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# Graphene-Alumina Nanocomposites with Improved Mechanical Properties for Biomedical Applications

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**Abstract:** The paper presents a study on graphene platelet (GPL)-reinforced alumina (Al<sub>2</sub>O<sub>3</sub>) ceramic composites and the relationships between the loading of GPL and both mechanical properties and in-vitro biocompatibility. Al<sub>2</sub>O<sub>3</sub> powders with different GPL contents were prepared and sintered using a gas protected pressure-less furnace. The examination of the results shows the density of the composites varying from 99.2% to 95.6% with the loading of GPL from 0.75 to 1.48 vol%. Raman studies show that moderate agglomerations of GPLs

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Yang Yang has contributed equally to the work as the first author.

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3 occur during the ball milling process and graphitic defects were produced during the high  
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5 temperature processing. Mechanical properties of the  $\text{Al}_2\text{O}_3$  matrix are significantly improved  
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7 by adding GPLs. A maximum increase of approximately 60% in flexural strength and 70% in  
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9 fracture toughness are achieved by introducing 0.75 vol% GPLs. In the biocompatibility tests  
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11 it was found that cells directly seeding on top of GPL/ $\text{Al}_2\text{O}_3$  samples showed better initial  
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13 attachment (3 hours after seeding) and viability (3 days after incubation) than the monolithic  
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15  $\text{Al}_2\text{O}_3$ , indicating that the GPL/ $\text{Al}_2\text{O}_3$  composites have comparable or more favorable  
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17 biocompatibility. The excellent mechanical and biomedical properties of the GPL/ $\text{Al}_2\text{O}_3$   
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19 composites may enable them to be applied to a wide range of engineering and biomedical  
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21 applications.  
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26 **Keywords:** Graphene platelets, Ceramic matrix composites; Mechanical properties; In-vitro  
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28 Biocompatibility; Pressure-less sintering.  
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### 33 34 **1. Introduction**

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36 Ceramics have been applied as biomaterials for many decades due to their superior  
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38 mechanical and biocompatible properties. One particular such application is ceramic heads of  
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40 hip joint prostheses<sup>1</sup>. In addition, ceramics have also been used clinically in the femoral  
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42 components of knee joint prostheses and ankle joint prostheses<sup>2</sup>. In other medical fields,  
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44 ceramics are used as artificial bone to repair cranial and orbital bone defects and as dental and  
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46 cochlear implants<sup>3-7</sup>. Despite the widespread usage, further applications of ceramics are  
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48 seriously limited by their intrinsic brittleness. In the last few decades, much effort has been  
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50 paid to improve the toughness of the ceramics using micro- and nano-structures as  
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52 reinforcements. These methods have the similar aim of providing additional intrinsic grain  
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54 boundaries or extrinsic interfaces that prohibit dislocation-movement and long paths of crack  
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3 propagation<sup>8-10</sup>.  
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6 Graphene has received enormous attention globally in recent years. Its excellent electrical  
7 and mechanical properties make it suitable for many applications<sup>11</sup>. Compared to monolayer  
8 graphene, graphene platelets (GPLs) are stacked graphene with thickness of up to  
9 approximately 100 nm<sup>12</sup>. It is reported that the Young's modulus of GPLs with thickness of  
10 2-8 nm is approximately 0.5 TPa<sup>13</sup>, higher than many ceramics. Analogous to ceramic  
11 composites reinforced with micro-fibres, incorporation of GPLs into ceramic matrices is  
12 expected to lead to considerable improvement in strength and toughness. However, the  
13 effectiveness varies in reported literature. It is suggested that three factors need to be  
14 carefully addressed in the fabrication process in order to utilize graphene's full potential to  
15 improve the mechanical properties of the ceramics<sup>12, 14-15</sup>. They are (i): homogeneous  
16 dispersion of GPLs in matrices to prevent GPLs agglomeration caused by Van der Waals  
17 forces; (ii) full densification of the composites; (3) retention of the graphitic structure during  
18 the high temperature process. Since the agglomeration of GPLs is the source of cracks and  
19 cause of the degradation of the mechanical properties, the preparation routes for producing  
20 GPL/ceramic composite powders are critical to ensure the excellent performance of the  
21 sintered composites. At present, a variety of the preparation methods such as conventional  
22 powder mixing, colloidal mixing and molecular mixing have been adopted to form a  
23 homogenous distribution of GPLs in the ceramic powder. Among them, powder mixing is  
24 considered the most efficient way and easy to scale up for volume production. In particular, it  
25 has been proved that when Dimethylformamide or N-methylformamide (NMP) is utilized as  
26 solvent, a homogenous dispersion of GPLs with reasonably small thickness in the powder  
27 mixtures can be obtained. Another vital factor is to best retain the GPLs during the  
28 densification of composites. This way, good bonding between the GPLs and matrices can be  
29 created and high stiffness of GPLs can be preserved, which facilitates load transfer at the  
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3 interface between the GPLs and matrix and provide good reinforcing and toughening  
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5 efficiencies.  
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9 Currently, many studies have been reported on GPL/polymer composites with enhanced  
10 mechanical and electrical properties while only few reports can be found to explore the  
11 GPL/ceramic composites. It is partly due to the fact that it is relatively more difficult to  
12 acquire sintered ceramic components than polymer one. Several researchers have made the  
13 attempts. Walker et al. employed aqueous colloidal processing methods to obtain  
14 homogeneous dispersions of GPLs and silicon nitride ( $\text{Si}_3\text{N}_4$ ) ceramic particles which were  
15 densified using spark plasma sintering (SPS)<sup>16</sup>. The measured fracture toughness of  $6.6 \text{ MPa m}^{1/2}$   
16 for the composite with 1.5wt% of GPLs was 136% higher than that of monolithic  $\text{Si}_3\text{N}_4$ .  
17 Dusza et al. prepared the GPL-reinforced  $\text{Si}_3\text{N}_4$  composite containing 1 wt% GPLs using hot  
18 isostatic pressing and reported a 44% increase in fracture toughness over pure  $\text{Si}_3\text{N}_4$ <sup>17</sup>. Liu et  
19 al. employed the SPS to fabricate the GPL/zirconia ( $\text{ZrO}_2$ )-toughened  $\text{Al}_2\text{O}_3$  composite with  
20 addition of 0.81 vol% GPLs and found an about 40% increase in fracture toughness<sup>12</sup>. More  
21 recently, Yadhukulak-rishnan et al. fabricated GPL-reinforced zirconium diboride composites  
22 at a high sintering temperature using SPS and investigated the effect of GPL content on  
23 mechanical properties of the composites<sup>18</sup> and found an about 110% increase in fracture  
24 toughness of the composite with addition of 6 vol% GPLs. Nieto et al. used SPS to  
25 consolidate GPL/tantalum carbide (TaC) ceramic composites and found the fracture  
26 toughness of the GPL/TaC composite containing 5 vol% GPLs was 99% higher than that of  
27 pure TaC ceramic<sup>19</sup>. In fabrication of GPL/ceramic composites, advanced consolidation  
28 techniques such as SPS, hot isostatic pressing and hot pressing are usually preferred since the  
29 sintering densification process can be realized in a short time and a vacuum or inert  
30 environment as well as high pressure can be maintained to protect the GPLs.  
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3 In the exploration of graphene composites for biomedical applications, a few studies indicate  
4 the cytotoxicity of graphene is dose-dependent and caused by oxidative stress generated  
5 incontact with graphene<sup>20-21</sup> while others showing enhanced adhesion and spread of the cells  
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22. The contradictory results might be associated with the types of cells and GPLs used in the experiments. Lahiri et al. prepared the GPL/ultrahigh molecular weight polyethylene (UHMWPE) composites for orthopedic implants and found that the viability of osteoblasts deteriorates up to 86% on the surface of the UHMWPE composite reinforced by 1.0 wt.% GPLs<sup>23</sup>. Zhang et.al fabricated GPL/hydroxyapatite (HA) composites and reported that an addition of 1 wt% GPLs led to significant improvement of osteoblast adhesion<sup>22</sup>.

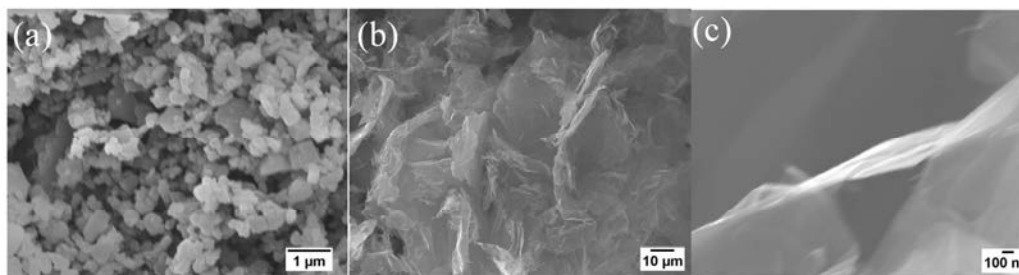
Very few reports can be found on pressure-less sintering of GPL/ceramic composites and the biocompatibility of GPL/ceramic composites and this study is set out to investigate them. Pressure-less sintering of GPL/Al<sub>2</sub>O<sub>3</sub> composites was carried out in a gas protected tube furnace and mechanical properties of the fabricated composites were studied. Raman spectroscopy was used to analyze the structural integrity of the GPLs. Phase composition and grain size of the ceramics were determined by the x-ray diffraction (XRD) and scanning electron microscopy (SEM) respectively. The effects of the GPL on the in-vitro biocompatible properties of the GPL/Al<sub>2</sub>O<sub>3</sub> composite materials were investigated using cell adhesion and 3-(4,5)-dimethylthiaziazolo (-z-y1)-3,5-di- phenytetrazoliumromide (MTT) tests. The role of the GPLs and their contributions towards the mechanical response and in-vitro biocompatibility of the GPL/Al<sub>2</sub>O<sub>3</sub> composite were analysed and discussed.

## 2. Experimental procedure

### 2.1 Starting materials

$\alpha$ -Al<sub>2</sub>O<sub>3</sub> powder (99.85%, 150 nm, Inframat Advanced Materials, Farmington, CT, USA) was used in this study. SEM images of Al<sub>2</sub>O<sub>3</sub> are presented in Figure 1a. GPLs were procured

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3 from Graphene Industries Ltd, Manchester, UK. SEM images of the obtained GPLs are  
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5 shown in Figure 1b and 1c. A general approach to produce the GPLs is via rapid thermal  
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7 expansion/exfoliation of graphite that has been intercalated using sulphuric acid. The resultant  
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9 platelets are stacks of graphene sheets about 6-8 nm in thickness and 15-25  $\mu\text{m}$  in diameter.  
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Figure.1 SEM images of Al<sub>2</sub>O<sub>3</sub> (a) powders and GPLs (b-c).

### 2.2 Powder mixture and green compact preparations

GPLs were first dispersed in NMP and sonicated for 40 minutes. Appropriate quantities of Al<sub>2</sub>O<sub>3</sub> nanoparticles were added and then the mixture was further sonicated for 10 minutes. This was followed by a ball milling procedure at 100 rpm in a planetary ball mill (PM 100, Retsch, UK) for 2 hours. The milling was carried out in a cylindrical ZrO<sub>2</sub> container using ZrO<sub>2</sub> balls under a ball-to-powder weight ratio of 2. The milled slurry mixture was dried at 90 °C in an oven for 3 days. The dried powder mixture was ground and sieved using a 140 mesh. Figure S1 shows the SEM images of the powder mixtures. A soft mould was then filled with the dried powders and subsequently underwent cold isostatic pressing to form green compacts.

### 2.3 Pressure-less sintering

Green compacts with nominal contents of 0, 1.17, 1.85 and 2.75 vol% GPLs were sintered using a tube furnace. Sintering temperature of 1650°C, soaking time of 2.5-3h and forming gas with flowing rate of 8 L/min were used to densify the powder mixtures.

