southern Greece | idem c.s. | 3.80 (2.18–7.06) | | 3.63 (1.70–5.80) | 748 (409–1222) | 52.35 (28.7–81.4) | | 11.18 (5.10–19.5) | | 19.46 (15.8–25.6) | Riga-Karandinos and Karandinos, 1998 |
| <i>Ramalina duriaei</i> (N) | Israel | SC | | | 5.5 | | | | | | | Kardish et al., 1987 |
| <i>Ramalina duriaei</i> (T) | Israel | AC | | | 12 | | | | | | | Kardish et al., 1987 |
| <i>Ramalina duriaei</i> (T) | Israel | Urban (U), rural (R) and sub-urban (SU) sites that are 4.5–24.5 km from a power plant | | (U) 11.7 (R) 12.8 (SU) 23.8 | (U) 13.6 (R) 12.9 (SU) 14.2 | | | (U) 49.5 (R) 14.5 (SU) 20.0 | (U) 165.4 (R) 30.3 (SU) 43.4 | | (U) 59.6 (R) 49.4 (SU) 59.0 | Garty, 1988 |
| <i>Ramalina duriaei</i> (N) | HaZorea (Northeast Israel) | SC | | 10.8 | 10 | | | 20.1 | 22.6 | | 33.1 | Garty, 1988 |
| <i>Ramalina duriaei</i> (N) | Israel | Urban- indus. area | | | | | | 14.3 | | 23.1 | | Garty et al., 1997 |
| <i>Ramalina lacera</i> (T) | Israel | Suburban area | | | | | | 15.3 | | 31.7 | | |
| <i>Ramalina lacera</i> (T) | Israel | Rural area | | | | | | 9.9 | | 6.4 | | |
| <i>Ramalina lacera</i> (T) | Israel | Southwest (AC) | | 6.53 (5.4–8.5) | 7.34 (5.2–12.4) | 1505 (1235–1954) | 32.51 (26.4–40.9) | 6.84 (2.1–12.2) | 34.95 (10.4–155) | 14.1 (6.7–26.2) | 60.1 (38–113) | Garty et al., 1998 |
| <i>Evernia prunastri</i> (T) | Rome | Area of 300 km ² of the big annular connection (AC) | | 14.05 | 13.37 | | | | 40 | 5.2 | 57.45 | Bartoli et al., 1994 |
| <i>Parmelia sulcata</i> (T) | Bern (Switzerland) | Urban area | | 28.5 | 47.5 | 833.5 | 18 | 106 | 189 | | 259 | Garty and Amman, 1987 |
| <i>Parmelia sulcata</i> (T) | Biel, Champagne Allee (Switzerland) | Urban area | | 22 | 84 | 890 | 21 | 22 | 172 | | 191 | Garty and Amman, 1987 |
| <i>Parmelia sulcata</i> (T) | Lauenen, (near Gstaad, Switzerland) | Suburban area | | 526 | 18 | 4613 | 48 | 667 | 29 | | 192 | Garty and Amman, 1987 |
| <i>Parmelia sulcata</i> (N) | Tuscany (provinces of Siena and Grosseto) (RM) | SC | | 3.6 | 9.1 | 1800 | 38.2 | | 15.9 | 2.32 | 65.7 | Loppi et al., 1999 |
| <i>Parmelia sulcata</i> (N) | Portugal | 228 sites along the Atlantic coast and interior of the country | | 5.77 (1.53–32.3) | | | | 3.92 (0.52–33.1) | 18.4 (2.0–142) | 14.5 (1.83–130) | | Freitas et al., 1999 |
| <i>Hypogymnia physodes</i> (N) | Village of Gusum (Sweden) | Surroundings of a brass foundry | 1.1 (0.4–1.7) | | 28.2 (11–79) | 832 (290–1300) | | 2.6 (1.7–3.9) | 22.6 (14–33) | | 232 (93–450) | Folkesson, 1979 |
| <i>Hypogymnia physodes</i> (T) | Bern (Switzerland) | Urban area | | n.d. | 23 | 864 | n.d. | 30 | 315 | | 224 | Garty and Amman, 1987 |
| <i>Hypogymnia physodes</i> (T) | Biel (Switzerland) | Urban area | | 30 | 41 | 1311 | 13 | 12 | 111 | | 159 | Garty and Ammann, 1987 |

Table 2 (continued)