#### 2.4 Mechanical measurements and microstructure characterization

The sintered samples were ground and polished to 0.5 $\mu\text{m}$  using silicon carbide papers and diamond suspension. The bulk density of the samples was measured using the Archimedes method with ethanol as the immersion medium. The densities of  $\text{Al}_2\text{O}_3$  and GPL adopted were 3.97 and 2.1  $\text{g m}^{-3}$ . The relative density was calculated by dividing the bulk density with the theoretical density of the powder mixture. Vicker hardness tests were carried out under a 1kg force. Flexural strength and fracture toughness of the sintered samples were obtained using four-point bending tests on an Instron 6025. In the flexural strength tests, specimens of 1.5  $\times$  2  $\times$  13 mm were machined. The span length and crosshead speed for the strength tests were 10 mm and 0.05  $\text{mm min}^{-1}$ . To avoid stress concentration, all the edges and corners of the specimens were chamfered using SiC grinding papers. The Single-edge V-notch beam method was used to determine the fracture toughness of the sintered samples at room temperature. Specimens of 2  $\times$  3  $\times$  15 mm were machined for toughness measurement. Notches at the centre of the test specimens were cut by a diamond wheel and further sharpened using a razor blade with the aid of diamond paste up to 1  $\mu\text{m}$ . The ratio between the notch depth and specimen thickness was approximately 0.25. Inner and outer spans of 6 mm and 10 mm, as well as a crosshead speed of 0.05  $\text{mm/min}$ , were applied in the toughness tests. Five samples were tested for each material. SEM was used to examine fracture surfaces of the ceramics and the polished samples thermally etched. Grain sizes were measured with the aid of the software (UTSHCSA, USA). At least 200 readings were taken to measure the grain sizes of each material. A Raman Microscope (Renishaw InVia Reflex and Witec Alpha 300R) was used to characterise the GPLs in the powder mixtures and sintered samples with the 532 nm laser wavelength excitation. Thermo gravimetric analysis (TGA, NETZSCH (STA 449C)) was carried out to examine the GPL contents within the sintered samples. The sintered samples were crushed, milled and sieved using a 140 mesh to produce powder



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3 mixtures for the TGA. In each test, a powder mixture of 100 mg was placed in an Al<sub>2</sub>O<sub>3</sub>  
4 crucible and examined in the temperature range between 200 and 800 °C in air. 800 °C was  
5 used as the upper limit in the TGA tests with the consideration that oxidation of GPLs should  
6 be completed at this temperature according to references<sup>24-26</sup>.  
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### 9 10 11 12 13 *2.5 In-vitro biocompatibility tests*

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15 Samples were machined to the disk shape with diameter of 18 mm and thickness of 2 mm for  
16 biocompatibility evaluation. All GPL/Al<sub>2</sub>O<sub>3</sub> composites and control group (the pure Al<sub>2</sub>O<sub>3</sub>  
17 ceramic) were supersonically cleaned in deionized water and acetone for 15 minutes  
18 separately, to remove any possible surface chemical groups or contaminations during  
19 fabrication and mechanical test procedures. Then all samples were sterilized by exposing to  
20 ultraviolet light overnight before the biological tests.  
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30 All biological reagents were obtained from Sigma Aldrich (Poole UK) unless otherwise  
31 stated. MTT assays were conducted to examine a pre-osteoblastic cell line (MC3T3) viability  
32 on the sintered samples. MC3T3 cells were thawed from liquid nitrogen stock (at passage  
33 number 10), cultured in dulbecco minimum essential medium (DMEM) (Sigma D6546,  
34 supplemented with 10% Fetal Bovine Serum, 2% L-glutamine, 2.4% 2-[4-(2-Hydroxyethyl)-  
35 1-piperazinyl] ethanesulfonic acid (HEPES) buffer and 1% penicillin-streptomycin) and  
36 incubated at 37°C in a humidified incubator with 5% CO<sub>2</sub>. After reaching 80% confluence,  
37 cells were trypsinised, centrifuged and re-suspended. Then cells were seeded onto prepared  
38 samples at the density of 50000 cells/cm<sup>2</sup>, cultured in an incubator for designed time periods  
39 and then washed gently with phosphate buffered saline (PBS) to get rid of floating cells  
40 before MTT assay.  
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55 10% w/v of MTT solution (5 mg/ml of MTT powder in PBS) was added into each assay.

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57 After incubation for 4 hours at 37°C, 1 ml of solvent (0.1 mol/L hydrochloric acid in 2-  
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3 propanol) was added into each assay and mixed for 2 minutes to dissolve formazan crystals  
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5 formed. Formazan solution was pipetted into a 96 well plate. Absorbance data was acquired  
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7 using a microplate spectrophotometer (Glowmax Multi, Promega Southampton, UK) at 560  
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9 nm reference wavelength.  
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12 Another set of cell adhesion tests were carried out using bone marrow stem cells (BMSCs).

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14 Primary cells were extracted from rat tibia and fibias and then cultured in alpha MEM  
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16 (supplemented with 10% Fetal Bovine Serum, 2% L-glutamine, 1% penicillin-streptomycin  
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18 and 2.4% HEPES Buffer) and incubated at 37°C in a humidified incubator with 5% CO<sub>2</sub>.  
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21 BMSCs were detached from cell culture flask and were then seeded onto the prepared  
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23 samples. Meanwhile, tissue culture plastic and Polydimethylsiloxane (PDMS) substrates were  
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25 used as positive and negative references respectively. Cells were cultured in an incubator at  
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27 37 °C with supplemented Alpha MEM. After 3 days of cell culture, cells were stained with  
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29 1ug/ml Calcein and 1ug/ml Propidium Iodide and inspected in florescence microscopes.  
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### 32 33 **3. Results and discussion**

#### 34 35 36 *3.1 Effects of content of GPLs on the microstructures and mechanical properties of* 37 38 *GPL/Al<sub>2</sub>O<sub>3</sub> composites.*

##### 39 40 *3.1.1 The content of GPLs within the sintered composites.*

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42 The results of TGA tests are shown in Figure S2. It is found that the sintered GPL/Al<sub>2</sub>O<sub>3</sub>  
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44 composites contain 0.4, 0.69 and 0.79 wt% GPLs respectively and the corresponding volume  
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46 percentages of GPLs in the sintered composites are 0.75, 1.3 and 1.48 vol%, which are lower  
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48 than the nominal content of GPLs within the composites and indicates part of the GPLs is  
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50 consumed during the high temperature sintering process.  
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### 3.1.2 Microstructures of the 'as-prepared' samples

The fracture surfaces of the sintered samples are shown in Figure 2. It can be seen that the pure  $\text{Al}_2\text{O}_3$  matrix exhibits an intergranular fracture mode while GPL-reinforced  $\text{Al}_2\text{O}_3$  composites present both an intergranular and intragranular fracture modes. The fracture mode is determined by the strengths of the ceramic matrix and the boundaries between them. For pure  $\text{Al}_2\text{O}_3$ , the strength of grain boundaries are weaker than the  $\text{Al}_2\text{O}_3$  grains, which enables cracks to propagate along the grain boundaries; while for GPL-reinforced  $\text{Al}_2\text{O}_3$  composites, the observed fracture mode implies the strengthened grain boundaries and moderate improvement in fracture toughness<sup>27</sup>. Meanwhile, it is noted that GPLs are well distributed in

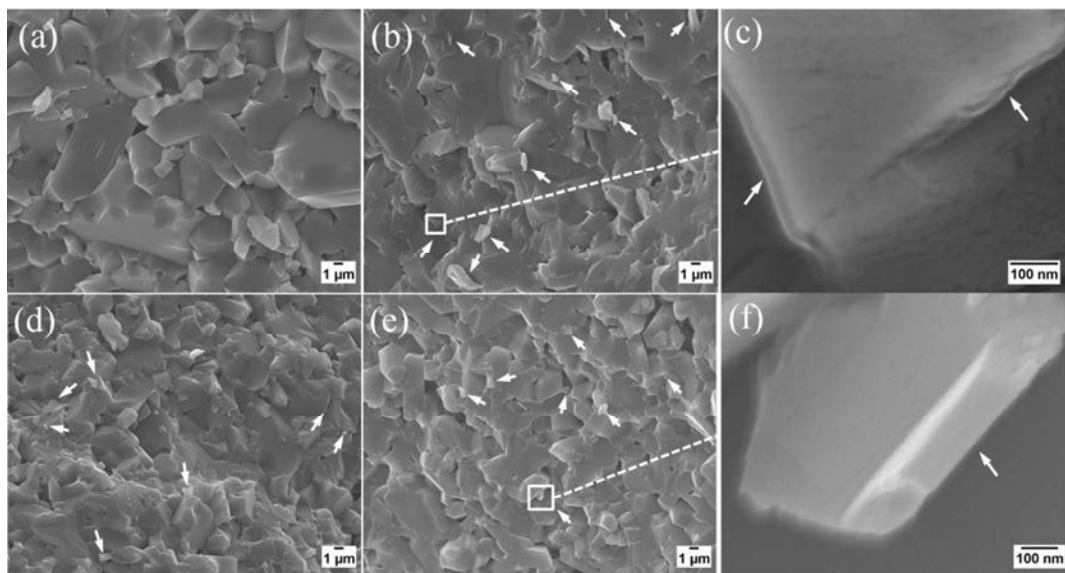
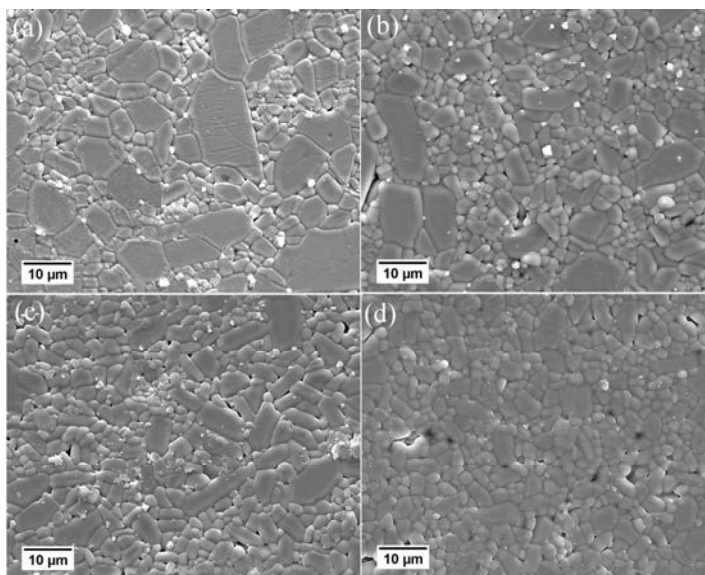


Figure 2. Fracture surfaces of sintered GPL/ $\text{Al}_2\text{O}_3$  composites. (a)  $\text{Al}_2\text{O}_3$ , (b) 0.75 vol% GPL/ $\text{Al}_2\text{O}_3$ , (d) 1.3 vol% GPL/ $\text{Al}_2\text{O}_3$ , (e) 1.48 vol% GPL/ $\text{Al}_2\text{O}_3$ . (c) and (f) are magnified images of square area in (b) and (e) respectively. White arrows indicate GPLs.

the  $\text{Al}_2\text{O}_3$  matrices, indicating the good dispersion of GPLs during the ball milling process. In addition, grain sizes of the  $\text{Al}_2\text{O}_3$  matrices decrease with the increasing percentage of GPLs,

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3 which is attributed to the pinning effect of GPLs in the grain boundaries. To obtain the  
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5 average grain sizes of the  $\text{Al}_2\text{O}_3$  matrices, polished and thermally etched surfaces of the  
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7 sintered samples are characterized using SEM. As shown in Figure 3, significant grain  
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Figure 3. SEM images of polished and thermally etched surfaces. (a)  $\text{Al}_2\text{O}_3$ , (b) 0.75 vol% GPL/ $\text{Al}_2\text{O}_3$ , (c) 1.3 vol% GPL/ $\text{Al}_2\text{O}_3$ , (d) 1.48 vol% GPL/ $\text{Al}_2\text{O}_3$ .

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decrease is observed by adding the GPLs. With the increasing percentage of GPLs the  $\text{Al}_2\text{O}_3$  matrices exhibit a more uniform and finer microstructure. The grain sizes of the  $\text{Al}_2\text{O}_3$  matrices are plotted in Figure S3. It can be seen that with the increasing concentration of GPLs from 0 to 1.48 vol%, the grain sizes decrease from 4.31 to 2.87  $\mu\text{m}$ . The significant refinement in the matrix microstructure is expected to contribute to the increase in both hardness and flexural strength. The microstructures obtained in present study using 150 nm raw  $\text{Al}_2\text{O}_3$  particles were compared with those in previous reports with raw  $\text{Al}_2\text{O}_3$  particle sizes of 140-150 nm<sup>28,29</sup>. In Santanach's work, the monolithic  $\text{Al}_2\text{O}_3$  with grain size of 7.5  $\mu\text{m}$  was produced using SPS at 1500 °C<sup>28</sup> while in this work the pure  $\text{Al}_2\text{O}_3$  prepared at 1650 °C using pressure-less sintering has much smaller grain size. This result implies that the

rate of mass transfer and grain growth in monolithic  $\text{Al}_2\text{O}_3$  is slower in pressure-less sintering than that in the sintering process with the help of pressure. However, the grain size ( $2.87 \mu\text{m}$ ) obtained for the 1.4 vol% GPL/ $\text{Al}_2\text{O}_3$  composite in present study is larger than that (below  $1 \mu\text{m}$ ) for the 0.59 vol% GPL/ $\text{Al}_2\text{O}_3$  composite produced at a lower temperature ( $1500 \text{ }^\circ\text{C}$ ) using SPS<sup>29</sup>. This fact indicates  $\text{Al}_2\text{O}_3$  grain growth is more sensitive to the sintering temperature than the GPLs content and the rate of grain growth in GPL/ $\text{Al}_2\text{O}_3$  composites can be higher in pressure-less sintering than that in pressure sintering due to a relatively higher sintering temperature.

### 3.1.3 Raman spectra of the pristine GPL and GPLs after ball milling and sintering.

Raman spectra of the pristine GPL and GPLs after the ball milling process are presented in Figure S4. It is very evident that D peaks of GPLs in powder mixtures show pronounced lower intensities than those of the pristine GPL.

To gain further insight into the structures of GPLs Raman parameters are compiled in Table 1. It is noted that GPLs in powder mixtures exhibit much lower ratio of intensities of D to G peaks ( $I_D/I_G$ ) compared to the pristine GPLs, indicating decreased defects of the GPLs<sup>18</sup>. The result can be explained by agglomeration of GPLs during ball milling, resulting in decreased

Table 1. Raman parameters of the pristine GPL and the GPLs in the powder mixtures

Materials	$I_D/I_G$	FWHM (G)	$\nu$ (G)	FWHM (2D)	$\nu$ (2D)	$I_{2D}/I_G$
Pristine GPL	0.26	20	1582	75	2714	0.65
1.17 vol% GPL/ $\text{Al}_2\text{O}_3$	0.04	17	1577	75	2712	0.63
1.85 vol% GPL/ $\text{Al}_2\text{O}_3$	0.06	18	1576	78	2704	0.63
2.75 vol% GPL/ $\text{Al}_2\text{O}_3$	0.04	18	1571	85	2695	0.52

number of edge defects. On the other hand, the  $I_{2D}/I_G$  of the GPLs in powder mixtures is lower than that of the pristine GPL which again indicates the occurrence of an agglomeration of GPLs. In addition,  $I_{2D}/I_G$  decreases with the increasing percentage of GPLs, implying addition of a high percentage of GPLs is more prone to causing formation of GPL agglomerates<sup>22</sup>. This argument can be further supported by the increasing full width at half maximum (FWHM) of 2D bands with an increasing percentage of GPLs.

The Raman spectra of the pristine GPL and GPLs in the sintered samples are compared in Figure 4. It can be seen that GPLs in the sintered samples exhibit much higher spectrum backgrounds compared to the pristine GPL. The increased spectrum backgrounds can be

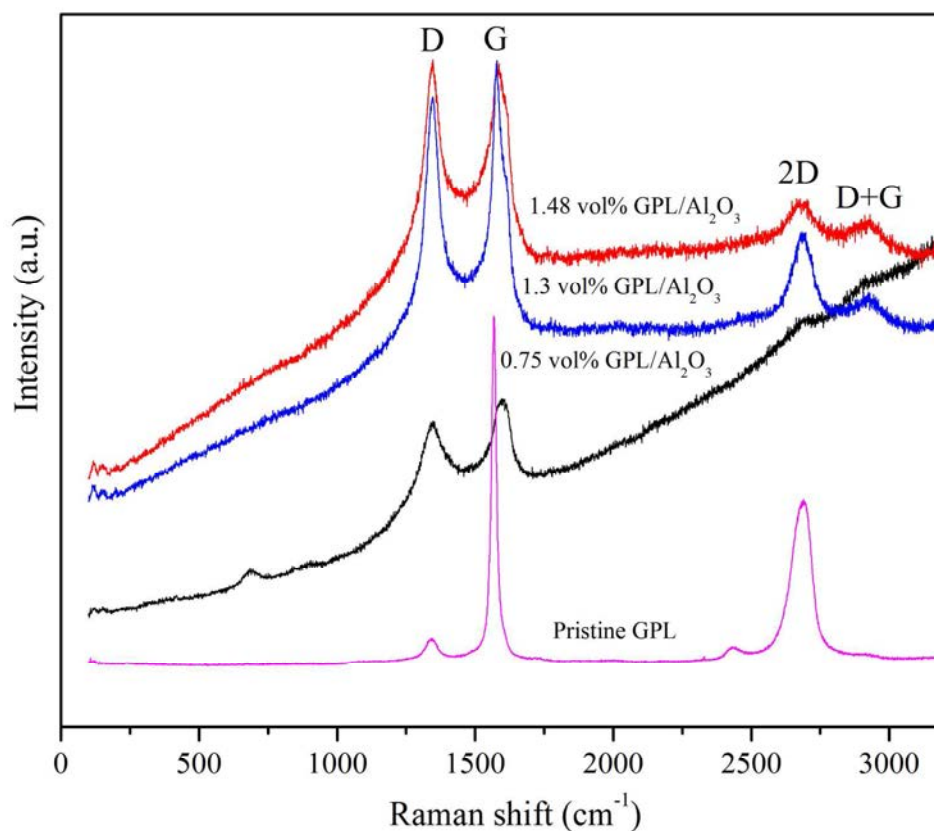


Figure 4. Raman spectra of the pristine GPL and GPLs after sintering.

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2  
3 attributed to interaction of GPLs with the  $\text{Al}_2\text{O}_3$  during the sintering process. Meanwhile, it is  
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5 evident that the after sintering GPLs show significantly higher graphitic defects ( $I_D/I_G$ ) and  
6  
7 the presence of D+G mode is observed in sintered samples. On the other hand, it is also noted  
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9 that GPLs in the sintered samples show far weaker signatures of G and 2D bands in  
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11 comparison with the pristine GPL, which implies that damage is induced during the sintering  
12  
13 process. Similar results were reported in Marc's work where reduced G and 2D peaks were  
14  
15 noted after damage was induced in GPLs by a pulsed laser<sup>30</sup>. It has been shown that  
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17 GPLs can readily absorb the oxygen molecules and hole doping in GPLs by oxygen  
18  
19 molecules can take place at a relatively low temperature (300 °C), which would alter the  
20  
21 integrity of the GPLs and cause the shift of the G band<sup>31-33</sup>. However, in our work a very  
22  
23 high sintering temperature (1650 °C) was used and it is expected that in addition to the hole  
24  
25 doping, the interaction between GPLs and oxygen will be significantly enhanced at such a  
26  
27 high temperature. As a result, GPLs might be able to react with the oxygen molecules to form  
28  
29 gas of carbon dioxide, leading to the loss of the GPLs.

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32  
33 Thinning effect of GPLs due to the interaction between the GPLs and ceramics was suggested  
34  
35 in many publications where a SPS furnace was used to consolidate the GPL/ceramic  
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37 composites in several minutes and at a relatively low sintering temperature (below 1650 °C)  
38  
39<sup>15, 18, 34</sup>. Given the fact that in the present work a much longer sintering time (3 hours) and a  
40  
41 relatively higher sintering temperature (1650 °C) were used in sintering process in a flowing  
42  
43 inert gas, the interaction between the GPLs and  $\text{Al}_2\text{O}_3$  would be considerably intensified,  
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45 causing the consumption of the GPLs. This argument can be indirectly confirmed by the high  
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47 background of the Raman spectra for GPLs in sintered samples.

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51 The shift of G bands is observed in Figure 4 and are usually attributed to the thermal stress  
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53 acting on GPLs incurred during thermal contraction of  $\text{Al}_2\text{O}_3$ , increased density of defects in  
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55 the GPLs and the change in the number of graphene layers<sup>18, 22, 35</sup>. It is worth to note that an

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3 additional factor which can lead to the shifting of the G band is associated with the hole  
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5 doping, which results from the interaction between the GPLs and oxygen molecules present  
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7 within the green compacts and the tube<sup>33</sup>.  
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### 9 10 3.1.4 XRD patterns of the pristine GPL and the sintered samples

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12 The XRD analysis of the pristine GPL, pure Al<sub>2</sub>O<sub>3</sub> and GPL/Al<sub>2</sub>O<sub>3</sub> composites is presented in  
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14 Figure 5. It can be seen that the GPL exhibits the same XRD pattern as the natural graphite  
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16 and there is no sign of the presence of new phases in the composites. It is reported that  
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18 aluminum oxycarbides are likely to be formed during a high temperature sintering process<sup>36</sup>.  
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20 However, no such phases are observed in the XRD patterns. The reason for this result might  
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22 be related to the low addition of GPLs which makes it hard to trace the reaction products.  
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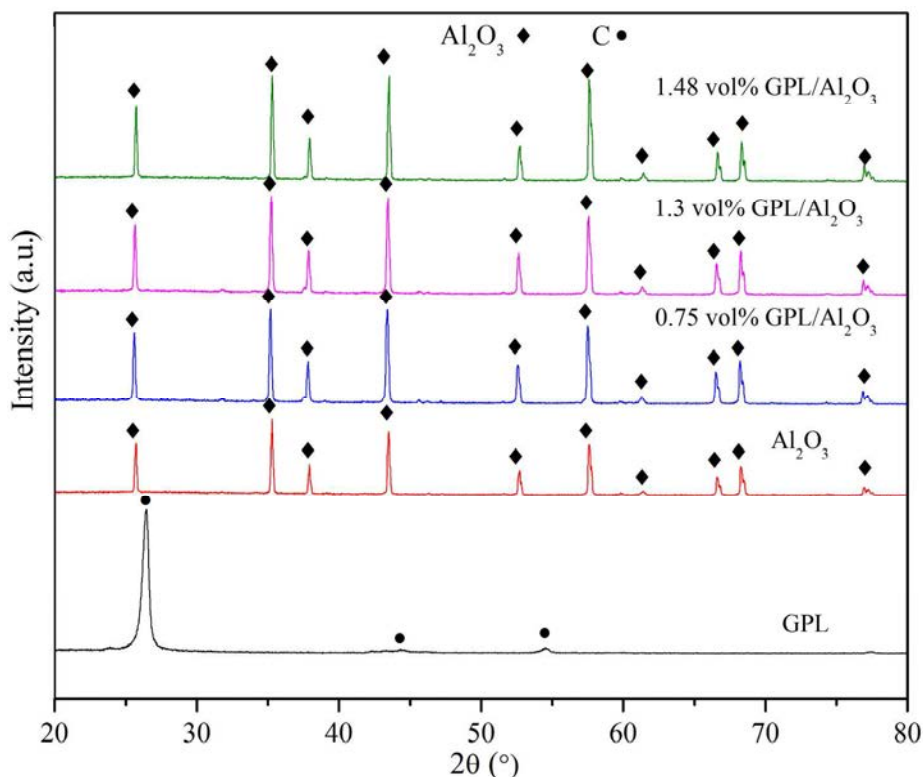


Figure 5. XRD patterns of the pristine GPL, pure Al<sub>2</sub>O<sub>3</sub> and GPL/Al<sub>2</sub>O<sub>3</sub> composites.