| Species | Site | Note | Cd | Cr | Cu | Fe | Mn | Ni | Pb | V | Zn | References |
|-------------------------------------|---|---|----------------------|---------------------|--------------------|--------------------|--------------------|---------------------|-------------------|------|--------------------|---------------------------|
| <i>Hypogymnia physodes</i> (T) | Slovenia | The sites were located at least 300 m away from main roads and at least 100–200 m away from dwelling (RM) | 1.05 (0.31–5.42) | 5.78 (2.33–21.8) | | 1253 (492–3756) | | | | | 90.2 (47.3–151) | Jeran et al., 1996 |
| <i>Parmelia caperata</i> (T) | Washington (near to the Potomac River) | The locations are near the Dickerson power plant | 1.2 (1.1–1.4) | 3.8 (2.0–5.1) | | 1400 (750–2090) | 240 (140–380) | | | | 64 (55–80) | Olmez et al., 1985 |
| <i>Parmelia caperata</i> (T) | Travale-Radicondoli (central Italy) | Area of 15 km ² near a geothermal power plant (RM) | 0.329 (0.11–0.69) | 4.51 (1.25–8.41) | 10.8 (4.5–25.4) | 1019 (275–2370) | 85.8 (10.9–280) | 4.41 (1.65–8.18) | 6.3 (2.1–19.7) | | 43 (22.2–63.8) | Loppi and Bargagli, 1996 |
| <i>Parmelia rudecta</i> (T) | Washington (near to the Potomac River) | The locations are near the Dickerson power plant | | 4.6 (2.8–7.3) | | 1620 (780–3090) | 230 (86–365) | | | | 68 (34–100) | Olmez et al., 1985 |
| <i>Cetraria cucullata</i> (N) | Northwest of Canada | | | | | 478.8 | 47.7 | 2.5 | 4.2 | 1.78 | 24.1 | Puckett and Finegan, 1980 |
| <i>Cetraria nivalis</i> (N) | Northwest of Canada | | | | | 257.7 | 84.5 | 2.7 | 5.6 | 1.2 | 25.0 | Puckett and Finegan, 1980 |
| <i>Cladonia stellaris</i> (N) | Northwest of Canada | | | | | 568.7 | 30.2 | 2.9 | 4.3 | 3.98 | 15.9 | Puckett and Finegan, 1980 |
| <i>Cetraria islandica</i> (N) | Switzerland (Devos) | Rural area | | 8 | 8 | 149 | 43 | 7 | 40 | | 60 | Garty and Amman, 1987 |
| <i>Usnea filipendula</i> | Slovenia | Surroundings of a brass foundry | 0.6 (0.4–1.0) | | | 22.4 (7–48) | 614 (250–1400) | 2.6 (1.7–4.1) | 27.0 (15–40) | | 182 (82–200) | Folkesson, 1979 |
| <i>Usnea sp.</i> (T) | Switzerland (Devos) | Rural area | | 18 | 14 | 402.5 | 26 | 32 | 120 | | 72.5 | Garty and Amman, 1987 |
| <i>Pseudovernia Furfuracea</i> (N) | Slovenia | Surroundings of a brass foundry | 0.6 (0.4–0.9) | | 35 (15–71) | 926 (504–1640) | | 2.8 (1.9–3.9) | 37.3 (20–53) | | 237 (125–444) | Folkesson, 1979 |
| <i>Pseudovernia Furfuracea</i> (T) | Switzerland | | | 41.8 | 34.7 | 3423 | 44.8 | 74.5 | 135.5 | | 159.5 | Garty and Amman, 1987 |
| <i>Cladonia rangiferina</i> (N) | Slovenia | Surroundings of a brass foundry | 0.5 (0.1–1.0) | | 14.5 (4–40) | 442 (160–1200) | | 1.5 (0.8–2.3) | 22.8 (11–36) | | 102 (51–204) | Folkesson, 1979 |
| <i>Umbilicaria deusta</i> (N) | Sudbury District (Northern Ontario, Canada) | Surroundings of a copper foundry | | | 65 | 1470 | | 37 | | | | Tomassini et al., 1976 |
| <i>Umbilicaria muhlenbergii</i> (N) | Sudbury District (Northern Ontario, Canada) | Surroundings of a copper foundry | | | 30 | 920 | | 16 | | | | Tomassini et al., 1976 |
| <i>Stereocaulon paschale</i> (N) | Sudbury District (Northern Ontario, Canada) | Surroundings of a copper foundry | | | 43 | | | 26 | | | | Tomassini et al., 1976 |
| <i>Cladonia alpestris</i> (N) | Mackenzie Valley (Ontario, Canada) | Boreal forest | | | 12.6 | 164 | | | | | | Tomassini et al., 1976 |
| <i>Cladonia alpestris</i> (N) | Sudbury District (Northern Ontario, Canada) | Surroundings of a copper foundry | | | 15 | 320 | | 11 | | | | Tomassini et al., 1976 |

(Table continued on next page)

Table 2 (continued)