### 3.1.5 Mechanical properties of the pressure-less sintered pure $\text{Al}_2\text{O}_3$ and GPL/ $\text{Al}_2\text{O}_3$ composites

Figure 6a shows the densities of pure  $\text{Al}_2\text{O}_3$  and GPL/ $\text{Al}_2\text{O}_3$  composites. It can be seen that the pure  $\text{Al}_2\text{O}_3$  has been almost fully densified while the GPL- $\text{Al}_2\text{O}_3$  composites present relatively lower densities, which decrease from 99.2 to 95.6% with the increasing

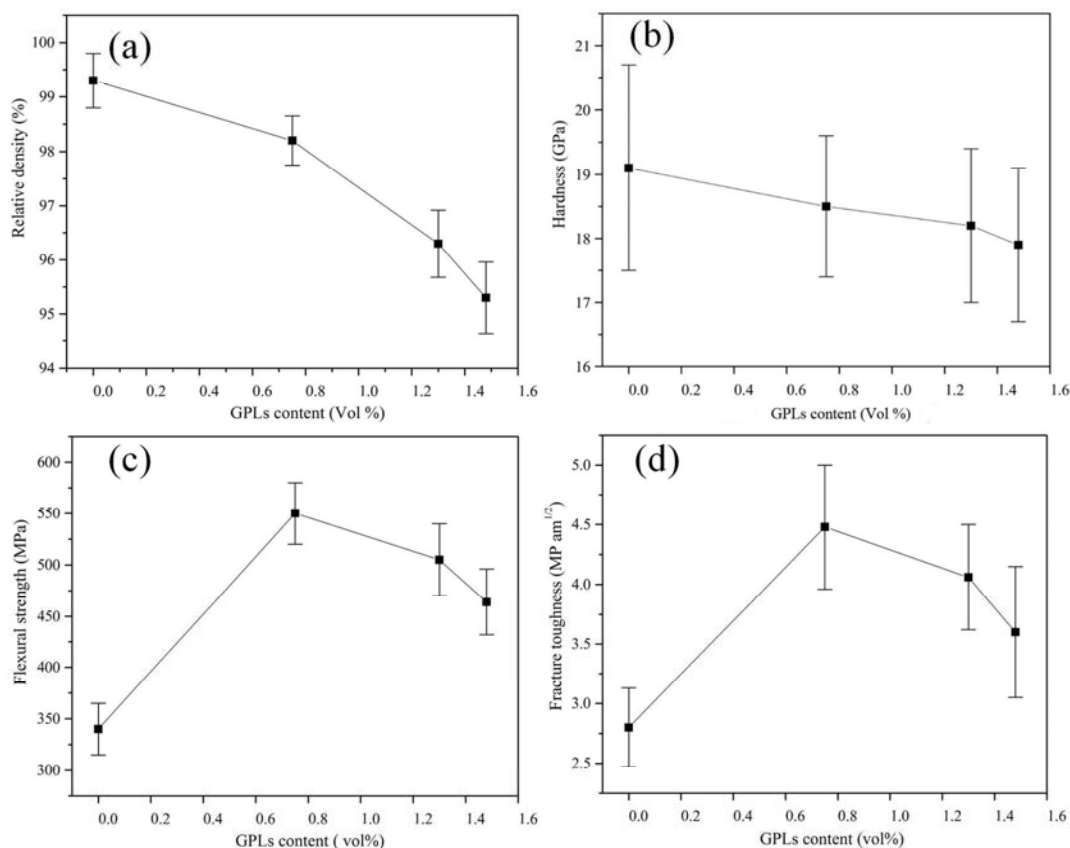


Figure 6. Densities and mechanical properties of the GPL/ $\text{Al}_2\text{O}_3$  samples as a function of GPLs content.

concentration of GPLs from 0.75 to 1.48 vol%. The result suggests that the addition of GPLs hinders the densification process of GPL/ $\text{Al}_2\text{O}_3$  composites during the pressure-less sintering. In authors' previous work, the 0.76vol% GPL/ $\text{Al}_2\text{O}_3$  ceramic composite with relative density of 99.92% was successfully fabricated at 1550 °C using SPS within 5 minutes<sup>14</sup> while in

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3 present study the relative density achieved for the 0.75vol% GPL/ $\text{Al}_2\text{O}_3$  ceramic composite  
4 prepared at 1650 °C and in 3 hours was only 98.23%. Therefore, it can be said that the  
5 densification rate during the pressure-less sintering is smaller than that in sintering with the  
6 aid of pressure and a higher sintering temperature or a longer sintering time may be required  
7 to obtain denser GPL/ $\text{Al}_2\text{O}_3$  samples.  
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14 Hardness of pure  $\text{Al}_2\text{O}_3$  and GPL/ $\text{Al}_2\text{O}_3$  composites are compared and plotted in Figure 6b. It  
15 is noted that the hardness decreases with an increasing percentage of GPLs. Usually higher  
16 density and smaller grain size would result in greater hardness. Although the decreased grain  
17 size brought by GPLs contributes to the increase of hardness, the lower density of the  
18 composites help decrease hardness.  
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27 Figure 6c presents flexural strength of the pure  $\text{Al}_2\text{O}_3$  and GPL/ $\text{Al}_2\text{O}_3$  composites. It can be  
28 observed that the flexural strength of the  $\text{Al}_2\text{O}_3$  has been significantly improved by adding  
29 GPLs and it increases considerably with a minor addition of GPLs and decreases with the  
30 further increase of GPLs. A maximum increase of approximately 60% in flexural strength  
31 was achieved by introducing 0.75 vol% GPLs. Similar to hardness, flexural strength is mainly  
32 affected by the grain size and residual pores. The smaller grain size with reduced flaw size  
33 would result in better flexural strength while the presence of pores would allow cracks to be  
34 formed easily and cause the small fracture energy due to the stress concentration around the  
35 pores<sup>37</sup>.  
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48 Fracture toughness of the sintered samples is shown in Figure 6d. It is noted that variation in  
49 fracture toughness exhibits the same trend as the flexural strength and all of the GPL/ $\text{Al}_2\text{O}_3$   
50 composites present improved fracture toughness in comparison with pure  $\text{Al}_2\text{O}_3$ . A  
51 significant increase in fracture toughness has been achieved by introducing 0.75vol% GPLs.  
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57 A further increase in GPLs leads to a decrease in fracture toughness. The obtained maximum  
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3 fracture toughness of the GPL/ $\text{Al}_2\text{O}_3$  composites is approximately 70% higher than that of  
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5 pure  $\text{Al}_2\text{O}_3$ .  
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7  
8 In this work, the hardness, flexural strength and fracture toughness achieved for the 0.75vol%  
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10 GPL/ $\text{Al}_2\text{O}_3$  composite reach 18.58, 550 Mpa and  $4.47 \text{ Mpa m}^{1/2}$  respectively and are higher  
11  
12 than those ( $17.46$ ,  $485 \text{ Mpa}$  and  $4.11 \text{ Mpa m}^{1/2}$ ) for the 0.76 vol% GPL/ $\text{Al}_2\text{O}_3$  composite  
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14 produced using SPS <sup>14</sup>. Therefore, it is expected that pressureless sintering can be able to  
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16 produce ceramic composites with better mechanical properties than SPS. In particular, given  
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18 the fact that advanced sintering techniques such as hot pressing and SPS are limited to the  
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20 fabrication of samples with simple shapes, pressure-less sintering is more advantageous when  
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22 complex shapes are desired for the sintered samples.  
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### 25 26 27 *3.1.6 Effects of GPLs on mechanical properties of the GPL/ $\text{Al}_2\text{O}_3$ composites*

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29 As mentioned in previous work <sup>12, 14</sup>, the mechanical properties of the densified GPL-  
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31 reinforced  $\text{Al}_2\text{O}_3$  are mainly dependent on the dispersion of GPL in the matrix and interaction  
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33 between GPLs and the ceramic matrix. In this study, well dispersed GPLs in the ceramic  
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35 matrix are achieved, as shown in Figures 7 a, c and e. A good dispersion of GPLs in the  
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37 ceramic matrix would contribute to the mechanical properties in two ways. On one hand, the  
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39 distribution of GPLs in the grain boundaries of the ceramic matrix prevents the migration of  
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41 grain boundaries during the long time sintering process and causes the formation of a fine  
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43 microstructure, which benefits the flexural strength and fracture toughness by decreasing the  
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45 defect size of the ceramic matrix and increasing the contact areas between the GPLs and the  
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47 ceramic matrix respectively. On the other hand, due to the GPLs' high Young's modulus, the  
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49 embedded GPLs in the ceramic matrix (Figures 7 b, d and f) can reinforce the matrix, leading  
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51 to improved flexural strength.  
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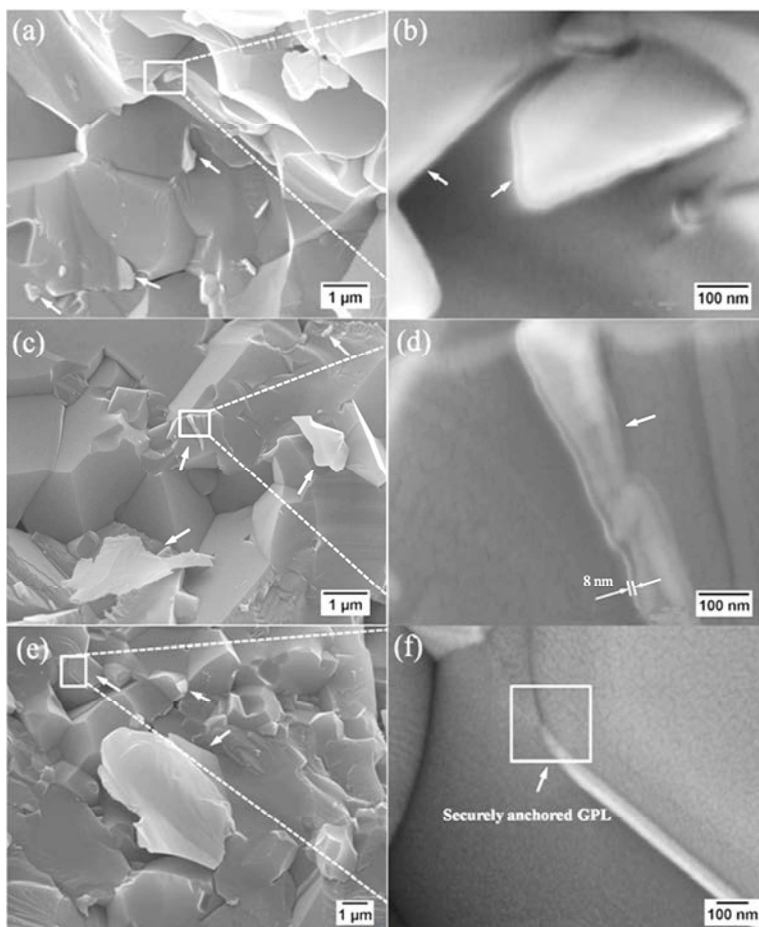


Figure 7. Fracture surfaces of GPL/ $\text{Al}_2\text{O}_3$  composites. (b), (d) and (f) are the magnified square area in (a), (c) and (e) respectively. White arrows indicate GPLs.

Meanwhile, due to the long sintering time at high temperature, GPLs are securely anchored in the ceramic matrix. Close interaction between GPLs and the ceramic matrix are observed (Figures 7 b and f). A good interaction between GPLs and the ceramic matrix would enable an efficient load transfer from the ceramic matrix to the GPLs, resulting in the improvement in flexural strength. Additionally, high energy is required to overcome the strong interfacial friction at the interface between the ceramic matrix and the GPLs to pull out the GPLs, leading to the increase in fracture toughness. It should be noted that GPL agglomerates

(Figures 7 d) are observed and embedded in the ceramic matrix. It is believed that during crack propagation interlayer sliding in the GPL agglomerates is likely to occur to help energy dissipation and thus contribute to the fracture toughness of the composites<sup>38</sup>.

### 3.2 In-vitro biocompatibility of GPL/ $\text{Al}_2\text{O}_3$ composites

MTT tests were used to assess cell viability by measuring the levels of active mitochondrial dehydrogenases to convert MTT into formazan crystals. Cell viability of MC3T3 cells on control and sintered samples was investigated with MTT assay after incubation for 3 hours, 1 day and 3 days respectively. The results are shown in Figure 8. As expected, the least viability is found for the negative control and the cell viability for the GPL/ $\text{Al}_2\text{O}_3$  samples is

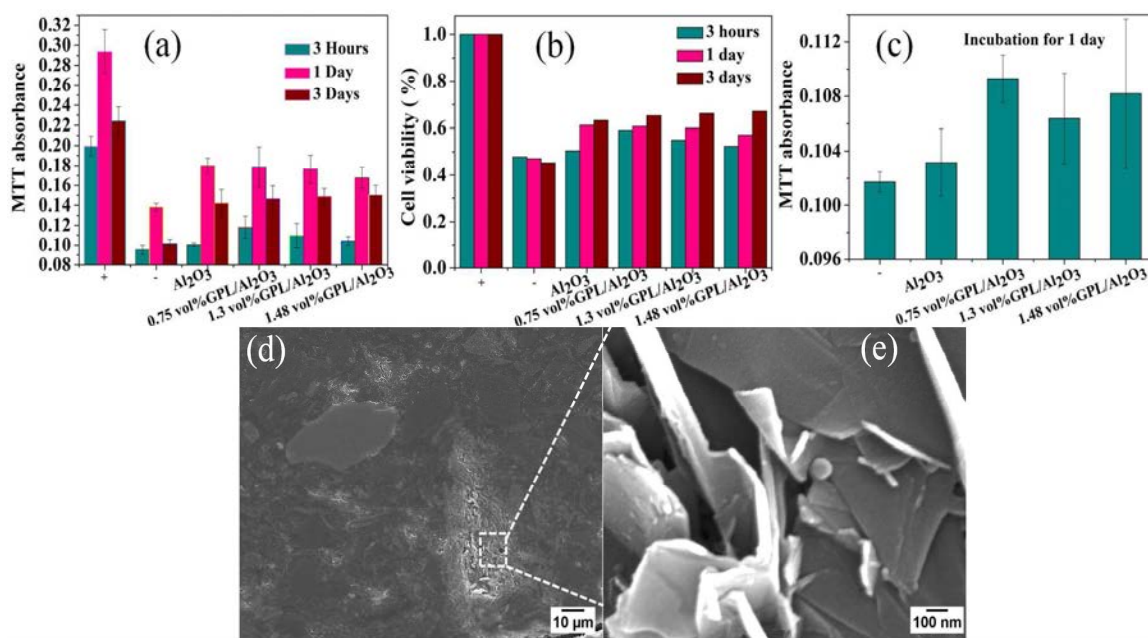


Figure 8. Cell viability on varied samples (a-c) and SEM images of the surfaces of the GPL/ $\text{Al}_2\text{O}_3$  samples(d-e).

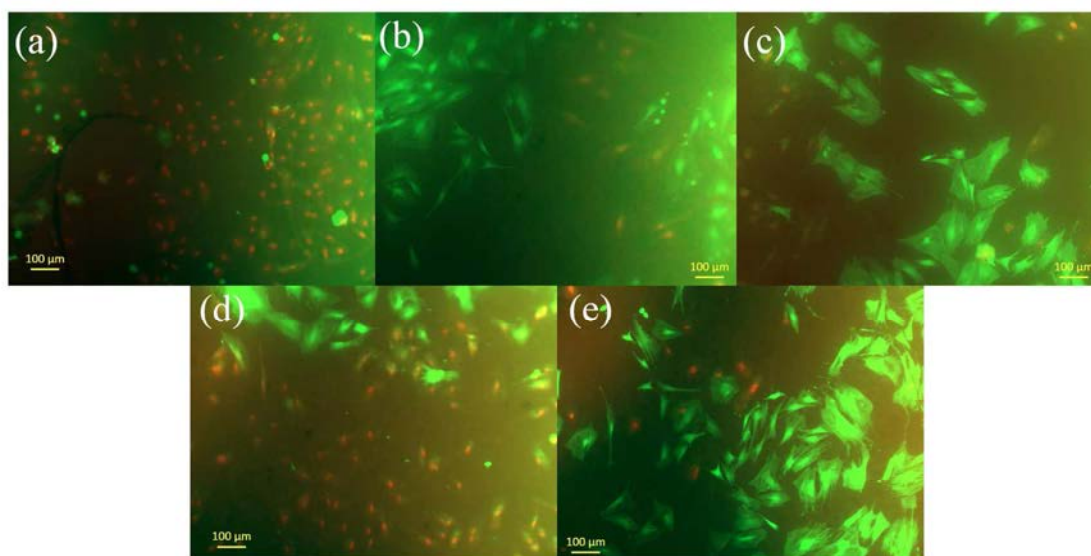
generally higher than that for the monolithic  $\text{Al}_2\text{O}_3$  after 3 hours or 3 days, which indicates a better initial attachment of cells and viability for GPL/ $\text{Al}_2\text{O}_3$  samples (Figure 8a and 8b).

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3 However, GPL/ $\text{Al}_2\text{O}_3$  samples exhibit slightly lower cell viability after 1 day of incubation.  
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5 Particularly, the 1.48vol% GPLs/ $\text{Al}_2\text{O}_3$  ceramic composite permits a lower survival rate of  
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7 cells in comparison to other samples. It implies that during the culturing period, cell  
8  
9 proliferated and expanded along the sample surfaces except where the sharp edges of the  
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11 GPL clusters exposed on the surfaces of the ceramic composites (Figure 8e). These exposed  
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13 GPL edges prohibit cell proliferation, which causes the decrease in cell survivability on the  
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15 surfaces. The 1.48 vol% GPLs/ $\text{Al}_2\text{O}_3$  ceramic composite contains a higher percentage of  
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17 pores in which more GPLs edges are likely to be exposed on the surface of the ceramic  
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19 composite. Therefore, lower cell viability was found for the 1.48vol% GPLs/ $\text{Al}_2\text{O}_3$  ceramic  
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21 composite.  
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26 Another parallel set of MTT tests were carried out using DMEM without addition of fetal  
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28 bovine serum, to investigate initial cell adhesion on substrates without the disturbance of  
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30 serum proteins (Figure 8c). The result suggests that more cells attached on GPL/ $\text{Al}_2\text{O}_3$   
31  
32 samples as compared to the monolithic  $\text{Al}_2\text{O}_3$  sample after incubation for 1 day. These results  
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34 show that the GPL/ $\text{Al}_2\text{O}_3$  ceramic has comparable or more favourable biocompatibility for  
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36 osteoblast precursor cells, which is in agreement with the reported work concerning the  
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38 graphene reinforced HA composites<sup>22</sup>.  
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42 Cell adhesion tests were also performed using BMSCs and the florescence images are shown  
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44 in Figure 9. It can be observed that cells with different morphologies are present on the  
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46 materials. The number of the cells is reduced with the addition of GPLs. This observation  
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48 indicates negative effect of GPL on the proliferation of stem cells. However, as suggested by  
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50 Figure 9c-e, cells on the surfaces of the GPL/ $\text{Al}_2\text{O}_3$  composites demonstrate better elongation  
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52 and expansion. This might indicate that the presence of GPLs stimulated adhered BMSCs in a  
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54 way that growing and expanding is encouraged while cell division is prohibited. The mixed  
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3 results suggest that other factors such as surface morphology [7], porosity, contact angle with  
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5 cell media and surface chemistry [8], accumulation of charge [9] and the grain size of the  
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7 ceramics might also play an important role in the interaction between stem cells and material  
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9 surfaces. Other tests need to be done in the future to examine the complex behaviour of stem  
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11 cells on the composite, for example cell differentiation identification.  
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Figure 9. Fluorescence microscopy images of the stem cells on the PDMS (a),  $\text{Al}_2\text{O}_3$  (b), 0.75 vol% GPL/ $\text{Al}_2\text{O}_3$  (c), 1.3 vol% GPL/ $\text{Al}_2\text{O}_3$  (d), 1.48 vol% GPL/ $\text{Al}_2\text{O}_3$  (e).

The survival rate of the cells was qualitatively calculated and shown in Figure S5 based upon the observed results shown in Figure 9. It is found that the number of attached cells and adherent live cells for the GPL/ $\text{Al}_2\text{O}_3$  composites is smaller than that for monolithic  $\text{Al}_2\text{O}_3$  (Figure S5a), indicating un-preferable attachment of stem cells on the GPL/ $\text{Al}_2\text{O}_3$  composites. However, the survival rate of the stem cells on composites with additions of 0.75 and 1.48 vol% GPLs are noticed to be higher than that on the monolithic  $\text{Al}_2\text{O}_3$  (Figure S5b). It is believed that the variation in the survivability of the stem cells (Figure S5b) can be explained by the double effects of the GPL content and the pores within the ceramics. The presence of pores

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3 will increase the likelihood of GPL edges being exposed to the cells, leading to the death of  
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5 the cells (Figure 8e) while the GPLs with flat surfaces can certainly facilitate the spread and  
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7 proliferation of the cells.  
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10 Previous reports have shown that both CNT/Al<sub>2</sub>O<sub>3</sub> composites produced by HIP<sup>2</sup> and  
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12 GPL/HA composites fabricated using SPS<sup>22</sup> exhibit attractive biocompatibilities and cell  
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14 viability has been increased due to the addition of carbon fillers. Similar results were  
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16 obtained in this study using pressure-less sintering. In particular, the enhanced spread of stem  
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18 cells on the GPL/Al<sub>2</sub>O<sub>3</sub> composites observed in the study is promising for the materials to be  
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20 used in a variety of biomedical purposes in the future and pressure-less sintering can certainly  
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22 be a possible means to produce such materials.  
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#### 26 **4. Conclusions**

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29 In the study, GPL/Al<sub>2</sub>O<sub>3</sub> composites were successfully fabricated using a pressure-less  
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31 sintering process. A minor addition of GPLs has demonstrated to be able to efficiently  
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33 improve the fracture toughness as well as flexural strength of the Al<sub>2</sub>O<sub>3</sub> ceramics. The pull-  
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35 out toughening mechanism and strong interaction between the GPLs and the matrix are  
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37 regarded as the reasons for the improvements. XRD and Raman results suggest that no  
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39 traceable reaction products in the GPL/Al<sub>2</sub>O<sub>3</sub> composites were found after the sintering and  
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41 graphitic defects in GPLs were produced during the high temperature processing. MTT tests  
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43 show that the GPL/Al<sub>2</sub>O<sub>3</sub> ceramic has comparable or more favourable biocompatibility  
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45 toward osteoblasts. However, the cell adhesion tests with stem cells indicate otherwise,  
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47 possibly caused by the pores and GPL edges exposed to the cells, leading to the narcosis of  
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49 the cells while the GPLs with flat surfaces can facilitate the spread and proliferation of the  
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51 cells. Other factors affecting the interaction between the stem cells and ceramics can be  
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3 surface morphology, porosity, contact angle with cell media and surface chemistry,  
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5 accumulation of charge and the grain size of the ceramics.  
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### 8 **Supporting information**

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11 SEM images of the powder mixtures, TGA of the sintered samples, Grain sizes of GPL/Al<sub>2</sub>O<sub>3</sub>  
12  
13 composites, Raman spectra of the pristine GPL and GPLs in powder mixtures and living and  
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15 dead BMSCs after culturing for 3 Days. The material is available free of charge via the  
16  
17 Internet at <http://pubs.acs.org>.  
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21  
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27  
28 project under grant number 710705.  
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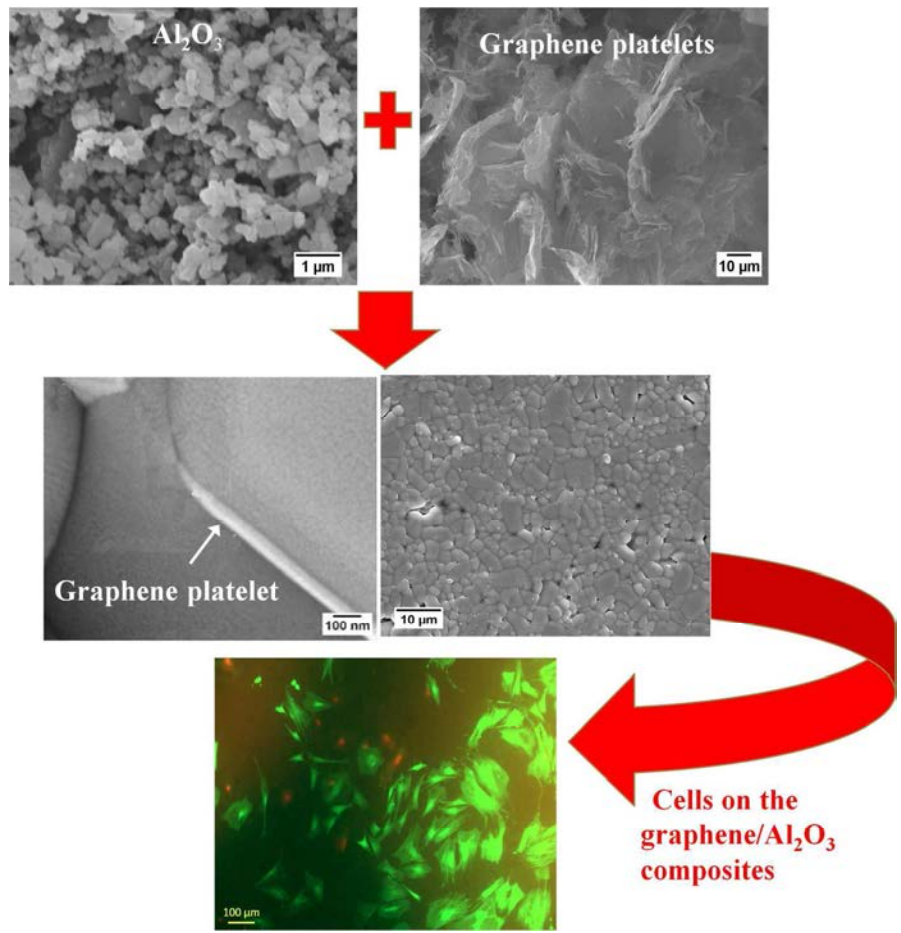
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# Table of Contents