| Species | Site | Note | Cd | Cr | Cu | Fe | Mn | Ni | Pb | V | Zn | References |
|------------------------------------|---|---|----|------|------|-----|------|------|------|---|------|------------------------|
| <i>Cladonia deformis</i> (N) | Mackenzie Valley (Ontario, Canada) | Boreal forest | | | 1 | 70 | | | | | | Tomassini et al., 1976 |
| <i>Cladonia deformis</i> (N) | Sudbury District (Northern Ontario, Canada) | Surroundings of a copper foundry | | | 21 | 220 | | 10 | | | | Tomassini et al., 1976 |
| <i>Cladonia mitis</i> (N) | Mackenzie Valley (Ontario, Canada) | Boreal forest | | | 20.6 | 170 | | | | | | Tomassini et al., 1976 |
| <i>Cladonia mitis</i> (N) | Sudbury District (Northern Ontario, Canada) | Surroundings of a copper foundry | | | 19 | 260 | | 10 | | | | Tomassini et al., 1976 |
| <i>Cladonia uncialis</i> (N) | Sudbury District (Northern Ontario, Canada) | Surroundings of a copper foundry | | | 19 | 210 | | 10 | | | | Tomassini et al., 1976 |
| <i>Cladonia uncialis</i> (N) | Mackenzie Valley (Ontario, Canada) | Boreal forest | | | 4 | 120 | | | | | | Tomassini et al., 1976 |
| <i>Diploschistes steppicus</i> (N) | Sede Boquer region (Negev desert, Israel) | In this area no industry exists, we may assume that the heavy metals derive partly from car traffic | | 17.2 | 66.9 | | 66.0 | 23.6 | 63.0 | | 33.2 | Garty, 1985 |
| <i>Teloschistes lacunosus</i> (N) | Sede Boquer region (Negev desert, Israel) | idem c.s. | | 15.2 | 4.7 | | 33.5 | 12.8 | 19.4 | | 30.3 | Garty, 1985 |
| <i>Squamarina crassa</i> (N) | Sede Boquer region (Negev desert, Israel) | idem c.s. | | 18.0 | 14.4 | | 68.6 | 21.2 | 31.5 | | 41.9 | Garty, 1985 |
| <i>Ramalina maciformis</i> (N) | Sede Boquer region (Negev desert, Israel) | idem c.s. | | 10.3 | 8.1 | | 8.1 | 11.9 | 39.6 | | 20.9 | Garty, 1985 |
| <i>Caloplaca ehrenbergii</i> (N) | Sede Boquer region (Negev desert, Israel) | idem c.s. | | 20.0 | 8.7 | | 42.4 | 14.4 | 35.3 | | 28.8 | Garty, 1985 |

^a T, transplanted; N, natives; AC, contaminated area; SC, control site; n.d., no detectable; RM, reference material used.

several elements in lichens. Garty et al. (1998a) found that chlorophyll integrity is inversely correlated with concentrations of Cr, Fe, Mn, Ni, Pb, and B. K concentrations, however, have a positive correlation with chlorophyll integrity (Kauppi, 1976; Garty et al., 1998a). In a study of *Cladina stellaris* samples that were transplanted to an area near to a fertiliser factory in Finland, Kauppi (1976) found that the high concentrations of K in the lichens corresponded to increases in chlorophyll content and in particular, to the chlorophyll *a*/chlorophyll *b* ratio.

Garty et al. (1998a), found differences of approximately factor 2 in electrical conductivity in lichens from industrially polluted sites (Israel) compared with those from rural sites. This indicates a process of cell membrane damage. Membrane integrity is highly correlated with the presence of Ca (Beckett and Brown, 1984), which is a macronutrient and has regulatory functions in sites of extracellular interchange on the surface level of alga cell walls or hyphae, and intracellular interchanges with proteins. Concentrations of several elements (S, B, Al, Cr, Fe, Si, Ti, and Zn) are positively correlated with cell membrane damage for *R. duriaei* (Garty et al., 1998b).

Various studies report that exposure of lichen thalli to chemicals in laboratory and using fumigation experiments, reveals subsequent damage to cell membranes corresponding to an increase in water electrical conductivity caused by the loss of electrolytes (Pearson and Henriksson, 1981; Hart et al., 1988; Garty et al., 1998b). In general, the loss of K is related to cell membrane damage and is inversely correlated with electrical conductivity (Garty et al., 1998b; Tarhanen et al., 1999). The extent of cell membrane integrity may be evaluated by measuring electrical conductivity.

Another field of study using lichens as biomonitors is that involving volcanic areas (Grasso et al., 1999) and in particular, the release of mercury. In the Hawaiian islands, Hg has been found in the interval of 8–59 $\mu\text{g g}^{-1}$, meaning that bioaccumulation is therefore more widely distributed in areas that are strongly affected by volcanic activity (Davies and Notcutt, 1996).

An important problem — as far as concerns the determination of trace elements — is the quality control of analytic methods as well as of the sampling strategies and treatment of samples (Wolterbeek and Bode, 1995). In the last few years, some 2000 articles on the analysis of lichens have been published. The high variability of the data found may not only be caused by the different distribution of pollution patterns, but it may also be the result of possible errors in analysis. It is in this sense that the European Commission, through its Standards, Measurement and Testing programme has developed a new certified reference material (CRM) of lichen (CRM 482) to determine and remove the major sources of error in lichen analysis (Quevauviller et al., 1996). In

addition to this, the International Atomic Energy Agency (IAEA) has recently initiated an interlaboratory study with developing countries on the reference material IAEA 336 lichen (Smoldis and Parr, 1999).

It should be noted that the problem of sample pre-treatment and collection methods, as well as the standardisation of analytical methods, is of fundamental importance above all when comparing lichens from different geographical areas (Puckett, 1988; Jackson et al., 1993). Literature contains little information above all about the techniques for washing lichens. Different lichen washing strategies may cause relevant changes in metal or sulphur content (Wadleigh and Blake, 1999) or in fluorides (Palomaki et al., 1992), as against unwashed samples.