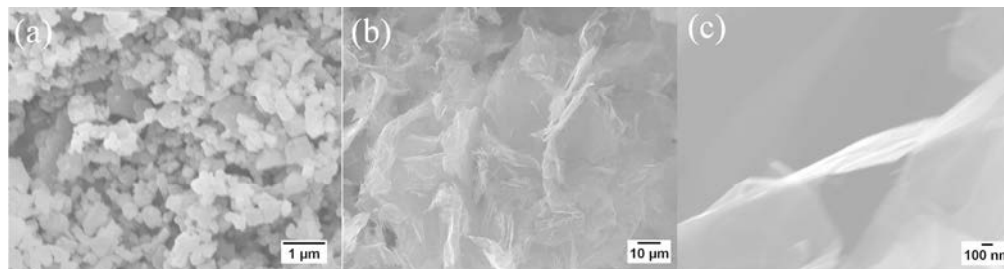


Figure.1 SEM images of Al<sub>2</sub>O<sub>3</sub> (a) powders and GPLs (b-c).  
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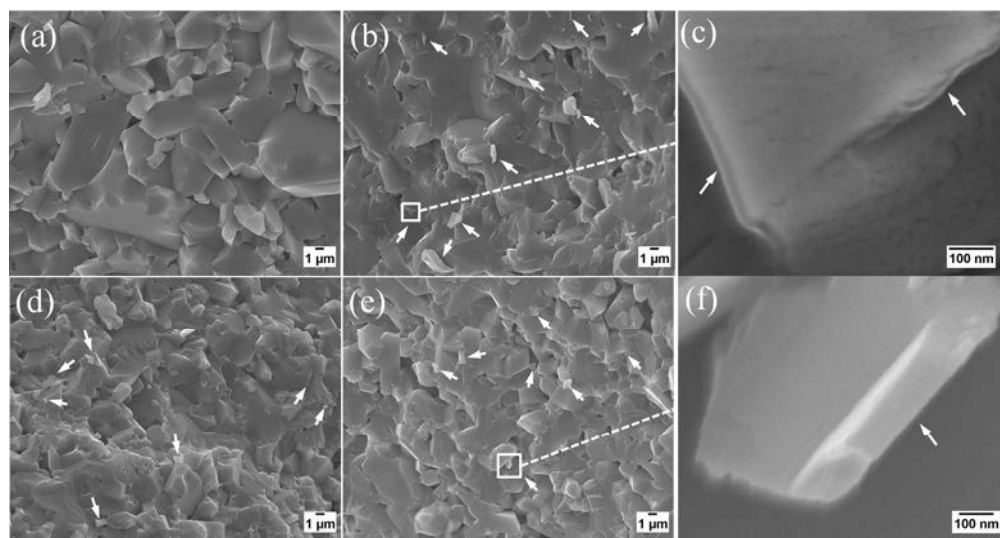


Figure 2. Fracture surfaces of sintered GPL/Al<sub>2</sub>O<sub>3</sub> composites. (a) Al<sub>2</sub>O<sub>3</sub>, (b) 0.75 vol% GPL/Al<sub>2</sub>O<sub>3</sub>, (d) 1.3 vol% GPL/Al<sub>2</sub>O<sub>3</sub>, (e) 1.48 vol% GPL/Al<sub>2</sub>O<sub>3</sub>. (c) and (f) are magnified images of square area in (b) and (e) respectively. White arrows indicate GPLs.

376x200mm (150 x 150 DPI)



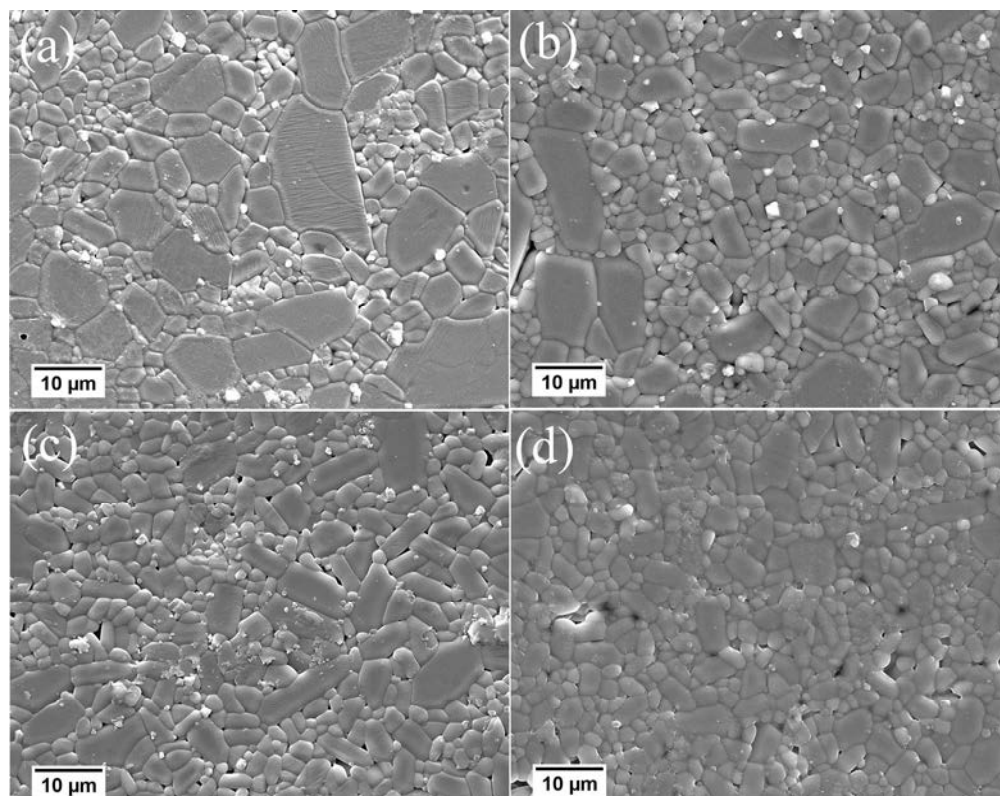


Figure 3. SEM images of polished and thermally etched surfaces. (a) Al<sub>2</sub>O<sub>3</sub>, (b) 0.75 vol% GPL/Al<sub>2</sub>O<sub>3</sub>, (c) 1.3 vol% GPL/Al<sub>2</sub>O<sub>3</sub>, (d) 1.48 vol% GPL/Al<sub>2</sub>O<sub>3</sub>.  
326x257mm (150 x 150 DPI)

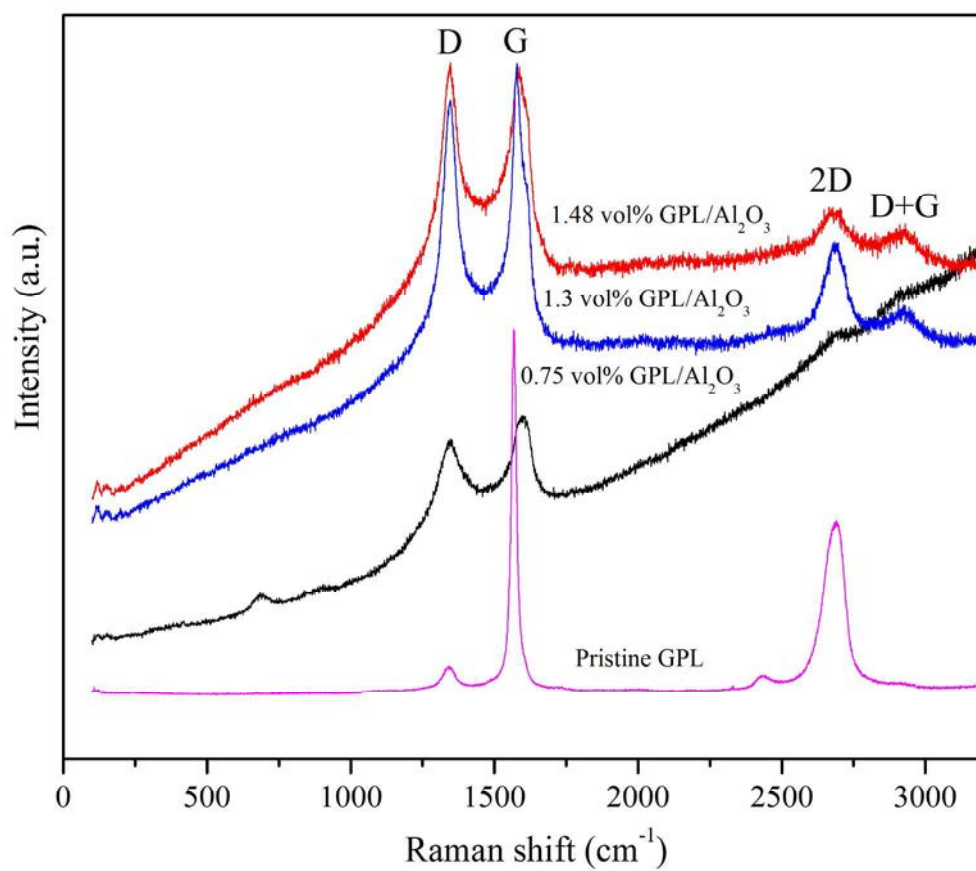


Figure 4. Raman spectra of the pristine GPL and GPLs after sintering.  
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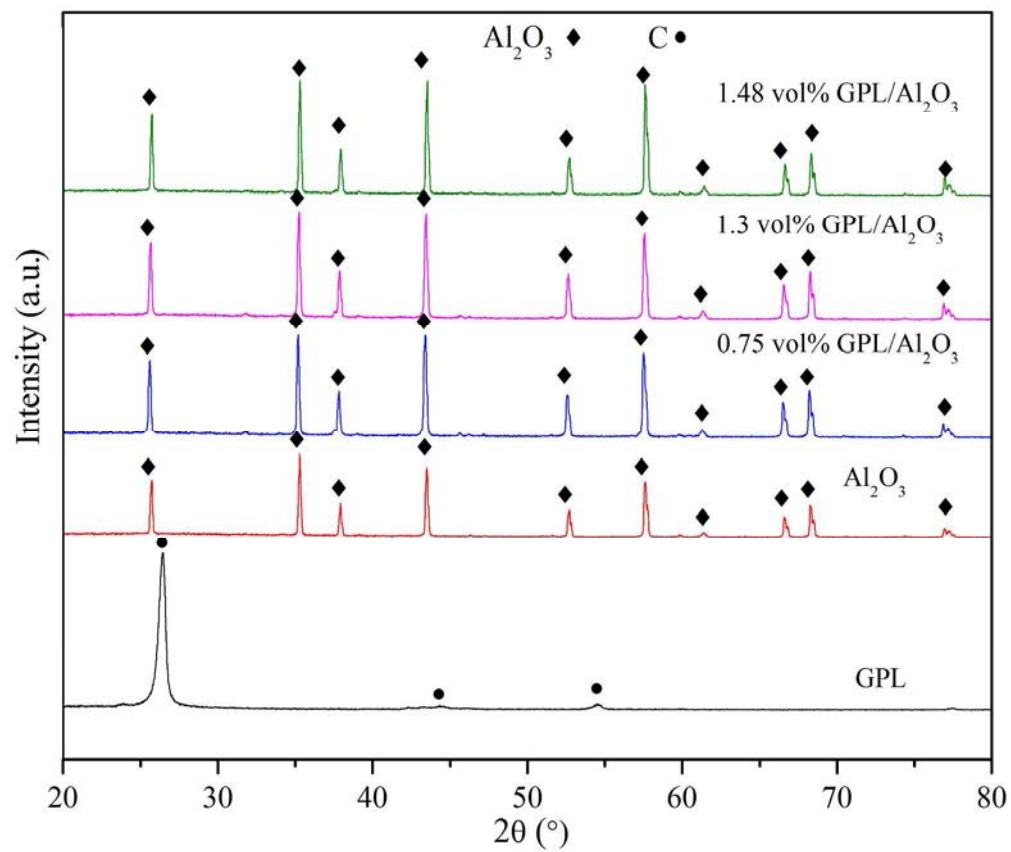


Figure 5. XRD patterns of the pristine GPL, pure Al<sub>2</sub>O<sub>3</sub> and GPL/Al<sub>2</sub>O<sub>3</sub> composites.  
209x178mm (300 x 300 DPI)

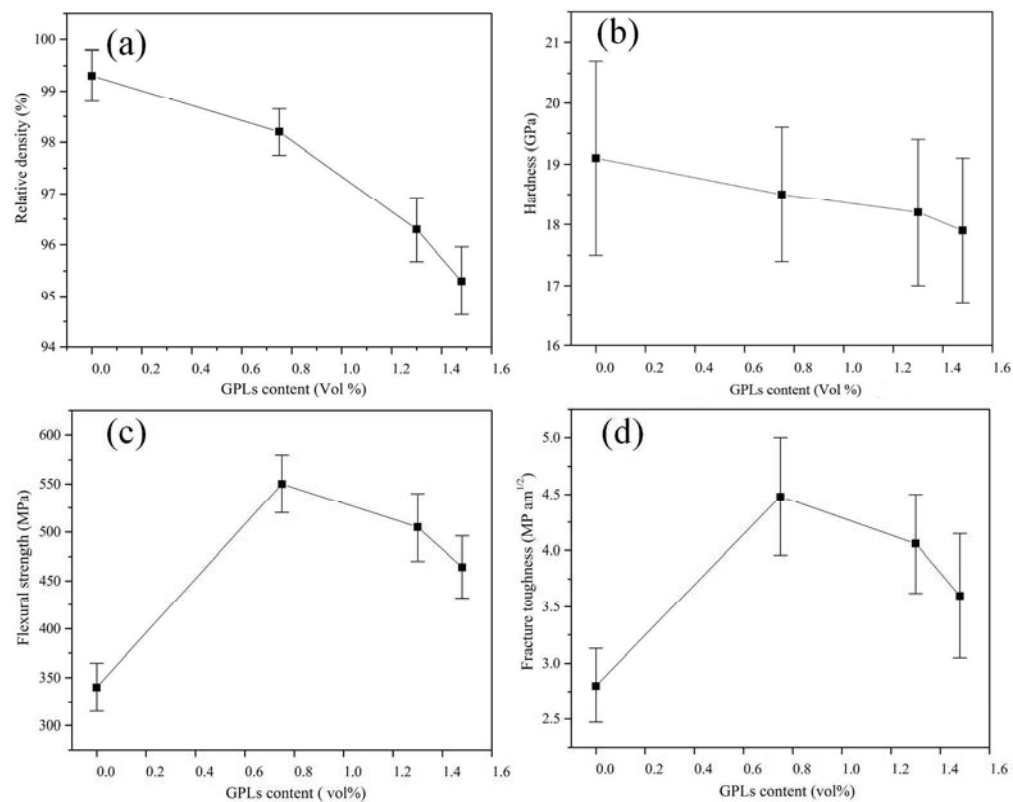


Figure 6. Densities and mechanical properties of the GPL/Al<sub>2</sub>O<sub>3</sub> samples as a function of GPLs content. 428x344mm (150 x 150 DPI)

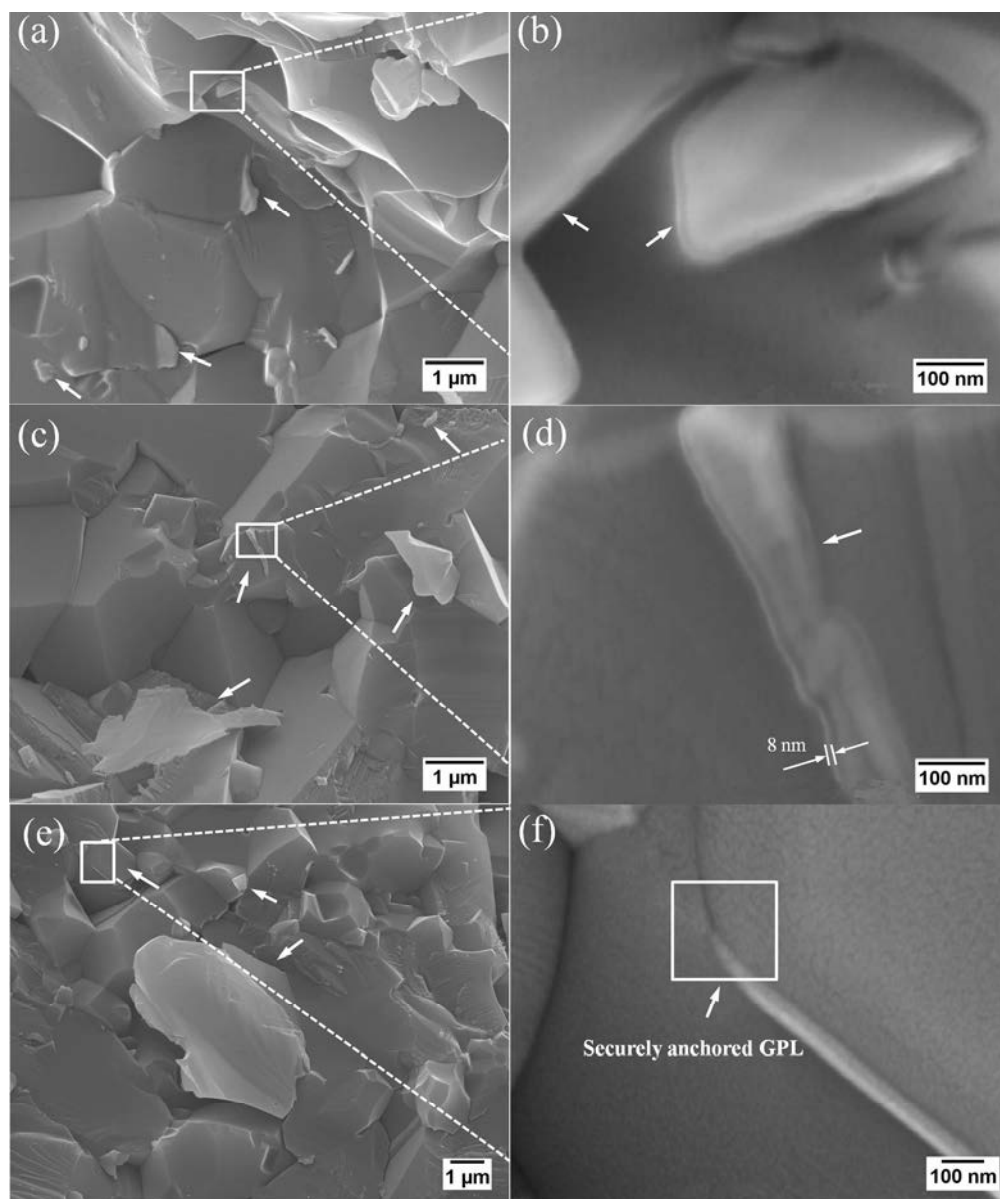


Figure 7. Fracture surfaces of GPL/Al<sub>2</sub>O<sub>3</sub> composites. (b), (d) and (f) are the magnified square area in (a), (b) and (e) respectively. White arrows indicate GPLs.  
326x388mm (150 x 150 DPI)

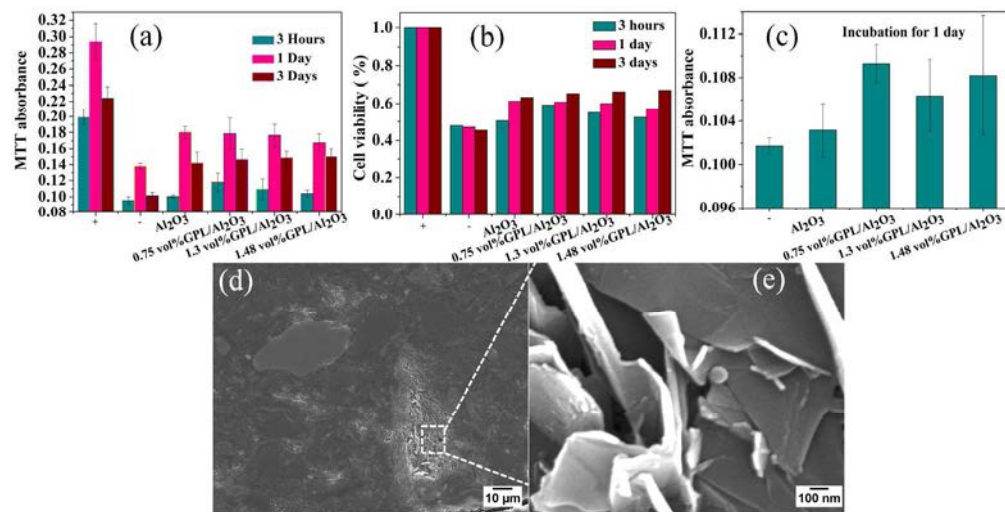


Figure 8. Cell viability on varied samples (a-c) and SEM images of the surfaces of the GPL/Al<sub>2</sub>O<sub>3</sub> samples(d-e).