Furthermore, the enormous developments in analytical techniques (atomic absorption spectroscopy, ICP-AES, neutron activation, etc.) should be taken into consideration. These developments have, in the last few decades, notably improved instrument detection limits, allowing increasingly accurate analyses and eliminating possible sources of error. When determining trace elements in lichens, some interesting analytical applications, such as the use of short-life radionuclides in neutron activation (Grass et al., 1994) and X-ray fluorescence spectrometry (Caniglia et al., 1994; Richardson et al., 1995), can supply valid information, in particular for the determination of macronutrients (K and C), or of trace metals (Cu, Pb, Zn) and non-metals (S).

4.4. Sulphur compounds

The effect of sulphur compounds on lichens have been extensively studied. For indeed, many studies are generally concerned with the effects of SO_2 fumigation of exposed lichens (Rao and LeBlanc, 1966; Henriksson and Pearson, 1981; Fields, 1988; Balaguer and Manrique, 1991; Gries et al., 1995), or with the effects of simulated acid rainfall (Scott and Hutchinson, 1987; Holopainen and Kauppi, 1989; Sanz et al., 1992; Tarhanen, 1998). Other works deal in the respiration rate (Baddeley et al., 1972), photosynthesis (Richardson and Puckett, 1973; Kauppi, 1976; Lechowicz, 1982) and chlorophyll fluorescence (Calatayud et al., 1996, 1999; Deltoro et al., 1999). For the most part, these studies aim to evaluate the effects of sulphur compounds on the physiology of lichen thalli and/or on the integrity of photobiont chlorophyll.

Chlorophyll analysis is usually carried out following the method proposed by Ronen and Galun (1984). Phaeophytin is a product of chlorophyll degradation. The variation in the normal chlorophyll–phaeophytin ratio is an indication of suffering in lichens. This method foresees the extraction of chlorophyll using 5 ml of solvent (DMSO). The ratio between chlorophyll *a* and phaeophytin *a* is measured using a spectrophotometer

(OD 435 nm/OD 415 nm); and this is considered to be an appropriate index for measuring the impact of high concentrations of SO₂ in lichens, or for evaluating the effects of heavy metal pollution in transplanted lichens (Garty, 1987). A ratio of 1.4 indicates that chlorophyll is unchanged. Any reduction in this value indicates chlorophyll degradation with ensuing stress to the organism (Boonpragob and Nash, 1991; Gonzales and Pignata, 1994, 1997; Levin and Pignata, 1995; Gonzales et al., 1996, 1998; Silberstein et al., 1996). Kardish et al. (1987) report a value of 1.44 for the Chl/Ph ratio of *Ramalina duriaei* in the control site, while for a polluted site with high levels of vehicle traffic, they found a value of 0.80. In general, an alteration of the Chl/phaeophytin ratio has been found, indicating the toxic effect of a combination of gaseous and non-gaseous pollutants.

Sulphur content is determined by transforming elementary sulphur into SO₄²⁻ ions, which occurs through the acid suspension method using barium chloride (Gonzales and Pignata, 1994; expressed in mg g⁻¹ dry weight).

Some authors (Levin and Pignata, 1995; Gonzales et al., 1996; Carreras et al., 1998) report that data relating to sulphur accumulation and obtained indirectly from the bioindicator, seem to show that the influence of SO₂ from industry (Córdoba, Argentina), is rather restricted compared to that which comes from vehicle traffic.

Sensitivity to SO₂ and to other atmospheric pollutants in general, varies according to species (Insarova et al., 1993). *Lobaria pulmonaria* is considered to be one of the most sensitive species according to the scale of Hawksworth and Rose (1976): 30 µg m⁻³ for average winter concentrations of SO₂. This species' high degree of sensitivity is probably due to the presence of isidia, which is a vegetative structure on the upper surface of the thallus and which plays a role in asexual reproduction. The isidia increases the absorption surface of the thallus per unit of mass (Riga-Karandinos and Karandinos, 1998). *Hypogymnia physodes* is, on the contrary, a species that is particularly resistant to SO₂. Indeed, it has been seen that exposure of this species to H₂SO₄ in highly acid conditions, produces no effect (Garty et al., 1995). *H. physodes* has also been used in the area surrounding a fertiliser plant in Finland, where sulphur levels of 3000 ppm had been found (Palomaki et al., 1992). Typical ultrastructural damage caused by the action of sulphur on photobiont cells is seen within the first two weeks of the transplant, without the sulphur concentration levels being particularly high.

Another interesting field of research is that which correlates sulphur content and the composition of sulphur isotopes (Case and Krouse, 1980; Krouse and Case, 1981; Takala et al., 1991). A recent study (Wadleigh and Blake, 1999) reports the spatial variation of the sulphur isotope composition of 83 epiphytic

lichen samples (*Alectoria sarmentosa*). The study reveals a positive correlation between isotope composition and different sources of sulphur emission in the site under study (Newfoundland Island, Canada). It is interesting to note the fact that lichens are also sensitive to the sulphur salts that come from the sea. Indeed the degree of sulphur concentration has been seen to decrease in lichens, the further they are from the coastal to internal areas of the island.