476x241mm (150 x 150 DPI)

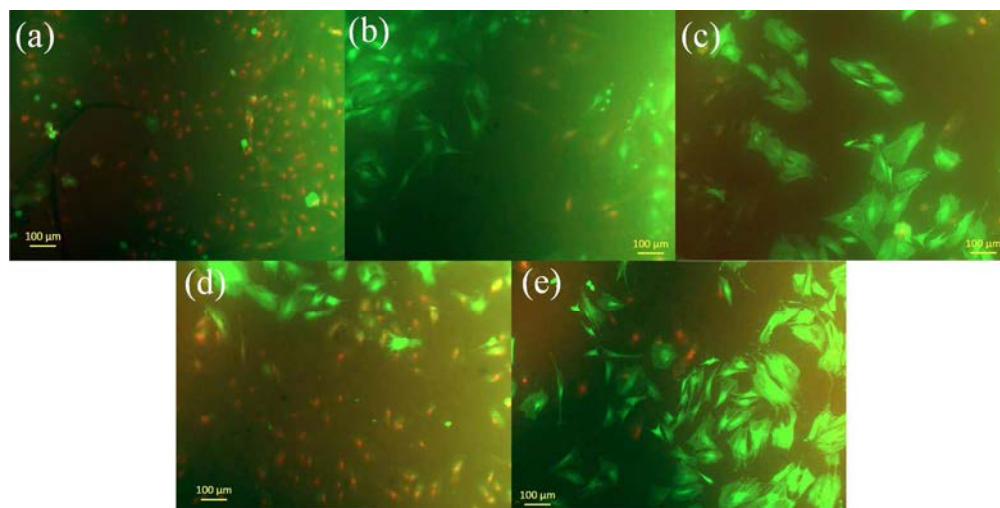


Figure 9. Fluorescence microscopy images of the stem cells on the PDMS (a), Al<sub>2</sub>O<sub>3</sub> (b), 0.75 vol% GPL/Al<sub>2</sub>O<sub>3</sub> (c), 1.3 vol% GPL/Al<sub>2</sub>O<sub>3</sub> (d), 1.48 vol% GPL/Al<sub>2</sub>O<sub>3</sub> (e).  
439x219mm (150 x 150 DPI)

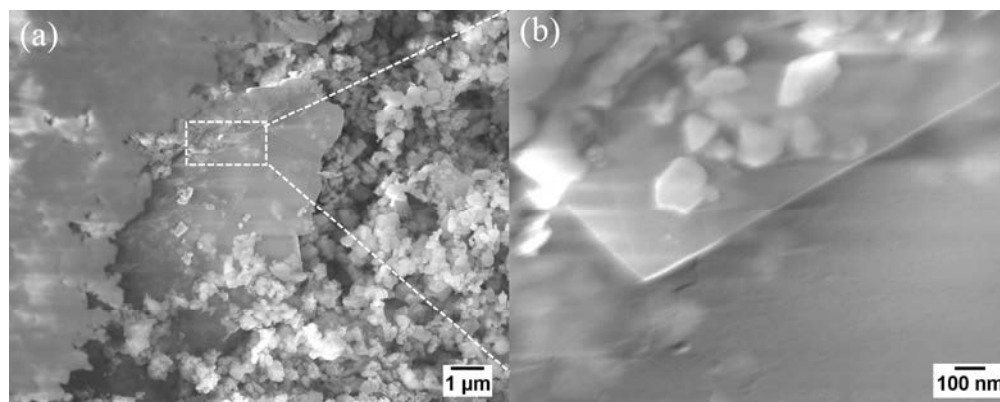


Figure S1. SEM images of the powder mixtures. (b) is the magnified square area in (a), showing GPL.  
449x179mm (300 x 300 DPI)



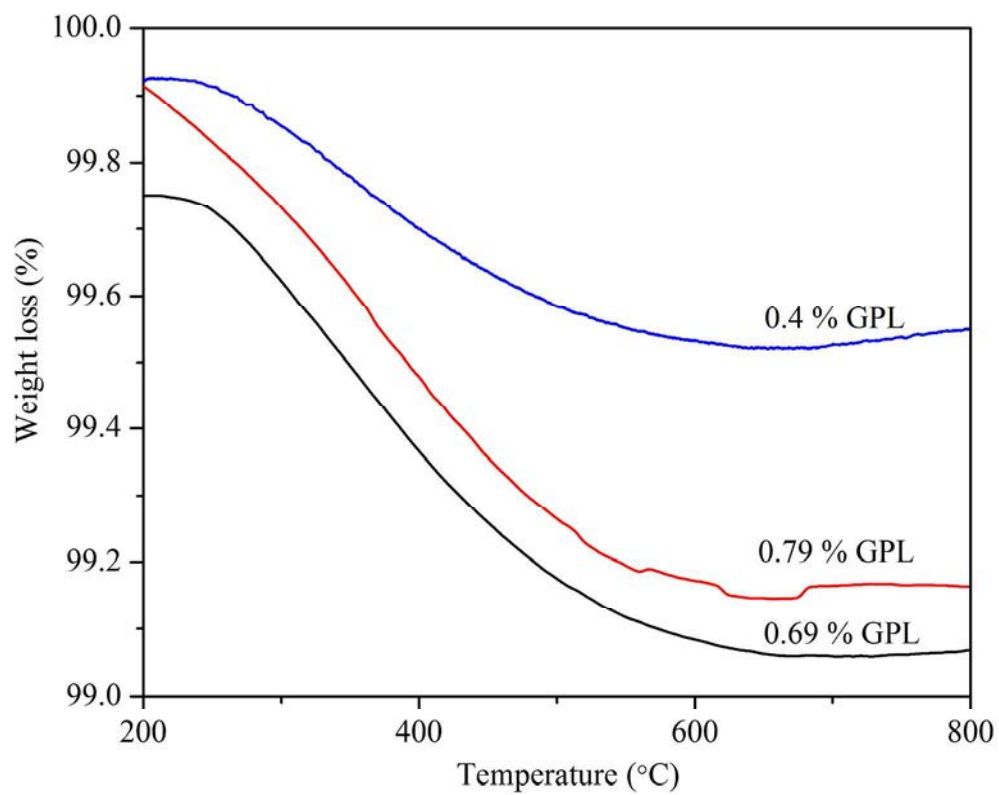


Figure S2. TGA of the sintered samples with different GPLs content.  
225x181mm (300 x 300 DPI)

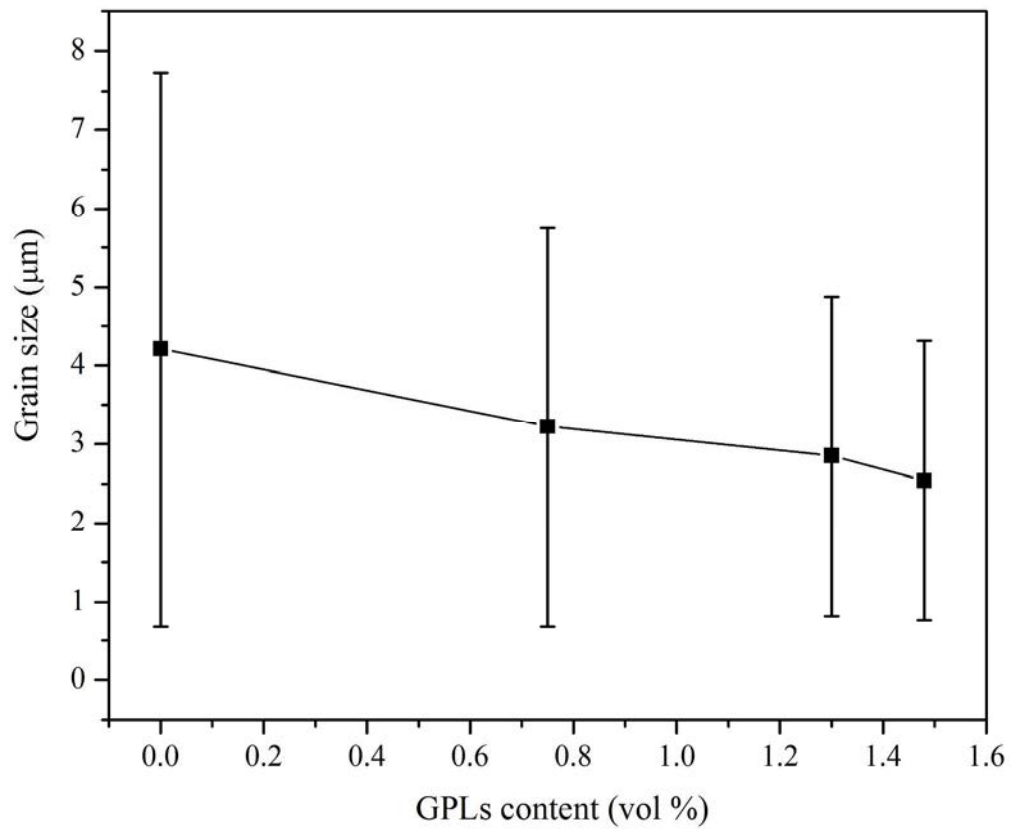


Figure S3. Grain sizes of Al<sub>2</sub>O<sub>3</sub> matrices as a function of content of GPLs.  
212x176mm (300 x 300 DPI)

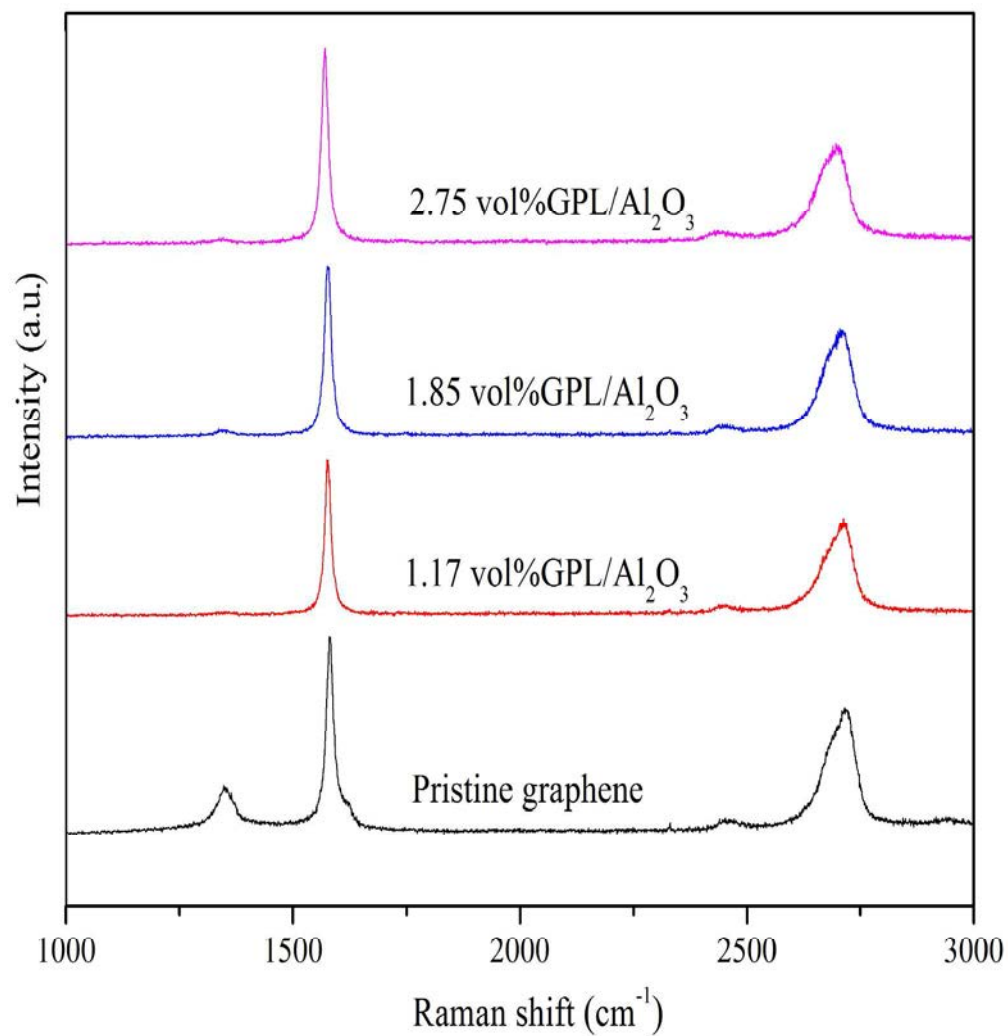


Figure S4. Raman spectra of the pristine GPL and GPLs in powder mixtures.  
215x224mm (300 x 300 DPI)

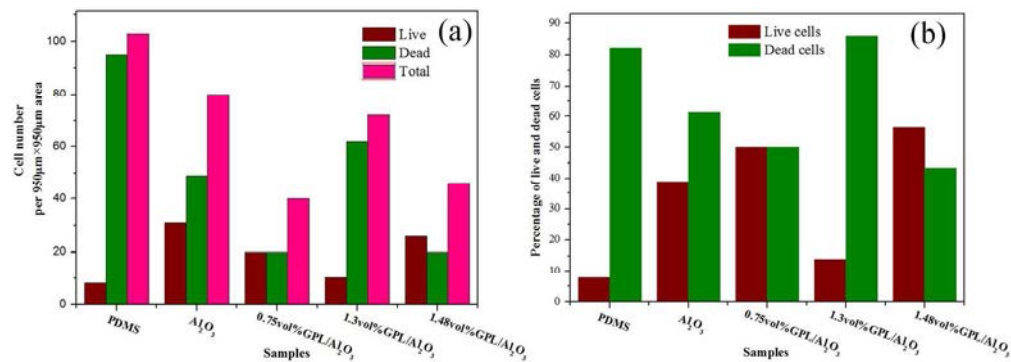


Figure S5. Living and dead BMSCs in a 950  $\mu m \times 950 \mu m$  area after culturing for 3 Days. (a) and (b) show the cell number and percentage of live and dead cells on the samples respectively.  
507x180mm (300 x 300 DPI)