The role of sugars in alga–fungus interaction is most important in lichen biology. Chronic SO₂ fumigation of lichens may cause interference in the flow of such nutrients as carbohydrates, thus creating symbiont damage. SO₂ causes reducing sugars to increase and non-reducing sugars to decrease. This effect is probably due to a breakdown in the polysaccharides that are rich in reducing sugars. Reducing sugars are determined by extracting 10 mg of lichen thallus with 1 ml of d-H₂O and centrifugation at 2000×g for 10 min in an Eppendorf vials. A mix of two solutions (4 ml; sodium potassium tartrate and an indicator) are added to the supernatant and the vials were bathed in darkness at 100°C for 3 min. After the proper cooling period, solution absorbance is measured at 660 nm (Riga-Karandinos and Karandinos, 1998).

Spectroscopic measurements, carried out to study changes in the spectral reflectance response of lichen thalli have been exposed in contaminated sites as against those exposed in control sites (Satterwhite et al., 1985; Garty et al., 1997). As a rule, lichen scanning revealed, as for the higher plants of uncontaminated sites, a significant drop of between 600 and 700 nm (which corresponds to the absorption region of the chlorophyll), and a net increase in spectral reflectance around 700 nm (red edge) together with a continual and relatively high reflectance in the near infrared between 700 and 1100 nm. The near infrared plateau is the result of differences between the varying refractive indexes of the internal components of the thallus (cell walls, chloroplast, air, water content, etc.). In spectra of lichens transplanted to contaminated sites, the red edge (700 nm) is much less pronounced and the plateau is very low. This indicates a clear situation of organism stress (Garty et al., 1997). In extreme cases, for example in plants that are subjected to high stress levels, or which are already dead, the spectrum shows a continuous line that rises gradually.

Membrane proteins may be damaged by the presence of SO₂, which may cause a reduction in protein biosynthesis in some lichens; or there may be negative effects on the nutritional interchange between symbionts with, as a consequence, an alteration of their delicately balance. To determine protein content, 100 mg of tissue are extracted using 3 ml of a phosphorous buffer solution at pH 7 and the extract is centrifuged at 1600×g for 5 min. The extracted solution (100 µl) is added to 5 ml

of Bradford solution (Bradford, 1976) and absorbance at 595 nm is taken and compared with a calibration curve made following with protein standard (e.g. bovine serum albumin-BSA; Riga-Karandinos and Karandinos, 1998). The structural proteins found in cell membranes and the lichens enzymes can have considerable damage in the presence of high levels of SO₂ concentration. These processes concern the delicate interchange of nutrients between symbionts and can damage the delicate equilibrium of the association (Fields and St. Clair, 1984).

Thus, the damage to cell membranes can be used as an indicator of environmental stress. Indeed, it has been demonstrated that SO₂, such as O₃ and NO₂, are powerful catalysts of lipid membrane peroxidation (Menzel, 1976; Gonzalez and Pignata, 1994; Gonzalez et al., 1996). Experiments where lichens were exposed to 1 ppm of SO₂ in aqueous solution show a slight reduction in the overall content of phospholipids and an increase in unsaturated fatty acids. This latter type of response to SO₂ may be considered to be of the adaptive type (Bychek-Guschina et al., 1999).

The effect of SO₂ can also be evaluated by dry weight/fresh weight ratio. This ratio has been proposed as an indication of the influence of the environment on the bioindicator. It has indeed been observed that in highly polluted areas (e.g. where traffic is intense), there is a tendency in lichens to lose moisture (Levin and Pignata, 1995).

Finally, the production of ethylene is another indicator of stress in lichens. Lichens exposed to solutions containing sulphur in an acid environment have different levels of ethylene production. In general, these solutions increase the solubility of the particles containing heavy metals that are trapped within the hyphae. This phenomenon may lead to an increase in the production of endogenous ethylene in lichens when they are exposed to chemical agents containing sulphur, to acid rain and to air polluted with heavy metals (Garty et al., 1995).

4.5. Nitrogen and phosphorous compounds

Although lichens have already been proposed as bioindicators of NH₃ (De Bakker, 1989), only in the last few years has a clear positive correlation been established between nitrophytic lichens and atmospheric NH₃ concentrations; even if responses are always greater to SO₂. Tree bark analyses in sites in Holland demonstrate that nitrophytic lichen species do not respond directly to nitrogen levels found in the environment, but that they are favoured by the high pH values in the bark, which are related to the high levels of NH₃ in the environment (Van Dobben and Ter Braak, 1998).

Cladonia portentosa is an excellent bioindicator for the study of precipitation chemicals and nitrogen and

phosphorous concentrations. Hyvarinen and Crittenden (1998a, b) have found concentrations in the range of 0.08–1.82% for nitrogen and 0.04–0.17% for phosphorous (per unit of dry weight) in apexes (5 mm top part) and thalli (bases) in various comparison sites. The concentration levels found for these elements are 2–5 times higher in the apexes than at the bases and furthermore, both the apexes and base parts show a high positive correlation between elements. The correlation between N deposition and the nitrogen accumulated in the lichens is positive; becoming higher when referred to concentrations found in the thalli rather than in the apexes. On the other hand, nitrogen concentrations in the thallus are little correlated with the N values in precipitation. The nitrogen found in thalli is, however, highly linked to moist nitrogen deposits, but it is also correlated positively with the NO₂ present in the air. As well as *C. portentosa*, *H. physodes* has also been proposed as a bioindicator of nitrogen total deposition (dry and wet; Sochting, 1995) as well as of nitrogen and sulphur (Bruteig, 1994).

High levels of SO₂ and NO_x can cause a reduction of pH of lichen thalli (see for example data respective to Peloponnesus area [Greece], Riga-Karandinos and Karandinos, 1998). To this respect, it shall be highlighted that atmospheric pollution of this kind has led to extinction of *L. pulmonaria* and *R. farinacea*. The measurement of the pH of lichen thalli can supply information with regard to the state of pollution of a site. To determine pH levels, 50 mg of lichen thallus is homogenised in liquid nitrogen and 2 ml of d-H₂O. After centrifugation at 100×g for 10 min, the pH value for the supernatant is read. Various authors report that *L. pulmonaria* is endangered in some sites subject to acid rain, and pH = 5 has been indicated as a threshold value below which lichen is unable to survive (Gauslaa, 1985; Gilbert, 1986).

4.6. Ozone

O₃ and NO₂ (see also Section 4.4) are powerful catalysts of lipid membrane peroxidation: the main effect of O₃ on lichens is indeed the damage of cell membrane.

It has been demonstrated that in biological systems the presence of oxidation products such as malondialdehyde is directly correlated to the start of the peroxidation of unsaturated fatty acids (Mehelman and Borek, 1987). Egger et al. (1994), reports an increased production of both malondialdehyde and superoxide dismutase in *Hypogymnia physodes* that was transplanted to highly polluted sites with monthly O₃ concentrations in the range of 20–198 µg m⁻³ (10–100 ppb). These compounds are products of lipid peroxidation and are indicators of oxidative damage to membranes and to the enzyme systems that protect against oxidation in plants.

Oxidation products are estimated by determining malondialdehyde (MDA), which is measured using the colorimetric method (Heath and Packer, 1968). MDA is determined using the extinction coefficient of $155 \text{ mM}^{-1} \text{ cm}^{-1}$ (Kosugi et al., 1989). The results are expressed in $\mu\text{mol g}^{-1}$ of dry weight.

Other important peroxidation products are hydroperoxy conjugated dienes (HPCD) which can be isolated by solvent extraction. Concentration is calculated using the molar extinction coefficient $\epsilon = 2.64 \times 10^4 \text{ M}^{-1} \text{ m}^{-1}$. Results are expressed in mmol g^{-1} of dry weight.

O_3 damage to the photochemical apparatus of lichens after repeated exposure to real doses has been well-documented (Scheidegger and Schroeter, 1995). In particular, this kind of damage has been studied in other species such as, for example, clover, where pollution causes typical and easily recognisable leaf damage (Karlsson et al., 1995). Furthermore, O_3 is the subject of monitoring and controls made by the European Community (Benton et al., 1995). Ross and Nash (1983), report a study where *Flavoparmelia caperata* was fumigated with O_3 for brief periods (10 and 12 h) and in quantities of $200 \mu\text{g m}^{-3}$ (100 ppb). This caused a 50% net decrease in photosynthesis. *Usnea ceratina* was fumigated for 6 h per day for a period of 5 days with concentrations of 100–200 ppb of O_3 , causing a notable reduction in net photosynthesis (Zambrano and Nash, 2000).

4.7. Fluorides, chlorides and other atmospheric pollutants

Literature regarding the bioaccumulation of fluorides is scarce. Asta and Garrec (1980) have demonstrated that fluoride concentration in the lichen thallus is dependent on both the lichen species and the environmental F levels. Fluoride levels above 360 ppm were found in lichens (*H. physodes* and *Bryoria capillaris*) transplanted to areas near to a fertiliser factory and mine (Palomaki et al., 1992). Fluoride accumulation in these sites reached maximum levels during the summer followed by decreases in the autumn. Already at levels of 30–40 ppm of dry weight it is possible to see the typical ultrastructural damage in the photobiont cells, caused by exposure to fluorides. It has also been found that fluoride content in lichens is inversely correlated to the distance from an aluminium processing plant. The losses in the lichen populations around this plant show a high level of correlation ($r^2 = 0.90$) with their F content (Perkins, 1992).

Studies inherent to the impact of chlorides are also rather scarce. To this regard, a biomonitoring study using the lichen *Parmelia sulcata* Tayl. and several mosses, carried out in 26 sites, shows a spatial and temporal correlation between chloride bioaccumulation and the environmental impact trend of a waste incin-

eration plant in the city of Grenoble, France (Gombert and Asta, 1997).

For other atmospheric contaminants, such as polychlorinated dibenzodioxins and polychlorinated dibenzofurans (PCDD_s/PCDF_s), bioindication studies using lichens are very scarce. In general, the available studies of PCDDs and PCDFs report the temporal variations of these pollutants in samples of different types (e.g. plants) taken in the vicinity of urban waste incineration plants (Schuhmacher et al., 1997; Schuhmacher et al., 1998). Furthermore, for example, the eggs of several bird species and in particular, those of the herring-gull, seem to be good bioindicators of the presence of PCDD_s/PCDF_s (Oxynos et al., 1997).

4.8. Radionuclides

Lichens are good bioaccumulators of radionuclides (Notter, 1988). This application concerns an important sector of research into the evaluation of the fallout of radionuclides, above all after the Chernobyl incident (Barci et al., 1988; Seaward et al., 1988, Mihok et al., 1989; Livens et al., 1991; Sloof and Wolterbeek, 1992; Hofmann et al., 1993; Triulzi et al., 1996; Sawidis et al., 1997). In lichens in several areas of Norway, after the Chernobyl incident of 1986, levels of Cs 134 and Cs 137 of two orders of size larger than those in vascular plants were found (Bretten et al., 1992). This phenomenon has caused a significant increase in the average concentration of radiocesium in reindeers, whose major food source are lichens (Jones et al., 1989; Rissanen and Rahola, 1989). *Parmelia sulcata* has been used as a bioindicator for the presence of radionuclides in areas close to Chernobyl where I 129 and Cl 36 have been measured. Regional distribution patterns of these radionuclides have shown a positive correlation with accumulated concentrations (Chant et al., 1996). One study reports values of the natural decontamination of radionuclides in lichens (Topcuoglu et al., 1995). The average biological life-span of Cs 137 is 58.6 months in *Xanthoria parietina*, which has shown itself to be the best bioindicator of radioactive fallout as against mosses.

It has been demonstrated that plutonium concentrations in *Xanthoria* spp. gathered in the vicinity of a nuclear arms deposit are inversely correlated with the distance from the contamination site, and that there is a direct correlation between concentrations of Pu 239 and Pu 240 with concentrations found on the soil surface ($r^2 = 0.767$; $P < 0.001$; Thomas and Ibrahim, 1995). Altitude is an important factor that is correlated with concentrations of Ra 226 and Ra 228 studied in lichens of the *Umbilicaria* species (Kwapulinski et al., 1985). Altitude is also correlated with levels of Cs 137 found in 1993 in samples taken in Italy in the province of Parma (Triulzi et al., 1996).

5. Concluding comments

The analysis of atmospheric pollutants using conventional analytical procedures allows data to be interpreted directly and results to be obtained rapidly. A summary of the most common conventional analytical techniques for the analysis of various environmental pollutants is given in Table 3.

On an ecological level, however, air quality studies using these methods may present the following problems:

1. space-time fluctuations could lead to sampling errors;
2. low concentrations of several microcontaminants (which could also change over time), could lead to difficulties in methodology;
3. it is also difficult to ascertain either the intermittent or sporadic emission of contaminants;
4. in this way the biological tolerance limits of the species concerned might not be taken into consideration; and

Table 3
Analytical techniques for the analysis of environmental pollutants

| Pollutant | Reference instrumental methods |
|--------------------------------|--|
| SO ₂ | Flame photometric Gas chromatography with flame photometric detector Spectrophotometric (pararosaniline wet chemical) Electrochemical Conductivity Gas-phase spectrophotometric |
| O ₃ | Chemiluminescent Electrochemical Spectrophotometric (potassium iodide reaction, wet chemical) Gas-phase spectrophotometric |
| NO ₂ | Chemiluminescent Spectrophotometric (azo-dye reaction wet chemical) Electrochemical Gas-phase spectrophotometric Conductivity |
| Fluorides | Potentiometric method |
| PAH | Gas chromatography associated with high resolution mass spectrometry |
| PCDD | Gas chromatography associated with high resolution mass spectrometry |
| PCDF | Gas chromatography associated with high resolution mass spectrometry |
| Metals | Atomic absorption spectrometry Atomic emission spectrometry Inductive coupled plasma emission spectrometry Inductive coupled plasma optical emission spectrometry Inductive coupled plasma mass spectrometry |
| Chlorine and hydrochloric acid | Volumetric method; spectrophotometric analytical method |
| Phosphorus and its compounds | Gas chromatography with Nitrogen/Phosphorous detector; X ray spectroscopy |

5. often the dose-effect ratio does not have linear response and it is thus possible to run into interpretation problems in evaluating damage to organisms and ecosystems.

The above points highlight the fact that traditional environmental monitoring methods require numerous and extensive samples to be taken in the areas under study and that these samples must also be taken over prolonged periods of time.

Furthermore, the use of mathematical models of contaminant dispersion in the environment should also be pointed out (Benedini and Cicioni, 1992). These models, based on physical and chemical properties, have produced excellent results in the last few years, especially as far as predictions of contaminant dispersion and potential bioavailability are concerned. However, these methods are little developed, above all with regard to effects on species and ecosystems (Conti, 1996).

These models of pollutant propagation and transport usually, however, concern punctual sources of contamination and they also require large quantities of information if they are to be applied. From this stems the importance of biomonitoring when establishing contaminant levels in organisms and to use in ascertaining possible toxicity in relation to organism placement within the ecosystem.

From an ecotoxicological viewpoint, it is not possible, starting from chemical analysis, to establish a valid model for toxicity capable to foresee the bioavailability and various complex synergies that run between the organisms present in an ecosystem.

The ability to predict the incidence of many human activities with regard to a species and above all, to an ecosystem, is very limited. The difficulty in establishing a cause-and-effect relationship derives above all from the systematic lack of information on the “state of health” of the environment being studied or from the nature of its biological processes (which do not have linear characteristics and which are discontinuous through space and time). From here stems the importance of environmental monitoring plans and biomonitoring plans that, if properly applied, can supply an overall complete picture of the possible interventions that may be required.

Although lichens are important sources for control and environmental biomonitoring, it is necessary to take various precautions when using lichens as a quantitative measure of a single contaminant. Bioindicators of air pollution can most certainly supply information of a qualitative type; nonetheless, correlation studies using environmental data from the sites concerned and taken over a prolonged period of time (months–years), can supply information about semiquantitative aspects.

Through IAP calculation, lichens allow us to evaluate air quality as far as regards the presence of different

environmental contaminants. As already mentioned, lichens do not react specifically to a particular contaminant, but rather to the overall toxic effect of a mix of contaminants (Amman et al., 1987). Levin and Pignata (1995) proposed the use of Pollution Index (PI) for the evaluation of air quality. PI is determined using the equation cited by Levin and Pignata (1995). This enables the evaluation of which of the biomonitoring areas has the better air quality by measuring lichen reactions to atmospheric pollutants.

$$PI = [(Pha/Chla) + S_p/S_c](HPCD_p/HPCD_c)$$

The phaeophytin *a*/chlorophyll *a* ratio can be changed to chlorophyll *b*/chlorophyll *a*:

$$PI = [(Chlb/Chla) + S_p/S_c](HPCD_p/HPCD_c)$$

Chl *b* and Chl *a* express chlorophyll *a* and *b* concentrations in mg g⁻¹ of dry weight. *S* is the sulphur content of lichens expressed as mg g⁻¹ of dry weight, while HPCD expresses the concentration of hydroperoxy conjugated dienes in mmol g⁻¹ of dry weight. The sub-index *p* indicates concentrations measured in samples transplanted to contaminated sites, while sub-index *c* indicates those measured in lichens transplanted to the control site.

Biomonitoring studies using lichens make it possible to verify, with our current state of knowledge, air quality and any improvements thereof. This is what happened in the case of the progressive improvement in air quality over the years (1989–1994) in several Italian cities (La Spezia; Palmieri et al., 1997). This improvement was due to the reduction in SO₂ emissions, which was partly linked to the increase in the use of methane gas for domestic heating and to the closure of a coal-fired power station. The city of Montecatini Terme (central Italy) has also improved its environmental situation: new lichen species have been found and the previous “lichen desert” situation has disappeared (Loppi et al., 1997b). This phenomenon has been correlated with low SO₂ emission levels (approx 15–20 μg m⁻³ from 1993 to 1996) and NO_x, which passed from 150 μg m⁻³ in 1993 to 100 μg m⁻³ in 1996. A similar marked improvement caused by decrease in environmental SO₂ levels was also found in Paris in the 1980s (Luxembourg Gardens) where lichen species from the previous century began to reappear (Seaward and Letrouit-Galinou, 1991; Letrouit-Galinou et al., 1992).

In general, lichen distribution in northern Italy seems mainly to be regulated by SO₂ pollution (Nimis et al., 1990, 1991; Bargagli et al., 1991). As far as regards central Italy, a study of *Parmelia caperata* made by Loppi et al. (1992a) found a strong correlation ($r^2=0.93$; $P<0.05$) between IAP values and the total heavy metal content (Cd, Cr, Cu, Hg, Ni, Pb, Zn).

Techniques for drawing up air quality maps using lichens, or the use of the transplant method, allow us to obtain information about a vast area in a short amount of time and at contained costs.

These methodological approaches, although they cannot be considered as replacements for standard atmospheric pollution monitoring carried out using control stations, are without a doubt valid environmental biomonitoring instruments in different cases:

1. as a preliminary evaluation, or rather as an estimate of the base impact in a set area, with the aim of preventing future human-derived impact;
2. to monitor an already-compromised environmental situation; and
3. to control the quality of reclamation efforts already carried out.

Application of the system approach to the solving of problems regarding atmospheric pollution is doubtless valid and fundamentally requires an evaluation of the progress made in the areas of study considered, the identification of pollution sources, and the cause/effect correlation of the same (Conti, 1996). Of course, from that mentioned in the above points 1–3, it can be seen that for this reason, necessary interventions must have three main objectives:

1. environmental prevention: with the aim of intervening at the impact source and thus in advance of the pollutant event;
2. environmental protection: to eliminate the effects of pollutant actions or to tend to minimise these effects; and
3. environmental restoration: with the aim of removing damages caused by previous actions.

The necessity to increase our knowledge of bioindication studies using lichens remains a fundamental point in the development of research. It is possible to say that for a large majority of pollutants and their effects upon lichens, our knowledge is at an advanced stage in its development in terms of both the quantity and quality of information.

Nonetheless, it is possible to point out that in a significant part of bioindication studies of lichens, there is a tendency to study the environmental effects of situations that have already been compromised. This signifies a scarce propensity to carry out studies that fundamentally have an eye to aspects of environmental prevention.

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